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KINETICS OF BIMOLECULAR EXCHANGE  
REACTIONS: A COMPUTATIONAL APPROACH

By

Chris Carruthers

A thesis submitted to  
the School of Graduate Studies and Research  
in partial fulfillment of the requirements  
for the degree of Master of Science

Department of Chemistry

University of Ottawa

Ottawa, Ontario



Christopher Carruthers, Ottawa, Canada, 1992



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ISBN 0-315-85809-5

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## Abstract

In order to better understand the kinetics of gas phase bimolecular exchange reactions a computer program has been written which allows easy exploration of the time dependent and vibrational level dependent details of this class of reactions. BIMSIM (for BIMolecular exchange reaction SIMulation) is intended as a very flexible "virtual laboratory" which can easily be configured and reconfigured for a wide range of different experiments (e.g., laser pulse or shock tube), different initial conditions (e.g., of temperature, reactant concentration, and molecular environment), for different reactions in this class, and for different levels of approximation. In order to test the validity and demonstrate the use of the program a reaction system was found for which appropriate input data is available and for which suitably detailed analytical calculations have been done. Agreement was found to be excellent. Using BIMSIM, results of chemical interest were obtained for the reaction  $\text{Br} + \text{HCl} \rightarrow \text{HBr} + \text{Cl}$ . It was found that non-equilibrium depletion of the vibrational levels of HCl are as much as a factor of 10 and that they depend on the relative amounts of Br, HCl and inert diluent He, as well as on the temperature, becoming more pronounced around 500 K. Interesting details of the time dependence of the fractional level populations are discussed.

## Acknowledgement

I would like to thank my supervisor Prof. Heshel Teitelbaum, especially for his inexhaustible patience. I would also like to thank my wife Karen McCarthy and my good friend and colleague Dr. Victoria Barclay. I gratefully acknowledge an Ontario Graduate Scholarship from the Government of Ontario. This research was also supported by the Natural Sciences and Engineering Research Council of Canada. Finally, the generous computing resources made available by the University of Ottawa which allowed the completion of this work are acknowledged.

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# Chapter 1. Introduction

## 1.1 Scope

The gas-phase exchange reaction is one of the most common, important, relatively simple, and widely studied chemical reactions (see, for example, ref. [1]). In order to better understand the kinetics of such reactions of the form



a computer program has been written by the author which allows easy exploration of the time dependent details of this class of reactions. BIMSIM (for BIMolecular exchange reaction SIMulation) is intended as a very flexible “virtual laboratory” which can easily be configured and reconfigured for a wide range of different experiments (e.g., laser pulse or shock tube), different initial conditions (e.g., of temperature, reactant concentration, and molecular environment), for different reactions in this class, and for different levels of approximation. See Appendix A for a detailed description of BIMSIM and Appendix C for the source code.

The need to study the kinetics of bimolecular reactions with their vibrational detail stems from two important observations: a) endothermic bimolecular reaction rates are enhanced by vibrational energy of the reagents; b) exothermic bimolecular reactions often produce species which are vibrationally and rotationally excited[2]. Until now kineticists have simulated real reaction mechanisms (consisting of several bimolecular reactions) by implicitly assuming that the vibrational and rotational levels of all multiatomic species are distributed according to Boltzmann statistics. This is an approximation which is not justified particularly for fast chemical reactions or for slowly relaxing species. Thus, it is generally true that the population of the high vibrational levels of reagents are depleted, whereas the high vibrational levels of products are overpopulated. Since the products of one reaction are sometimes

the reagents for subsequent reactions, the whole process is rather complicated. Depending on the reaction conditions different scenarios can be envisaged. The ‘rate coefficient’ becomes distorted from the kineticist’s impression of it. Indeed the whole concept of a rate constant comes into question. In order to see how serious such effects can become, it is important to investigate the kinetics at its microscopic detail.

The approach, admitting to non-Boltzmann distributions, can be termed, by definition, “non-equilibrium kinetics”. Although this term encompasses all kinds of energy levels, including electronic and translational, BIMSIM is restricted to the cases of chemical interest and does not consider plasma temperatures where electronic and translational effects become important[3]. What is known currently about vibrational/rotational non-equilibrium kinetics is restricted in the literature to the field of unimolecular reactions. BIMSIM generalizes the techniques to include bimolecular reactions.

## 1.2 Non-equilibrium Kinetics of Unimolecular Reactions

Figure 1.1 is a schematic diagram showing the energy levels of molecule BC, and the barrier to rupture the weakest bond,  $E_0$ . For most molecules of chemical interest  $E_0 \gg k_B T$  (where  $k_B$  is the Boltzmann constant and  $T$  is the absolute temperature). Because of the high barrier, reaction occurs by stepwise climbing of the vibrational ladder. It is only near the barrier that excited molecules actually can dissociate collisionally and only above the barrier that excited molecules fall apart spontaneously. The whole process is a competition between vibrational excitation and reaction. At the low pressure limit of a unimolecular reaction, the processes which determine the rate of reaction are energy transfer steps. At this limit, energy transfer (vibrational excitation or replenishment of the population of those levels near the dissociation limit depleted by reaction) is inefficient. The balance is struck leaving a steady non-Boltzmann distribution. The reaction can be modeled by solving the very large

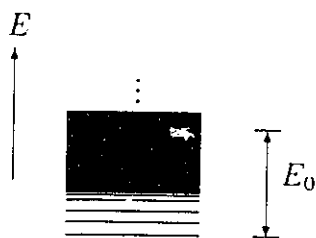


Figure 1.1: Schematic Energy Levels of a Molecule

set of differential rate equations, commonly called the master equation. For  $E < E_0$

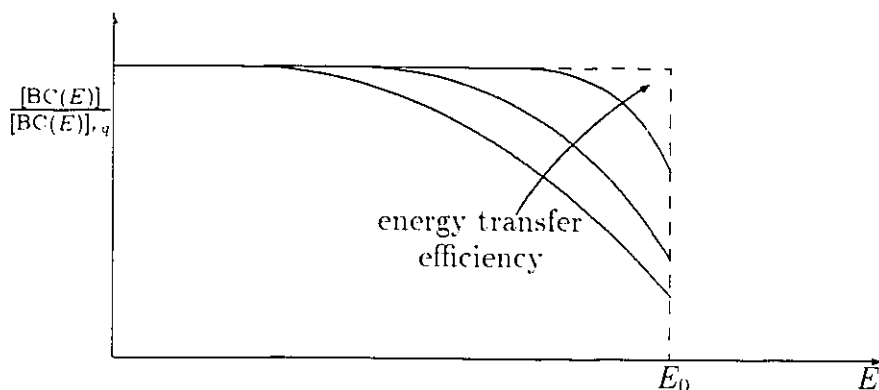
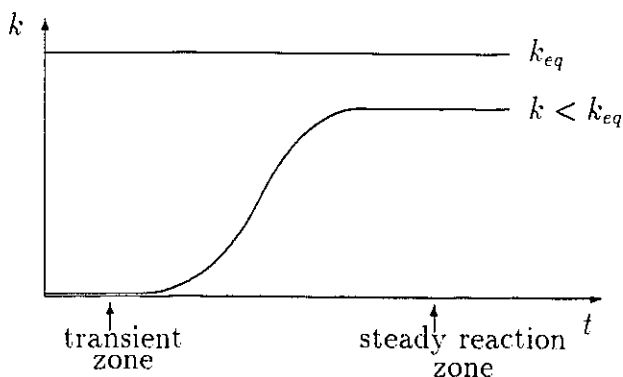
$$\frac{d[\text{BC}(E, t)]}{dt} = \sum_{E'} k(E' \rightarrow E)[\text{M}][\text{BC}(E', t)] - \sum_{E'} k(E \rightarrow E')[\text{M}][\text{BC}(E, t)]. \quad (1.2)$$

For  $E \geq E_0$

$$\begin{aligned} \frac{d[\text{BC}(E, t)]}{dt} = & \sum_{E'} k(E' \rightarrow E)[\text{M}][\text{BC}(E', t)] \\ & - \sum_{E'} k(E \rightarrow E')[\text{M}][\text{BC}(E, t)] - k(E)[\text{BC}(E, t)]. \end{aligned} \quad (1.3)$$

Here  $[\text{BC}(E, t)]$  is the concentration of the molecule with internal energy  $E$  at time  $t$ ,  $[\text{M}]$  is the concentration of an inert bath gas,  $k(E)$  is the specific rate constant for spontaneous dissociation of the molecule,  $k(E \rightarrow E')$  is the rate constant for collisional energy transfer from energy level  $E$  to energy level  $E'$ , and  $k(E' \rightarrow E)$  is that for the reverse process.  $k(E \rightarrow E')$  and  $k(E' \rightarrow E)$  are related by microscopic reversibility. This relationship guarantees that, in the absence of reaction, a Boltzmann distribution results at  $t = \infty$ .

Equation 1.3 can be solved analytically[4] under the assumptions that BC is big, energy transfer is poor,  $k(E' \rightarrow E)$  decreases exponentially with energy gap, BC is highly dilute in the bath gas M, BC is translationally equilibrated, M is thermally equilibrated, and the reverse recombination of B and C is unimportant. Because of the large number of levels involved one often replaces the summation by an integration. For the special case of the steady state limit (when the distribution

Figure 1.2: The Depletion of BC at high  $E$ Figure 1.3:  $k$  vs.  $t$ 

becomes invariant in time) the following non-Boltzmann distribution

$$\frac{BC(E)}{BC(E)_{eq}} \approx 1 - \frac{k_B T}{\langle \Delta E \rangle + k_B T} \exp\left[-\frac{(E_0 - E)}{k_B T}\right] \quad (1.4)$$

and the non-equilibrium rate coefficient  $k$  (compared to  $k_{eq}$ , the one where a Boltzmann distribution prevails):

$$\frac{k}{k_{eq}} \approx \frac{\langle \Delta E \rangle}{k_B T} < 1 \quad (1.5)$$

where  $\langle \Delta E \rangle$  is the average energy transferred per collision ( $\ll k_B T$ ). These results are shown schematically in Figures 1.2 and 1.3. Figure 1.3 additionally shows the time-evolution of the non-equilibrium rate-coefficient,  $k$ . This is a phenomenological quantity defined as  $k = -Rate/([BC][M])$ . Equation 1.5 shows that  $k$  is depressed

from  $k_{eq}$ . It turns out that this depression becomes more and more prominent as the temperature increases, resulting in convex Arrhenius plots. It is only after the transient process is finished that  $k$  becomes essentially constant.

As important as unimolecular reactions are, they form, after all, only a small fraction of chemical reactions. For every molecule that self-destructs there is an almost countless number of possible chemical partners with which it can react bimolecularly. So what is known about non-equilibrium kinetics is a small fraction of what should be known. The same guiding principles successfully used for unimolecular reactions (competition between energy transfer and reactivity) apply in general to all reactions, and suggest that manifestations of this should be looked for in the case of bimolecular reactions. It is the programme of our research group to start to fill in this gap in knowledge.

### 1.3 Comparison Between Unimolecular and Bimolecular Reactions of BC

The following table illustrates the expected similarities and differences between uni- and bimolecular reactions.

	Unimolecular Reaction $BC + M \rightarrow B + C + M$	Bimolecular Reaction $A + BC \rightarrow AB + C$
Competition between energy transfer and reactivity	Yes	Yes
Distortion of population distribution	Yes	Yes
$k$	$< k_{eq}$	$< k_{eq}$
[collider]	$[M] = \text{constant}$	$[A](t)$
$E_0$	$\gg k_B T$	no restriction
Important levels which react	only those close to the dissociation limit	any level can react with A
Scientific activity	well-studied	hardly studied at all

## 1.4 Non-Equilibrium Kinetics of Bimolecular Reactions

### 1.4.1 What are the Failings of Classical Chemical Kinetics?

Classical chemical kinetics is based upon two or three fundamental equations which are used (almost) universally without question. For the reaction  $A + BC \rightarrow AB + C$

$$\text{Rate} = k[A][BC]. \quad (1.6)$$

If the reverse reaction is important

$$\text{Rate} = k_f[A][BC] - k_r[AB][C]. \quad (1.7)$$

Here  $k_f$  and  $k_r$  are the rate coefficients for the forward and reverse processes respectively. The kinetic mass-action law relates  $k_f$  to  $k_r$  by:

$$\frac{k_f}{k_r} = K_c \quad (1.8)$$

where  $K_c$  is the equilibrium constant.

These fundamental equations are predicated upon the implicit assumption that the internal energy states of BC are in a Boltzmann distribution during the reaction. From the above considerations this is not strictly correct.  $k < k_{eq}$ , and this distortion is dependent upon the chemical environment which determines how efficient vibrational (say) excitation is. So,  $k$  is really dependent upon the concentrations of all species present, some of which may be varying with time. In other words,  $k = k([A](t), [BC](t), \dots)$  and is not a true constant, preventing direct integration of (1.6) for use in analysing experimental data. Even if the form of equation 1.6 were correct, it is not obvious that under conditions of non-equilibrium distributions, forward and reverse rates are additive. Furthermore, the distortion to  $k_f$  may not be of the same magnitude as to  $k_r$ , so that (1.8) is not strictly correct either.

### 1.4.2 Manifestations of the Failings of Classical Kinetics

It has already been mentioned above that  $k$  differs from  $k_{eq}$  by varying amounts depending on the temperature. This could result in non-Arrhenius behaviour. Theories of chemical reaction rely on the equilibrium assumption and fine-tune themselves by trying to reproduce experimental results. Curvature of such plots might be falsely attributed to special features of the transition state or to quantum mechanical tunneling rather than to simple non-equilibrium effects. Consequently theoretical approaches might be led astray.

Consequences for experimental kinetics are no less severe. It may legitimately be asked if the rate coefficients in the literature, for the elementary reactions, are correct. Are they  $k_{eq}$ , or somewhat less than  $k_{eq}$ ? Were they determined under conditions where non-equilibrium effects were absent? May they be used to predict the behaviour of complex systems such as combustion or atmospheric processes? In other words, even if  $k_{eq}$  for an elementary process were properly extracted, is it appropriate to use it for simulating a complex process in which non-equilibrium effects are likely to show up? Certainly the chemical environment of such systems is highly varied. Furthermore the products of elementary reactions may be produced in strongly non-Boltzmann distributions, especially if exothermic. Consequently, secondary reactions may not even proceed initially from a Boltzmann distribution. Each subsequent step's effect is compounded.

This thesis is designed to produce a program which allows our research group to explore conditions and give guidelines to chemists for deciding when non-equilibrium effects are important in bimolecular reactions.

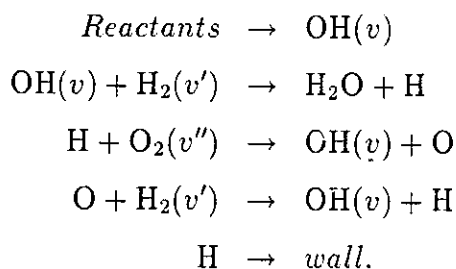
### 1.4.3 Implications for Kinetic Modeling

If the vibrational/rotational substructure is not ignored, then a simple mechanism is transformed into a nightmare. It becomes necessary to measure or calculate many state-selected rate constants instead of a single thermal rate coefficient for each elementary reaction. It becomes necessary to solve a far more complicated set of simultaneous differential rate equations than before. Rather complicated (less

transparent) dependencies of species concentration on chemical environment will result.

#### 1.4.4 Example

The first explosion limit for the  $\text{H}_2/\text{O}_2$  reaction is considered to consist of the following steps (here only the vibrational structure is explicitly mentioned):



Even if only 3 levels for each of  $\text{H}_2$ ,  $\text{O}_2$  and  $\text{OH}$  are considered as playing a role then there are 31 different rate constants to determine, and 12 differential equations to solve, some of which have a large number of terms (e.g. 28 for  $\text{H}$ ). This is indeed a more difficult problem to solve than the classical problem.

At this point a choice can be made regarding tactics. One can try to revert to the 5-equation approach, accounting for non-equilibrium effects by using an improved analytical rate law for each elementary step as does Teitelbaum[5]; or else one can rigorously solve the 12 simultaneous differential equations. This study takes the rigorous approach. It makes one less assumption than the analytical approach does—namely it does not assume the existence of a steady state. It follows the entire time history of the evolution of population distributions, and so it can be used more generally. In this study it is specifically used to compare with the analytical expression of ref. [5], and then to learn some details of the reaction  $\text{Br} + \text{HCl} \rightarrow \text{HBr} + \text{Cl}$ .

## 1.5 Previous Numerical Studies

There has been very little work done which solves the master equation for this class of reactions numerically.

The first study of this type was done by Porter *et al.*[6] who considered the reaction



at 400 K. They solved a vibrational level master equation numerically, and used quasiclassical trajectory calculations to obtain the rates for the reactive process. They found that reaction proceeds predominately from  $v = 5$  initially but that the population in this level is rapidly depleted. Thus actually about 56% of reaction proceeds from  $v = 4$  but the authors did not state what effect this depletion has on the phenomenological rate constant,  $k$ .

Later Finkelman *et al.*[7, 8] addressed the validity of the phenomenological rate law for the general reaction



A vibrational ladder model was also employed in this study (in which reactive and non-reactive processes were considered). The system was composed of structureless atoms (X), diatomic molecules (YZ), products (XY, Z), and an inert gas (M). The reactant diatomic was treated as a truncated harmonic oscillator with 4 or 7 vibrational levels. Landau-Teller scaling was used to obtain the energy transfer rate constants which were also assumed to be proportional to  $[\text{M}]$ . The master equation for the system was integrated using a numerical integrator based on Gear's method[9] (as does the present study). They found that after a transient period the system obeys a second-order phenomenological rate law and that for some conditions the phenomenological rate coefficient averaged over the course of reaction represented the observed rate constant, whereas for extreme conditions (for example, high temperature) it did not. In the most extreme case they found that the final value of the phenomenological rate coefficient was almost 1000 times smaller than the maximum value at  $t = 0$ . They also found that the phenomenological rate coefficient depended on the initial conditions of the experiment simulated.

At about the same time, Gutkowitz-Krusin[10] attempted to understand how the phenomenological rate equation arises from the underlying microscopic description by studying a simple 3-component multistate model for the general reaction 1.10. In

this model the reactant has only two states and the product has one. Only collisions of the type  $X-Y^*$  and  $XY-Z$  were considered. Gutkowitz-Krusin found that after a transient period the phenomenological rate equation always holds but that the transient period can be a macroscopically long time and that the duration of the transient period, as well as the magnitude of the rate constant, depended on the conserved quantities, such as the sum  $[X] + [XY]$ .

More recently, Lim and Truhlar[11, 12] reported a more realistic calculation for the system



at 300 K in an excess of Ar. Their differential rate equation accounted for state-to-state reactive processes and for simultaneous intramolecular vibration-rotation energy transfer as well as intermolecular vibration-vibration energy transfer processes and analogous processes for the reverse reaction. The phenomenological rate coefficient for this system was calculated by numerical integration of the rate equations using a method derived from the work of Gear[9], as well as by iteration to a quasisteady state and by eigenvalue analysis of a linearized master equation. They found multiple quasisteady states and, therefore, that the phenomenological rate law did not hold with a single pair of rate constants (one for forward reaction and one for reverse) throughout the reaction. They claim that the rate coefficient was distorted by no more than 1%, though, under typical experimental conditions. However, with the myriad of input rate constants involved, it is extremely difficult to “see the forest for the trees”. Moreover, they chose a temperature of 300 K and they concentrated on the exothermic direction. Both of these choices were self-defeating and led to negligible non-equilibrium effects. Notwithstanding these shortcomings this reaction system is a fascinating one because of the potentially large effects. Further study is warranted if one can concentrate on the endothermic direction,  $\text{Br} + \text{HCl} \rightarrow \text{HBr} + \text{Cl}$ , and on higher temperatures, as well as on the detailed time history of the individual vibrational levels of HCl. Spectacular effects can be expected.

Teitelbaum[13] solved the master equation for bimolecular exchange reactions

by treating the molecule as a harmonic oscillator truncated at the level from which reaction is primarily occurring. He obtained closed-form expressions for calculating the non-equilibrium correction factor to the rate coefficient over a wide range of temperatures and reagent concentrations—but only if the reaction can be isolated from secondary or back-reactions. Later work[5, 14] allowed reaction to occur from any level.

Other theoretical but less closely related work using techniques such as eigensystem analysis has been done by Present and Morris[15], Widom[16, 17], Snider and Ross[18], Shizgal[19] and Poulsen[20].

## Chapter 2. Methodology

### 2.1 The Master Equation Approach

The master equation (1.3) describes the rate of change of concentration of molecule BC in a particular internal energy state when BC undergoes unimolecular reaction and energy transfer. The same formalism can be used for bimolecular reactions if  $k(E)[BC(E, t)]$  is replaced by  $k_v[A][BC(v, t)]$ . This study adopts the philosophy of addressing the basic physics of the problem and leaves refinements to future work. BIMSIM can be easily extended to solve more general problems but is at the moment restricted to accommodate only those processes for which rate constants can realistically be expected to be found in the literature. Thus, for now, only single quantum energy transfer processes are considered. For purely vibrational processes this means that reaction 1.1 can be decomposed into the following elementary processes:



where  $BC(v)$  is diatomic species BC in vibrational level  $v$ , A is an atom or radical and R represents all relaxer species present (possibly including A, BC, inert diluent M, etc.). Then the differential equation describing the rate of change of the population,  $N_v$ , of BC in level  $v$  is given by

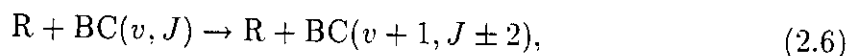
$$\begin{aligned} -\frac{dN_v}{dt} = & k_{v,v-1}N_vN_R + k_{v,v+1}N_vN_R \\ & - k_{v-1,v}N_{v-1}N_R - k_{v+1,v}N_{v+1}N_R + k_vN_A N_v. \end{aligned} \quad (2.4)$$

In equation 2.4 each term containing  $N_R$  is actually a sum of terms such that

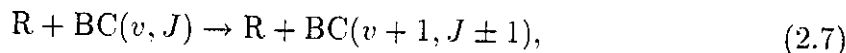
$$k_{i,j}N_R \equiv k_{i,j}^A N_A + k_{i,j}^{BC} N_{BC} + k_{i,j}^M N_M + \dots \quad (2.5)$$

(that is, one term per relaxer species present); the rate constants for processes 2.1, 2.2 and 2.3 are  $k_v$ ,  $k_{v,v-1}$  and  $k_{v,v+1}$ , respectively;  $n$  is the uppermost vibrational level for which data are available; and terms of the form  $N_X$  represent the concentration of species X. The set of such equations for  $v = 0, 1, 2 \dots n$  is called the master equation for the reaction. This master equation can be solved numerically (that is,  $N_v$  for  $v = 0, 1, 2 \dots n$  as a function of  $t$  can be generated). An application of this approach is found in Chapter 3.

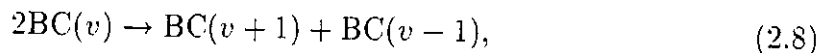
Rather than allowing only VT (that is, vibrational-translational) relaxation processes (equations 2.2 and 2.3) one can also allow VRT processes (in this context, R for rotational) such as



or



for homonuclear and heteronuclear BC respectively, and VV processes such as



as well as all of the analogous processes for the reverse reaction. As more and more terms are introduced into the master equation to account for these other processes, the master equation becomes less and less of an approximation to reality. Currently, BIMSIM allows for processes 2.1, 2.2, 2.3, 2.6 but not 2.7 nor for VV processes and not for any of the processes for the reverse of reaction 2.1. On the surface, this seems to restrict BIMSIM to reactions with a slow reverse process, but in actual fact it is not so restrictive. It really means that the reverse reaction is slow enough not to interfere with the establishment of a steady forward process. In the example described in this study,  $\text{Br} + \text{HCl}$ , the reverse process is very fast, but like most other bimolecular reactions, it does not interfere with the details of the forward reaction[13]. Notwithstanding this, in keeping with the policy stated above, it is important to study the forward reaction in isolation in order to establish a baseline for comparison with the more complex case.

## 2.2 Gear's Method

Some systems of ordinary differential equations (o.d.e.'s) are called "stiff". This term has a formal mathematical definition (see, for example [21]), but of more practical concern is that o.d.e.'s (often describing physical systems) that locally behave exponentially and that have components decaying at very different rates are known to be stiff. The master equation approach described in the previous section is such a system. Conventional numerical methods for solving o.d.e.'s are very inefficient and have a tendency to be unstable when used to solve stiff o.d.e.'s. The method chosen by the author which is a commercial implementation of the work of Byrne and Hindmarsh[22] (which in turn is based on the work of Gear[9, 23]) was developed to handle stiff o.d.e.'s. Gear's method is an extremely efficient variation of the Adams predictor-corrector formulas. The algorithm adjusts the order and step size to allow the integration to proceed as quickly as possible while still keeping the estimated error within user-requested limits. It represents the state of the art in the numerical solution of systems of stiff o.d.e.'s[24].

## 2.3 Testing of the Numerical Method

BIMSIM was tested for accuracy in several ways. The initial step size and global error could be controlled by the user. In this study both the initial step size and tolerance parameter space were explored to find sufficiently stringent (small) values such that the end result was invariant. Values of these parameters were found which allowed the end result to be invariant to better than 1 part in  $10^6$ .

Additionally one other numerical test was attempted. It was possible to check for the *chemical* accuracy by comparing the results of  $\sum_v N_v$  (i.e., the sum of the integral of equation 2.4) with the integral of

$$\frac{dN_A}{dt} = \sum_v k_v N_A N_v. \quad (2.9)$$

The two results were identical within 1 part in  $10^{10}$  by the end of the integration, indicating that round-off errors did not accumulate to serious levels, and that the

reagents were being converted into products as chemically required.

## Chapter 3. Results and Discussion

### 3.1 The Br + HCl Reaction

In order to test the validity and demonstrate the use of the program a reaction system must be found for which appropriate input data is available and for which suitably detailed calculations have been done. There are several such systems available[5, 14]. The system



was studied by Teitelbaum *et al.*[5] who recently solved the steady-state master equation describing bimolecular exchange reactions and used the analytical expression thus obtained for the phenomenological rate coefficient to calculate it under various conditions. This system is also used in the present study for the purpose of testing BIMSIM against the analytical result and to extend Lim and Truhlar's study to the more interesting endothermic direction, and to more interesting temperatures. BIMSIM is sufficiently flexible that it can calculate [HCl] as a function of time using the same approximations and initial conditions as those of the above study. This is done in order to a) see if the same results expected from the steady-state calculations are obtained, at sufficiently long times, and to b) see if the steady-state approximation of ref. [5] is valid and if so under what conditions. Other approximations are as follows: HCl is treated as a Simple Harmonic Oscillator truncated at  $v = 4$ ; only VRT energy transfer processes are considered; Landau-Teller scaling of energy transfer rate constants is employed and the species are in thermal equilibrium initially. Similarly, the initial conditions are: a range of five temperatures from 300–3000 K; a choice of species, R, responsible for vibrational relaxation (either He or HCl); and a choice of three mole fractions (that is  $[\text{Br}]/[\text{R}]$ ) from 0.01–1.0. The analytical solution[5] further assumed that either He or HCl was the collider

but not both. In the case of the numerical solution, a finite amount of HCl must be present even if  $R = \text{He}$ , but the amount of HCl present must be such that it does not contribute significantly to relaxation in order that the present study can be validly compared to the analytical study. Therefore, whatever value is chosen for  $[\text{HCl}]$ , the ratio  $[\text{HCl}]/[\text{He}]$  must be sufficiently small—at least as small as  $k_{10}^{\text{He}}/k_{10}^{\text{HCl}}$  or  $2.8 \times 10^{-3}$ . Through a process of trial and error it was determined that a value of  $[\text{HCl}]/[\text{He}] = 10^{-6}$  is sufficiently small.

Values for reaction rate constants and relaxation rate constants used as input to the program are given in Tables 3.1 and 3.2. These input parameters are the same as those used to obtain  $k$  analytically[5, 25].

Table 3.1: Relaxation Data[5]

$T/\text{K}$	$k_{10}^{\text{HCl}}/\text{dm}^3\text{mol}^{-1}\text{s}^{-1}$	$k_{10}^{\text{He}}/\text{dm}^3\text{mol}^{-1}\text{s}^{-1}$
300	$1.44 \times 10^7$	$4.10 \times 10^4$
500	$2.04 \times 10^7$	$5.31 \times 10^5$
1000	$1.05 \times 10^8$	$1.75 \times 10^7$
2000	$8.91 \times 10^8$	$2.66 \times 10^8$
3000	$3.0 \times 10^9$	$9.52 \times 10^8$

Table 3.2: Microscopic Reaction Rate Constants at all Temperatures

$v$	$k_v/\text{dm}^3\text{mol}^{-1}\text{s}^{-1}$
0	$1 \times 10^{-2}$
1	$2.0 \times 10^4$
2	$1.0 \times 10^9$
3	$7.0 \times 10^9$
4	$1 \times 10^{10}$

## 3.2 Data Reduction and Results

The phenomenological rate law for reaction 3.1 is

$$rate = k[Br][HCl] \quad (3.2)$$

where  $k$  is the phenomenological rate coefficient for the reaction. Therefore, at any given time,  $k = rate/([Br][HCl])$  and since the rate is defined as  $-d[HCl]/dt$ ,

$$k = -d[HCl]/dt/([Br][HCl]). \quad (3.3)$$

Although not obvious, it can be shown[5] that  $k$  is made up of contributions from each vibrational level of HCl. Thus

$$k = \sum_{v=0}^n y_v k_v. \quad (3.4)$$

Here,  $n = 4$ ,  $y_v$  is the fractional population in level  $v$ , and  $k_v$  is the level-selected rate constant for reaction 2.1 from level  $v$ . This is a quantity that can be readily calculated by BIMSIM. It turns out that  $y_v$  and hence  $k$  are dependent on time and initial conditions. To compare results with those of the analytical results[5] a single value of  $k$  must be chosen for each set of initial conditions. Figures 3.1 and 3.2 are plots of  $k$  normalized by its initial value  $k(0)$  for the six sets of initial conditions at  $T = 300$  K. Note that the abscissa is actually  $\log \xi$  where  $\xi$  is the extent of reaction. Extent of reaction is here defined as  $([Br]_0 - [Br])/[Br]_0$  if Br is the limiting reagent, or  $([HCl]_0 - [HCl])/[HCl]_0$  if HCl is the limiting reagent, with  $[Br]_0$  and  $[HCl]_0$  being the initial reagent concentrations. This choice of abscissa allows for all the significant events that occur over the course of the reaction to be displayed on a single graph and also makes it easy to compare the results obtained from one set of initial conditions with those obtained from another. As can be seen from Figures 3.1 and 3.2,  $k/k(0)$  is a relatively simple function; it starts out at unity and decreases monotonically to a smaller constant value; at later time the value of  $k/k(0)$  stays the same (Figure 3.1) or increases to its equilibrium value  $k_{eq}/k(0)$  (Figure 3.2), depending on the initial conditions. ( $k_{eq}$  corresponds to a Boltzmann

distribution. So  $k_{eq} = k(0)$ .) The increase back to the equilibrium value is discussed below. Teitelbaum's steady-state distribution of  $y_v$  is defined when  $\dot{y}_v \approx 0$ . Because of the linear relation between  $y_v$  and  $k$  (equation 3.4) steady state corresponds to times after which  $\dot{k} = 0$ . Hence, the obvious value to choose for comparison is the minimum value of  $k/k(0)$  obtained for each set of initial conditions. The results are presented in Table 3.3. The graphs of  $k/k(0)$  vs.  $\log \xi$  for all initial conditions are given in Appendix B.

### 3.3 Comparison with Other Work

Since the analytical results of Teitelbaum *et al.* are mathematically exact, at least within the framework of the approximations given above, and since the same approximations were used in the present study it is expected at first glance that values obtained for  $k$  in the two studies should be identical. This is, in fact, not the case; while the agreement obtained might be considered excellent (within 1 part in  $10^5$  in most cases) it is not exact. There are two reasons why this is not surprising. (1) The analytical results are based on the steady-state assumption, which, as the results in Appendix B show, is not valid at all times. A steady reaction begins only after a finite time. (2) Furthermore, the analytical results shown in Table 3.3 were obtained using initial reagent concentrations and do not account for the disappearance of Br (or HCl). Consequently there is only a limited time interval where the analytical results are applicable. This time interval can be so short (see especially Figure B.19) that the value of  $k/k(0)$  can be distorted, in principal, from the steady-state value. Alternatively it is possible that under some conditions the reaction is so fast that a steady distribution is not established in time, i.e., the reaction could proceed during the transient phase. Therefore we are not searching for exact agreement but rather for reasonable agreement. The results agreed for the most part to better than 1 part in  $10^5$  and where they did not the analytical results were always smaller than the numerical results. Furthermore, the cases where they did not agree corresponded to the cases (such as the one in Figure B.19) where the  $k/k(0)$  curve decreased, and, rather than forming a plateau, reached a minimum and immediately increased

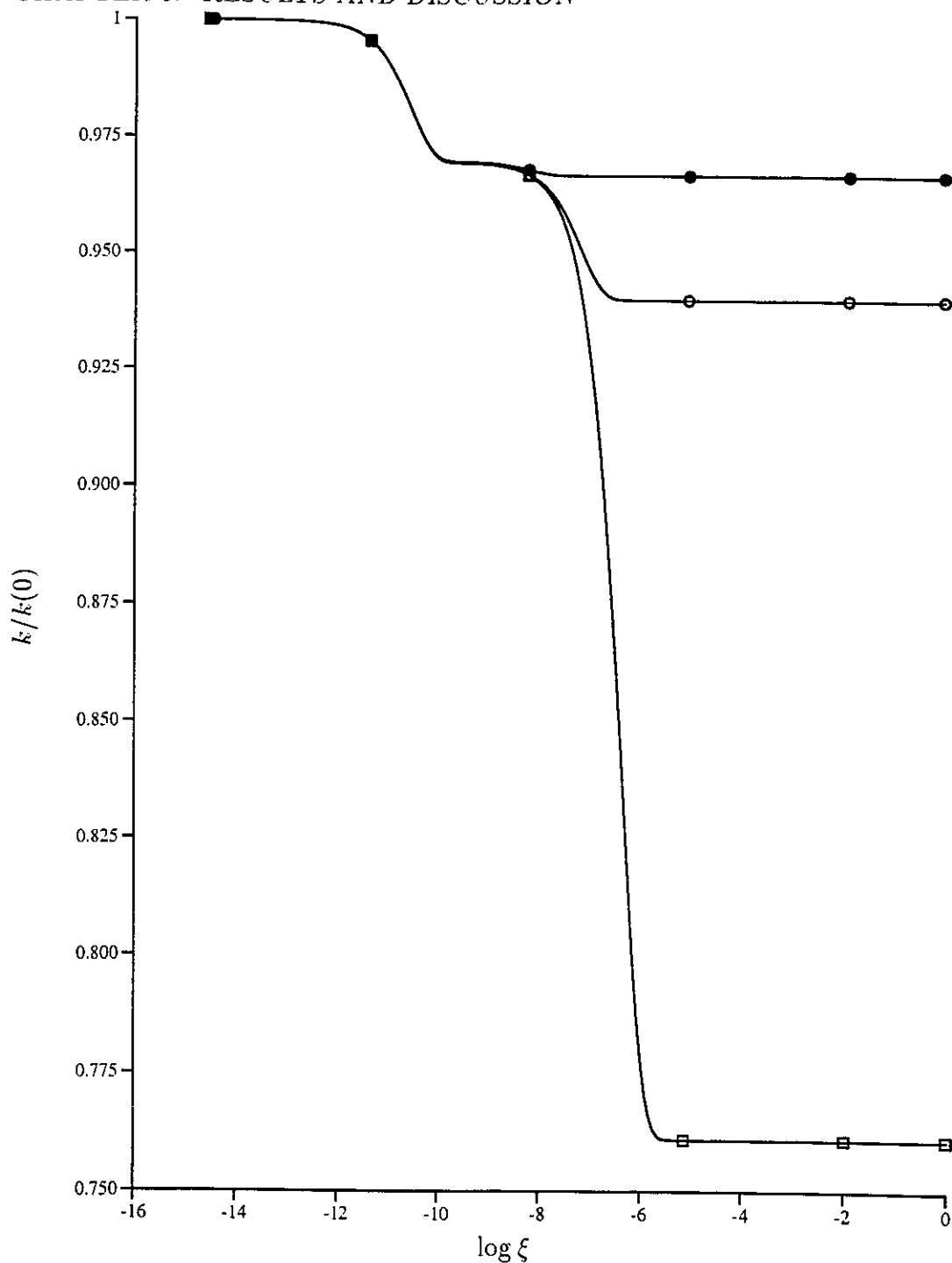


Figure 3.1: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $\text{R} = \text{He}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : (●) 0.01, (○) 0.1, (□) 1.0.  $\xi$  is the extent of reaction as defined on page 18. Representative points are shown merely to distinguish among curves.

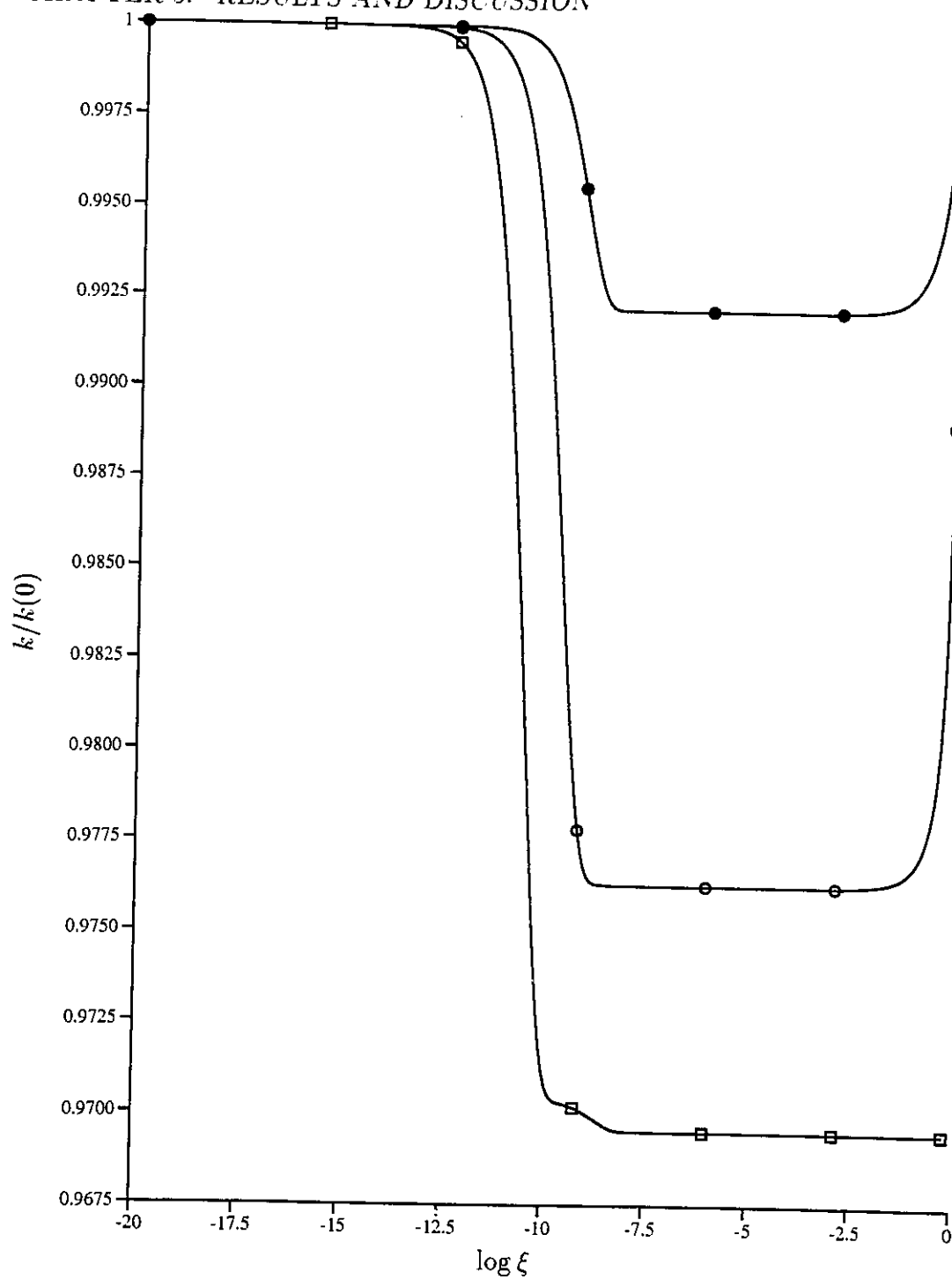


Figure 3.2: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $\text{R} = \text{HCl}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0.  $\xi$  is the extent of reaction as defined on page 18. Representative points are shown merely to distinguish among curves.

Table 3.3: Comparison of Numerical and Analytical Results

R	T/K	[Br]/[R]	$k/k(0)$	HT Results[25]
He	300	0.01	0.96672	0.96672
		0.1	0.939979	0.939975
		1	0.76083	0.76083
	500	0.01	0.16409	0.16409
		0.1	0.084968	0.084968
		1	0.073802	0.073802
	1000	0.01	0.72021	0.72021
		0.1	0.22937	0.22937
		1	0.030605	0.030605
	2000	0.01	0.90912	0.90912
		0.1	0.55283	0.55283
		1	0.15738	0.15738
	3000	0.01	0.9547148	0.9547150
		0.1	0.690782	0.690786
		1	0.22999	0.22999
HCl	300	0.01	0.99215	0.99215
		0.1	0.97628	0.97628
		1	0.96953	0.96953
	500	0.01	0.81716	0.81716
		0.1	0.34289	0.34289
		1	0.11164	0.11164
	1000	0.01	0.93196	0.93134
		0.1	0.61890	0.61747
		1	0.153344	0.153346
	2000	0.01	0.97308	0.97032
		0.1	0.79318	0.78081
		1	0.33730	0.33731
	3000	0.01	0.98748	0.98510
		0.1	0.88672	0.87085
		1	0.43926	0.43926

again; i.e., where significant reaction took place during the transient phase.

### 3.4 Internal Validation

It is possible to adjust the input parameters to BIMSIM in various ways to check that the behavior obtained agrees with the behavior expected. Three such ways were employed.

First, the absolute concentration of the species was changed (keeping the relative concentrations the same) over 12 orders of magnitude. The expectation (at least computationally) is that there will be no change in the value obtained for  $k$ , or in the shape of the curve for  $k(t)$ . (In fact, initially the results were not concentration-independent and this led to the discovery of a rather subtle bug in the program.) Second, the value of  $k_{10}$ , the energy transfer rate constant for the transition from  $v = 1$  to  $v = 0$ , can be made larger and larger. As the relaxation rate gets larger and larger, there is less and less depletion of the high vibrational levels caused by reaction from those levels and, therefore, the value of  $k$  obtained should approach  $k_{eq}$  and the  $k(t)$  curve should approach the trivial straight line  $k = k_{eq}$  if the program is working correctly. Third, the ratio  $[\text{He}]/[\text{HCl}]$  can be varied from 0 (where HCl is the species principally responsible for relaxation) to  $10^6$  (where He is the principal relaxer). In this case, it is expected that the value of  $k$  obtained and the shape of the  $k(t)$  curve will make a smooth transition from the R = HCl behavior to the R = He behavior. In all three cases, the results obtained agree with the expected results.

The temperature dependence of  $k/k(0)$  also turned out as expected, with the value becoming smallest at temperatures  $\sim 500$  K[5]. It is at around 500 K that the disparity between  $k_{10}$  and  $k_{eq}$  becomes largest. The disparity is smallest at 300 K, thus explaining why Lim and Truhlar found only 1% effects for their room-temperature study[12].

### 3.5 Time Dependence of $k$

The time dependence of  $k$  can be explained completely (at least qualitatively) in such a way that no features of a plot of  $k$  vs.  $t$  remain unexplained. This explanation also serves to further validate the program.

All of the features of such curves can be explained by looking at plots of  $y_v/y_v(0)$  vs.  $\log \xi$ , where  $y_v$  is the amount of HCl in level  $v$  relative to the total amount of HCl present, and  $y_v(0)$  is the value of  $y_v$  at  $t = 0$ . In order to best visualize the features of such plots some function of  $y_v/y_v(0)$  is chosen as the ordinate rather than  $y_v/y_v(0)$  itself. Figures 3.3 and 3.4 show how the fractional populations of each of the five vibrational levels (corresponding to  $v = 0-4$ ) varies as a function of time (actually as a function of  $\log \xi$ ). As with Figures 3.1 and 3.2, the figures are for the six sets of initial conditions for which  $T = 300$  K. The features of the plot shown in Figure 3.1 for  $R = \text{He}$  and of the plot shown in Figure 3.2 for  $R = \text{HCl}$  are significantly different and will be examined separately in the next two sections.

#### 3.5.1 $R = \text{He}$

For  $R = \text{He}$  (Figure 3.1) there are three features that require explanation: the dip at  $\log \xi = -11$ ; the second dip at  $\log \xi = -7$ ; and the fact that the curves for the three mole fractions remain the same until  $\log \xi = -8$  and then split showing three final values of  $k/k_{eq}$ .

The first feature is easily seen from Figure 3.3 as the effect of the depletion of  $y_2$ ,  $y_3$  and  $y_4$  at  $\log \xi = -11$ , and the second feature also is easily seen as the effect of the depletion of  $y_1$  at  $\log \xi = -7$ . The third feature is harder to explain. However, close examination of the  $y_2$  curve and especially the  $y_3$  and  $y_4$  curves in Figure 3.3 reveals that these levels are seriously depleted (by as much as a factor of  $10^{15}$  in the case of  $y_4$ ) and so all these values decrease at  $\log \xi = -11$  to the point where none of them makes any contribution to the phenomenological rate coefficient,  $k$  (and therefore the contribution made for all three mole fractions is the same). Note that for the conditions existing when  $R = \text{He}$ , i.e.,  $[\text{HCl}] \ll [\text{He}]$  such that  $[\text{HCl}]$  is also

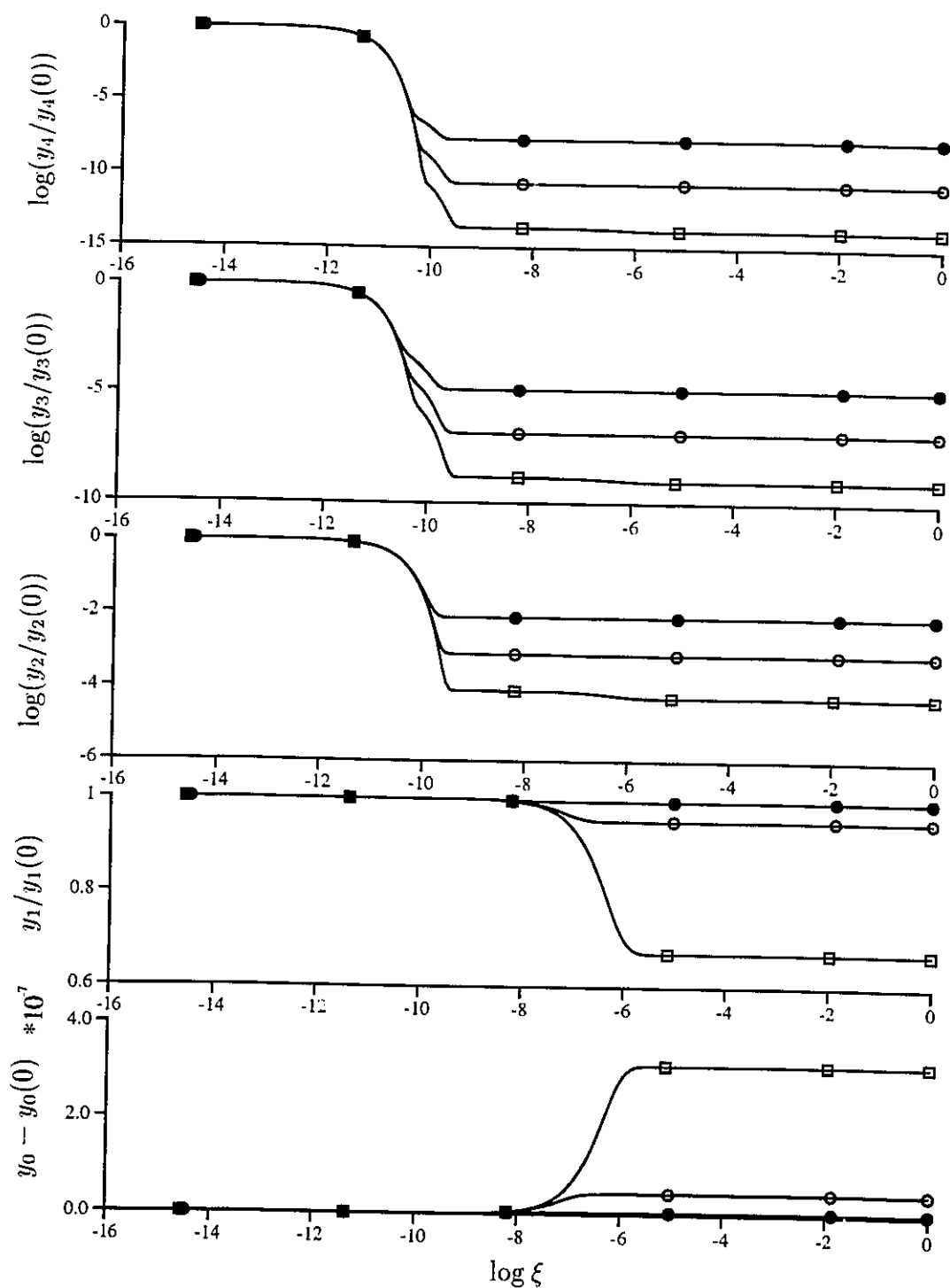


Figure 3.3: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{He}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

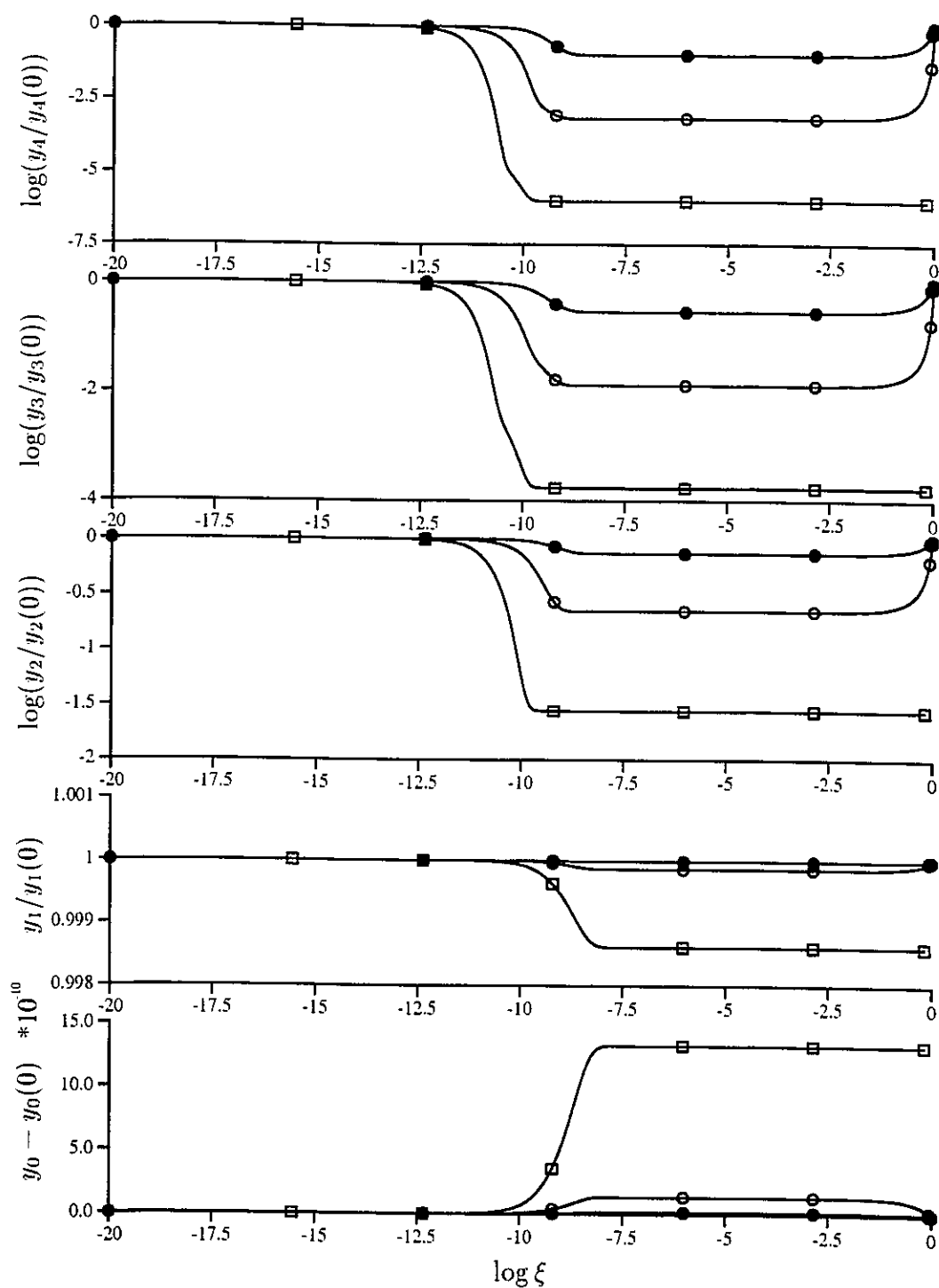


Figure 3.4: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{HCl}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : (●) 0.01, (○) 0.1, (□) 1.0

$\ll [\text{Br}]$  (see p.17) then  $[\text{Br}]$  hardly changes as the limiting reagent  $\text{HCl}$  disappears. Consequently the relative rate of reaction and relaxation remains constant at all times, and the steady state values of  $k$  and of  $y_v$  show an exceptional constancy that is not in general a necessary feature of steady states (see below).

### 3.5.2 R = HCl

For  $\text{R} = \text{HCl}$  (Figure 3.2) there are again three specific features that require explanation: the shoulder on the curve for  $[\text{Br}]/[\text{R}] = 1$  at  $\log \xi = -9$ ; the difference in behavior between this curve and the other two at  $\log \xi = -1$ ; and why these curves separate much earlier ( $\log \xi = -13$ ) than the corresponding  $\text{R} = \text{He}$  curves.

The  $y_2$ ,  $y_3$  and  $y_4$  curves of Fig. 3.4 clearly cause the bulk of the change in  $k$  at  $\log \xi = -10$  (the first feature). Examination of the  $y_1$  curve shows that (relatively speaking) the depletion for the  $[\text{Br}]/[\text{R}] = 1$  curve is much larger than the other two curves and this depletion means that  $y_2$  makes a slightly smaller contribution to  $k$  after  $\log \xi = -9$  than before. The third feature is again a direct reflection of the behavior of the  $y_1$ ,  $y_2$ ,  $y_3$  and  $y_4$  curves of Fig. 3.4. For the  $[\text{Br}]/[\text{R}] = 1$  curve,  $[\text{Br}] = [\text{HCl}]$  throughout the course of the reaction but for the other curves,  $\text{HCl}$  is in excess. This means that as soon as a finite amount of reaction has taken place ( $\log \xi = -1$ ), the amount of  $\text{Br}$  begins to decrease but the amount of  $\text{HCl}$  does not decrease significantly. However, reaction rates depend on  $[\text{Br}]$  ( $\text{rate} \propto [\text{HCl}(v)][\text{Br}]$ ) and the relaxation rates do not ( $\text{rate} \propto [\text{HCl}(v)][\text{HCl}]$ ). Therefore, the overall reaction rate decreases in proportion to the overall relaxation rate as time increases. This, in turn, increasingly allows relaxation to “catch up” to reaction. In other words, the reaction behavior becomes less and less non-equilibrium (that is, less and less depletion of levels  $v = 1, 2, 3, 4$ ) so that  $k \rightarrow k_{eq}$  as  $\log \xi \rightarrow 0$ . Note that, even though  $k \rightarrow k(0)$  and  $y_v \rightarrow y_v(0)$  at late times, this is still the steady-state regime, since once a steady distribution is established then a steady state prevails thereafter even if reagents disappear significantly. A steady state  $k$  does not necessarily mean a constant  $k$ . For example Teitelbaum’s steady state expression still applies, as long as the appropriate reagent concentrations are input.

## Chapter 4. Conclusions and Future Work

### 4.1 Conclusions

BIMSIM has been validated at the same time showing that the analytical expression of [5] is valid. BIMSIM was used for extending the calculations of Teitelbaum to include time dependent details and for extending the calculations of Lim and Truhlar to the more interesting case of the higher temperature endothermic reaction  $\text{Br} + \text{HCl} \rightarrow \text{HBr} + \text{Cl}$ . It was found that non-equilibrium effects can be impressive. The magnitude depends on temperature,  $[\text{Br}]$ ,  $[\text{HCl}]$ , and  $[\text{He}]$ . The bulk of the reaction occurs in the steady-state regime, but interesting effects show up as the reagents begin to disappear.

It is very difficult to produce a computer program which is correct—that is, one which calculates what you intend it to calculate. In other words, one which does not produce output which is in error. Even when independently calculated or measured results are available, if they differ from the results of the program in question one cannot say with certainty which result is correct and which is not (assuming that not both results are incorrect!). One has a renewed respect and healthy scepticism at the same time for other existing computational work—especially more complicated work such as *ab initio* calculations.

### 4.2 Contribution to Knowledge

BIMSIM is a computational tool which can be easily used to learn the details of non-equilibrium processes in bimolecular reactions. It is clear that the non-equilibrium effect is very sensitive to the chemical environment. Experimentalists now have a tool to guide them to extract steady-state rate constants and to allow them to intelligently compare results with other workers for whom conditions may be significantly

different. Theoreticians are also guided to proper comparisons with experiments.

### 4.3 Future Work—Chemical Aspects

Work yet to be done can be broken down into three categories: further analysis of already obtained data; using BIMSIM's existing capabilities to explore other initial conditions or levels of approximation; and extending BIMSIM, adding features and removing currently existing approximations.

It would be useful to analyse the data already obtained to see if one can use the relative populations to calculate a single "vibrational" temperature (by assuming a Boltzmann distribution, obtaining a best fit to the data and looking at the statistical error associated with this best fit) as a function of time. This might, for example, lead to an analytical expression for  $T(t)$  and, therefore, to an improved explicit expression for the correction factor obtained by Teitelbaum[13]. It would also be useful to examine the contribution each level makes to  $k(t)$  more closely in order to answer questions such as can a single level,  $m^*$ , from which reaction is primarily occurring always be identified (this is an assumption of ref. [13] but not of ref. [5]).

It may be that if a particular vibrational level is selectively excited (through a laser pulse, for example) that a steady state is never established. Therefore, this is a particularly interesting set of initial conditions that needs to be explored. It can be done using BIMSIM's current capabilities.

Several desirable enhancements to BIMSIM can be itemized:

- Include back reaction, especially to allow for closer examination of conditions where  $k$  is still changing when  $\xi$  becomes large enough to allow significant back reaction.
- Build a sensitivity analysis capability into the program.
- Explore the possibility of using Groebner bases[26] as a way of merging existing analytical (e.g., the work of Teitelbaum *et al.*[13, 5, 14]) and theoretical treatments (e.g., this work).

- Include VV processes.
- Allow products AB and C to be relaxers.
- Calculate  $k_f/k_r$  to see if the kinetic mass action law is valid during non-equilibrium chemical reactions.
- Make a direct comparison, if possible, with Lim and Truhlar's method, using their input rate constants in BIMSIM.

#### 4.4 Future Work—Computational Aspects

The author would like to make various computational improvements to BIMSIM as follows:

- Modify BIMSIM so that it can generate a separate version of itself for each reaction to be explored. In this way, only the input parameters that will change within a series of experiments performed need be provided, all the others having been determined once and isolated in a separate input file.
- The input mechanism needs to be more robust to allow more natural input, more errors in input detected, and error recovery where possible.
- The code which prompts the user when in interactive mode needs to be rewritten to allow the program to be smaller and more easily extended.
- Convert the code from Fortran to FWEB (an extension of Fortran which allows the system documentation for the program to be automatically generated from the source code).

## **Appendix A. BIMSIM Description**

The important features of BIMSIM will be outlined in this appendix.

### **A.1 Program Flow**

BIMSIM is a conventional program in the sense that it begins by obtaining input parameters (routines which obtain such parameters mostly have names starting with the letters "PIK" in BIMSIM) to indicate what reaction system, what temperature, etc., then it proceeds to calculate what is required (in this case what is required is the integration of the master equation—the name of the routine here is SOLVER) and writes results as they are available to various files (routine DUMP).

### **A.2 Flexible Input**

BIMSIM has two ways of obtaining input parameters: the user can be prompted for input from the keyboard (interactive mode), or input can be from a file (batch mode). Or, in fact, input can be a combination of the two; if the user selects batch mode and end-of-file is reached before all parameters are specified then the program switches to interactive mode.

In any case, the program always echoes input to another file, formatted in such a way that the user can use this file as input on the next run. When using the program interactively, this allows the program to be interrupted and later restarted bringing the user back to the point where he left off. It also allows the use of interactive mode to build an input file through prompts from the program, making the program easy to use but not necessarily forcing the user to respond to a long series of prompts each time the program is run.

In fact, each prompt has a default value associated with it (obtained by hitting

<ENTER>) and so a valid input file can be built simply by selecting the default at each prompt. This mechanism has the added benefit that, if desired, a user of BIMSIM can begin using the program right away without having to read any documentation or having to create an input file (or even understanding what the program does!).

In addition, when the input expected by BIMSIM is a physical quantity that has units associated with it, the user must supply the units, but many different units are allowed by the program and converted to the equivalent SI units (the program always uses SI internally). For example, pressure can be specified in Torr, kiloPascals, atmospheres, p.s.i., and bars.

Another innovative feature of the program is that it has so-called novice and expert modes. If novice mode is selected, more information is given to the user at each prompt to help him make his selection—in particular the units which are allowed at a given prompt are displayed when in novice mode. In expert mode, this information is suppressed—presumably once a user has been using BIMSIM long enough he will not need this information cluttering up his screen and can request that it not appear.

BIMSIM is designed to handle a very wide range of initial conditions, experiments, reactions, and levels of approximation. There are no less than 115 separate prompts in the program; even if each prompt allows for only 2 branches this means there are  $2^{115}$  different situations that BIMSIM can handle. For example, when determining the energy level of each vibrational level, the user has six choices: SHO, 3-parameter Morse function, partial spectroscopic fit (vibrational constants only), file of  $v$ ,  $E(v)$  pairs, full spectroscopic fit, or a file of  $v$ ,  $J$ ,  $E(v)$  triples. In some of these cases the energies of the rotational sub-levels are not yet determined and the user is further prompted to choose from a rigid rotor approximation or a partial spectroscopic fit to determine these energies.

### **A.3 Awareness of Experimental Conditions**

The program easily allows the simulation of typical experimental conditions such as those of laser pulse experiments where one level is selectively excited by a laser from the ground vibrational level to some higher level. It also allows input data to be sparse and does various interpolations to provide any missing values, if requested. It recognizes that data may not be available in some cases, for example, even if BIMSIM calculates that 11 vibrational levels exist below the dissociation energy, it will allow for a model which has fewer than 11 levels (according to the data available).

### **A.4 Portability**

BIMSIM conforms to the FORTRAN77 standard. The code contains no system-dependent sections. A port to a new system requires the equivalent of a CMS EXEC be provided to link FORTRAN I/O units to external files (and to the terminal), and to execute the program (assuming it has been successfully compiled). BIMSIM also requires any system to which it is to be ported have available the IMSL subroutine library. The IMSL subroutine library is a commercial product available for many different systems from IMSL, Inc., 2500 CityWest Blvd., Houston, TX.

### **A.5 Design Aspects**

BIMSIM is designed to be easily modified by those who use it (and not just by the author). There are extensive comments throughout the program. It consistently uses modular structured programming techniques. In many cases, routines which take user-defined action are already built into the program—the user has only to modify the existing shell, inserting the desired code.

The program is also designed to be robust and correct rather than fast. For example, a routine called EXIST is used in many places to greatly simplify the coding: if a value is to be calculated for each existing transition the program does

something like (in pseudo-code)

if exist( $v, J, v', J'$ ) then calculate value

where  $v, J$  are the quantum numbers for the initial level and  $v', J'$  are the quantum numbers for the final level of a particular transition. Input is extensively validated and there are many checks in the program to make sure input data is reasonable and that no data are accidentally missing from the input.

## Appendix B. Graphs

In this appendix graphs of  $k/k(0)$  vs.  $\log \xi$  and graphs of various functions of  $y_v$  vs.  $\log \xi$  are presented for all the initial conditions specified in section 3.1. In each graph  $\xi$  is the extent of reaction as defined on p.18, and representative points are shown merely to distinguish among curves.

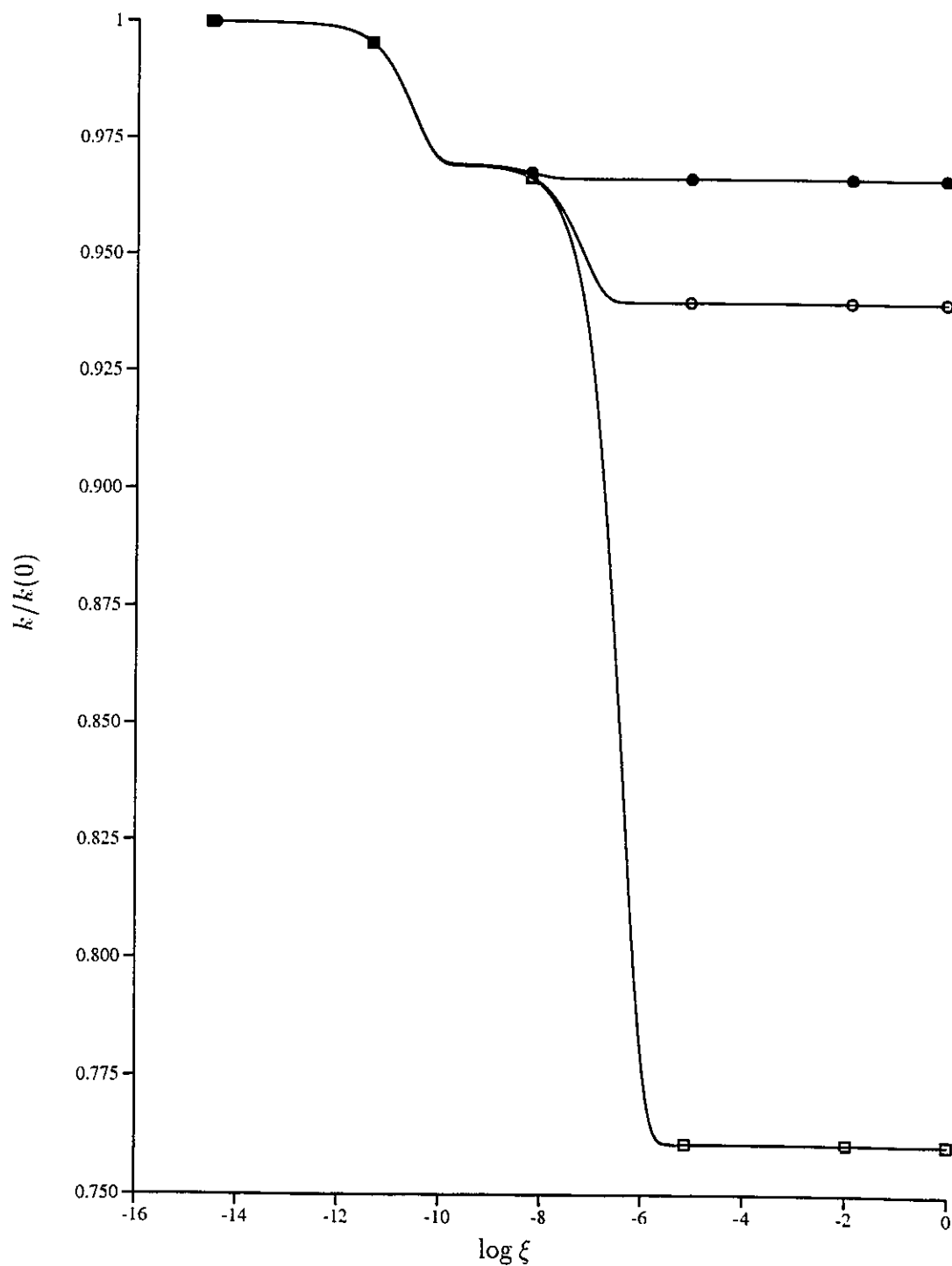


Figure B.1: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $\text{R} = \text{He}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

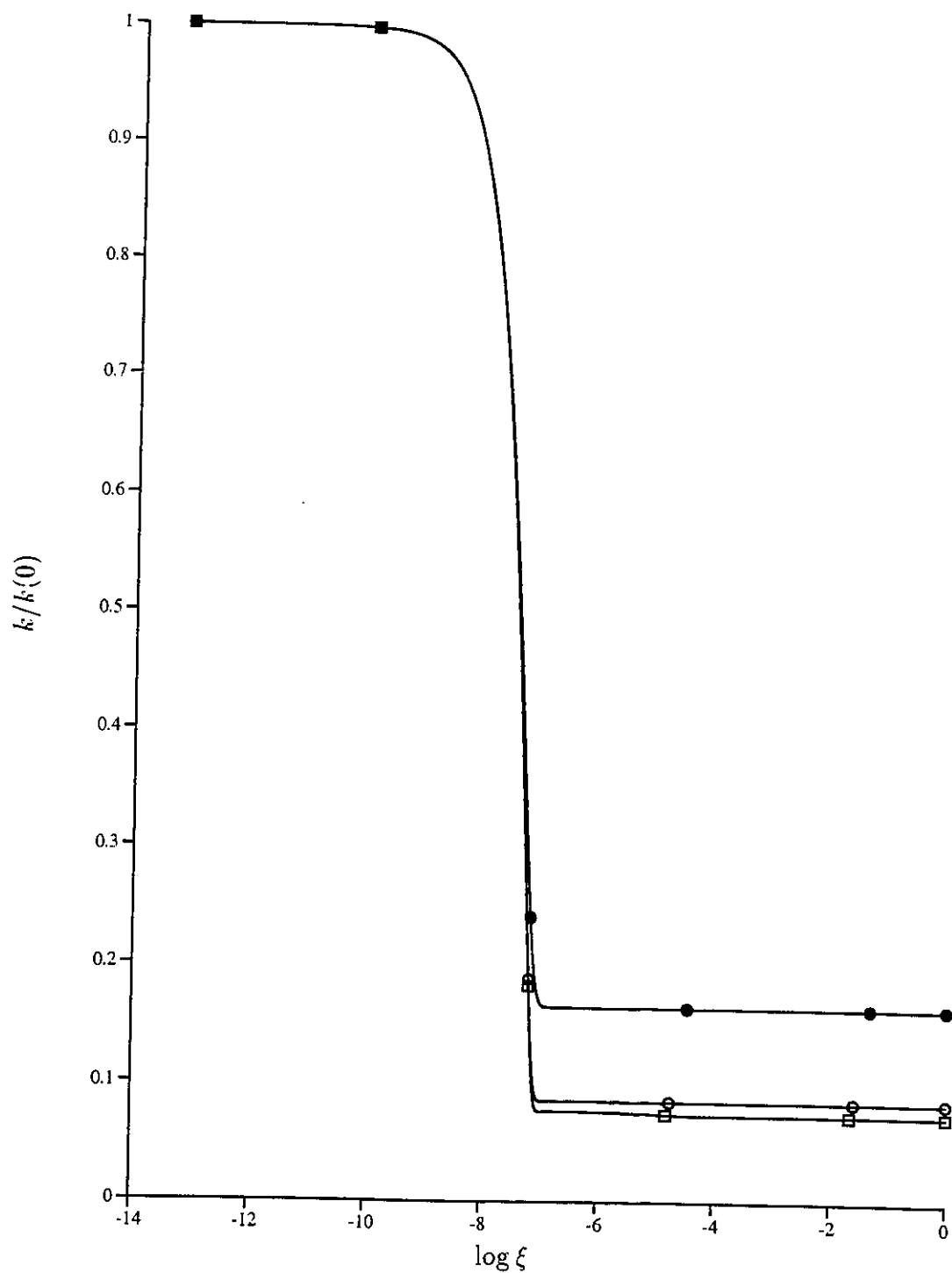


Figure B.2: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $R = \text{He}$ ,  $T = 500 \text{ K}$  and various values of  $[Br]/[R]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

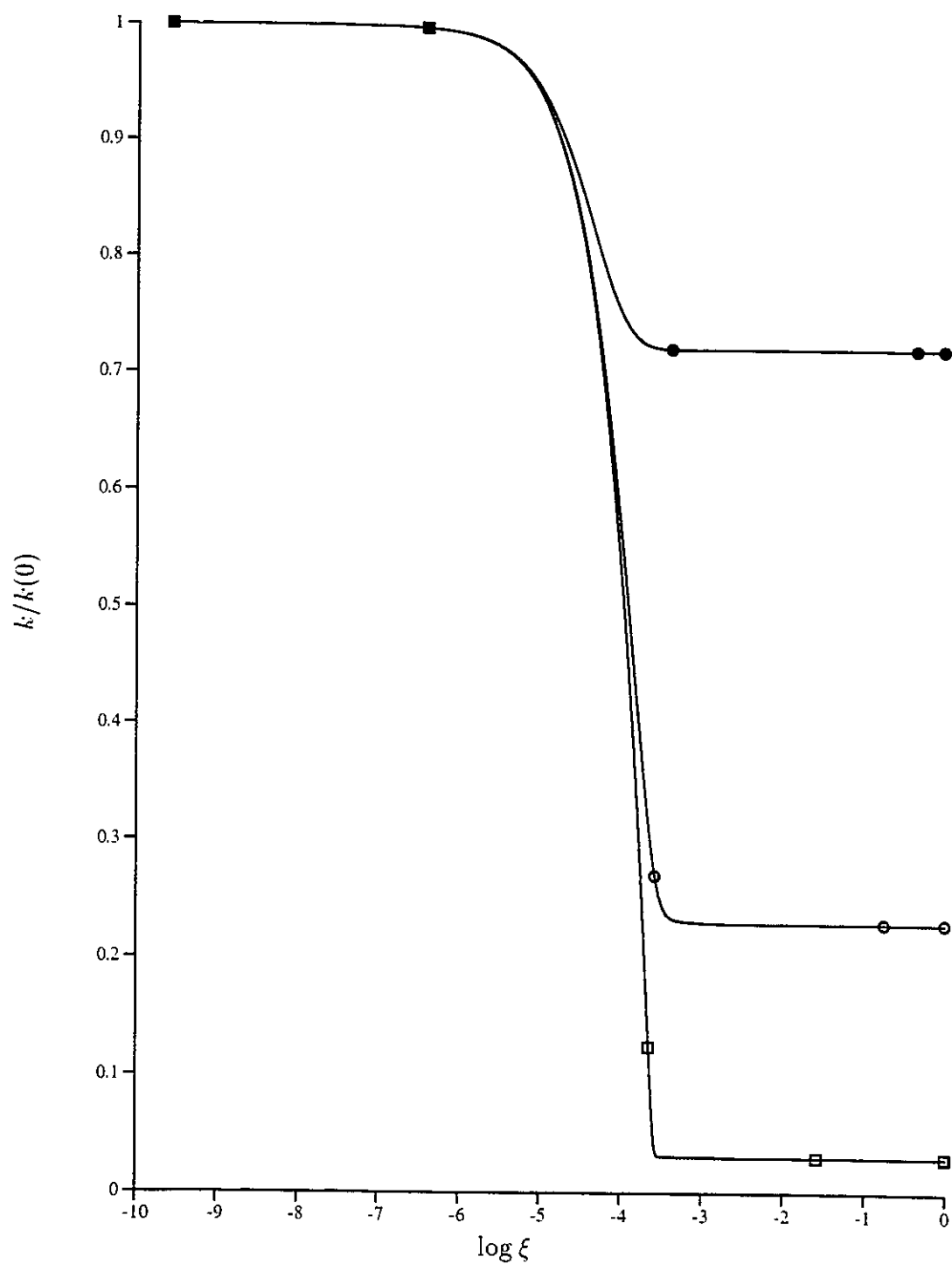


Figure B.3: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $\text{R} = \text{He}$ ,  $T = 1000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

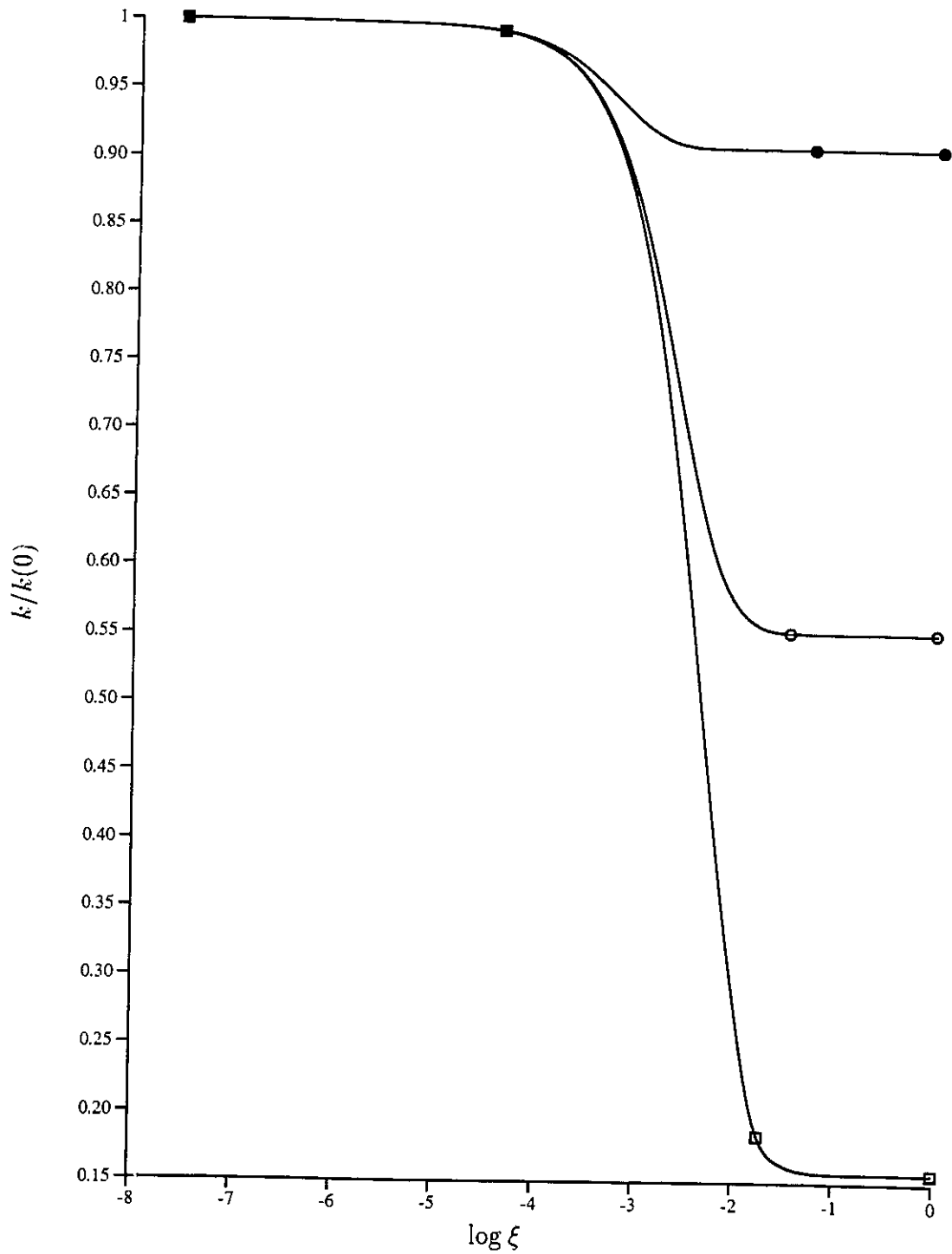


Figure B.4: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $R = \text{He}$ ,  $T = 2000 \text{ K}$  and various values of  $[\text{Er}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

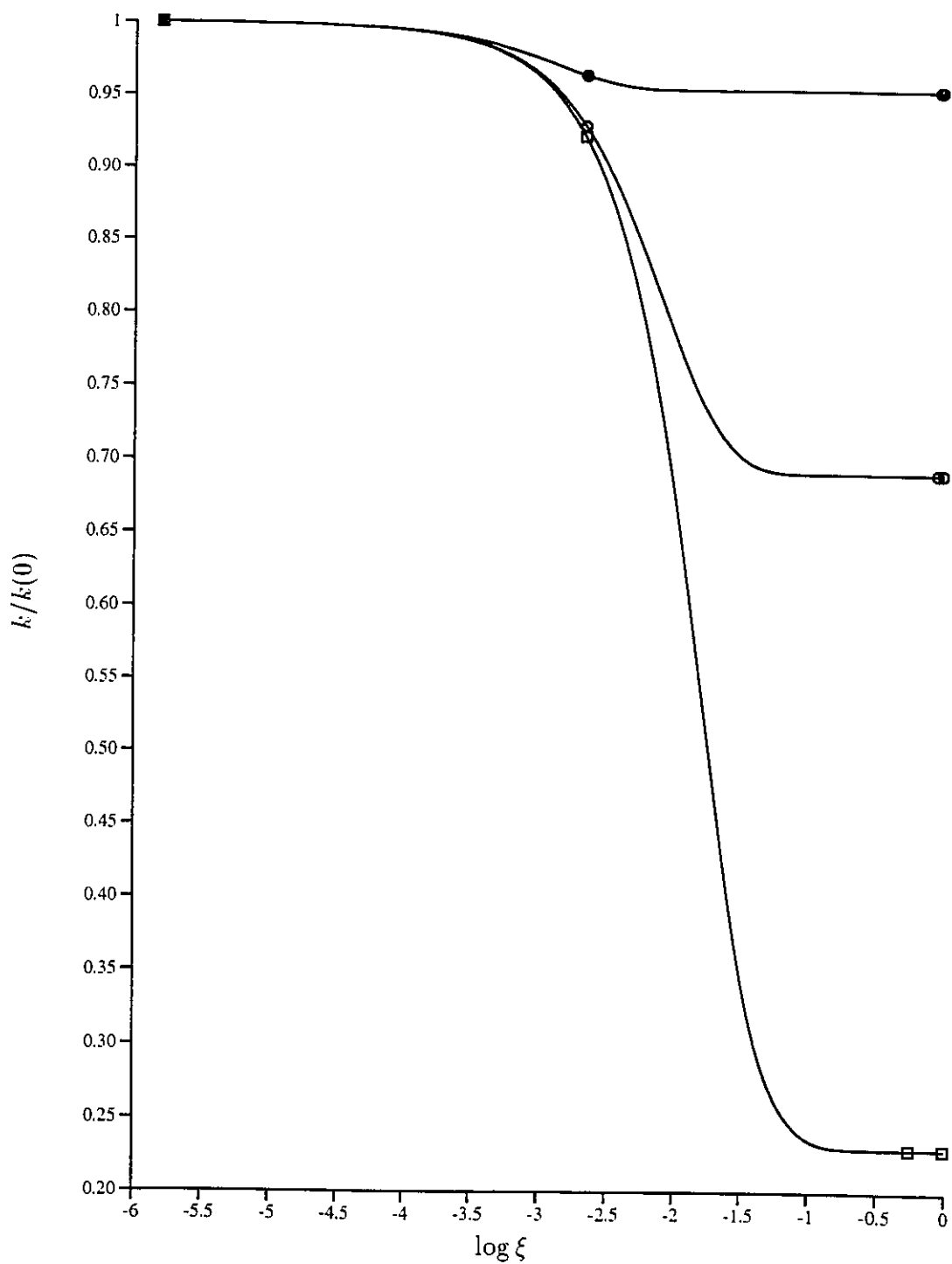


Figure B.5: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $R = \text{He}$ ,  $T = 3000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

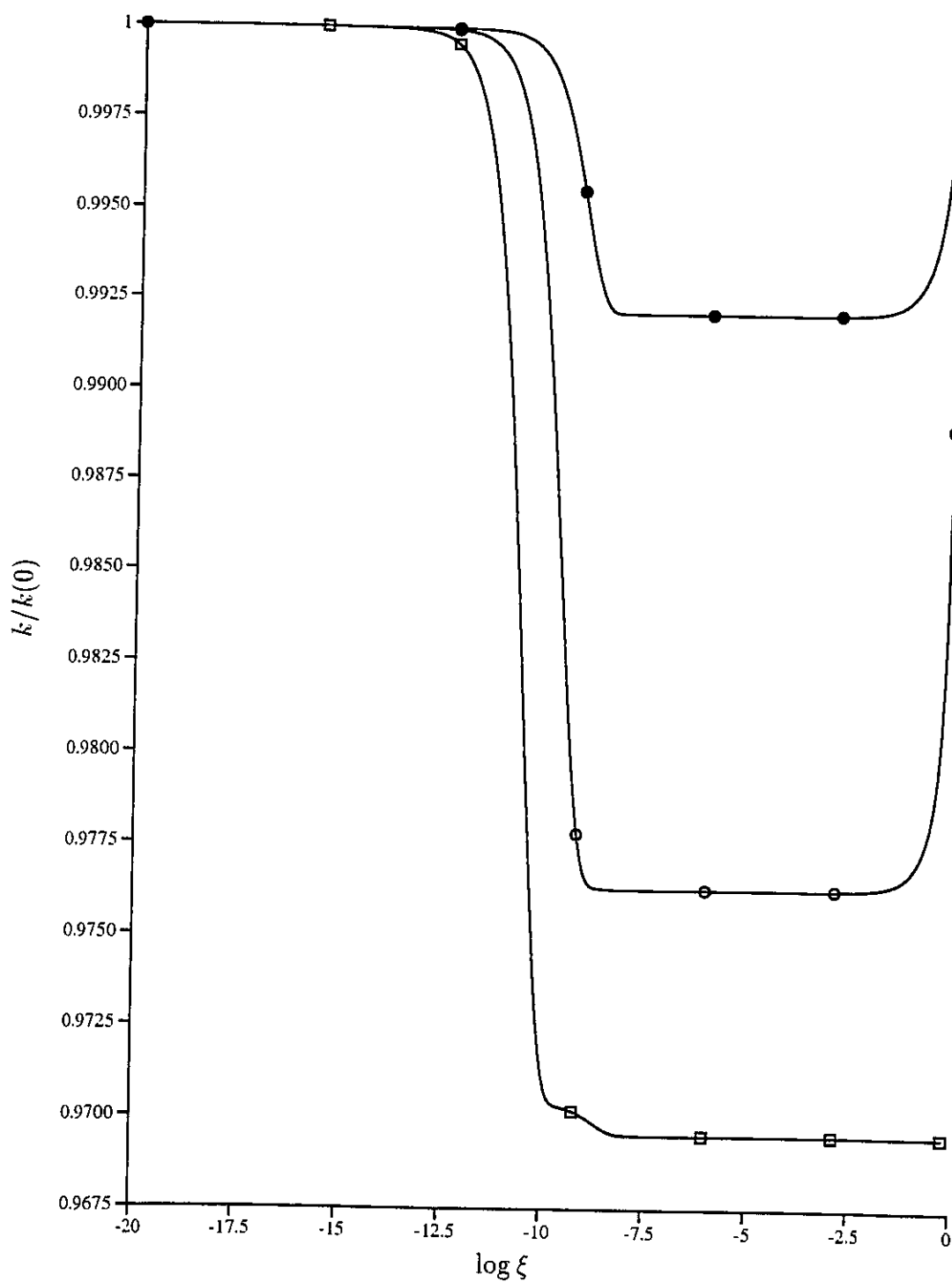


Figure B.6: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $R = \text{HCl}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

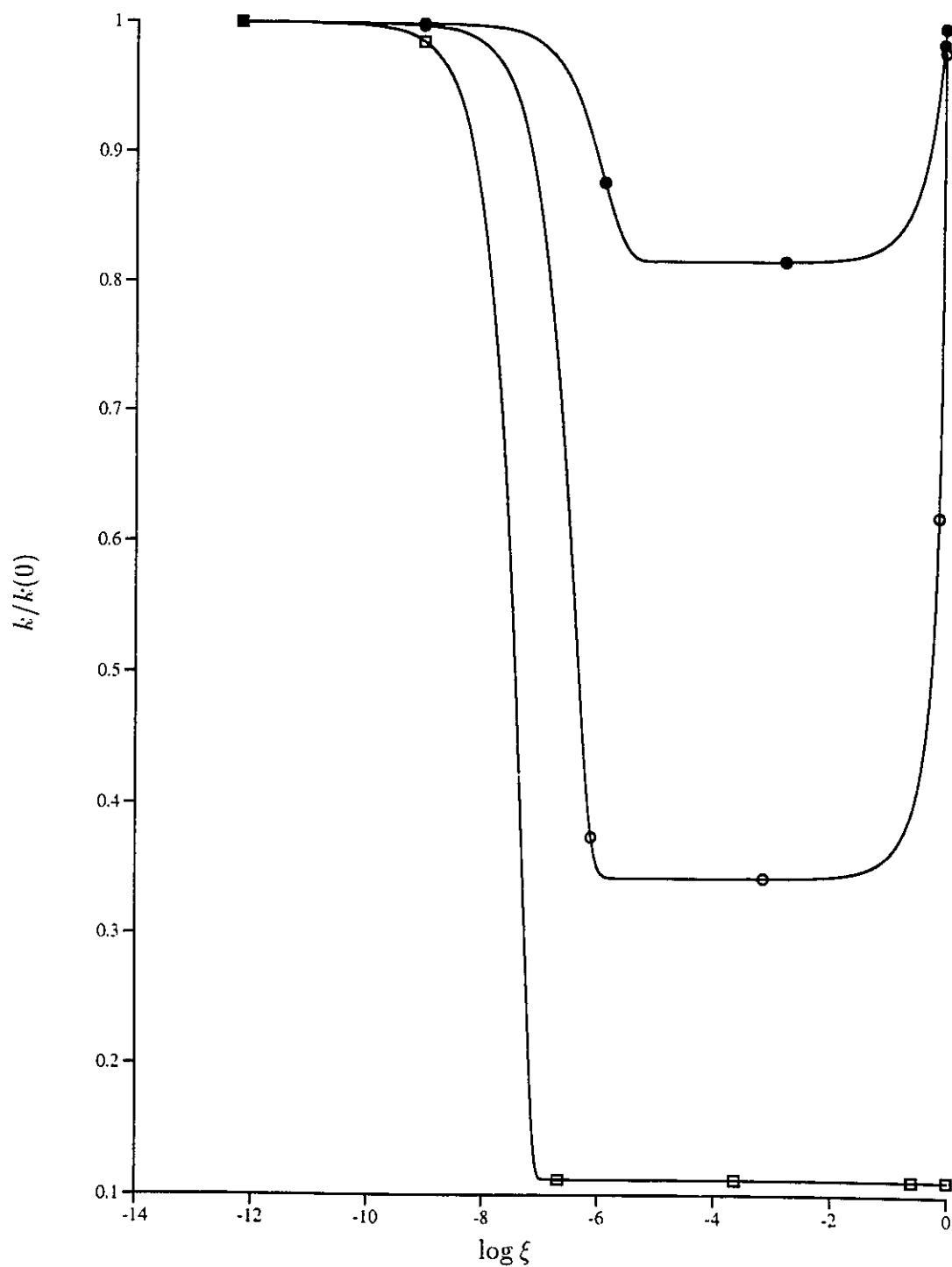


Figure B.7: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $R = \text{HCl}$ ,  $T = 500 \text{ K}$  and various values of  $[Br]/[R]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

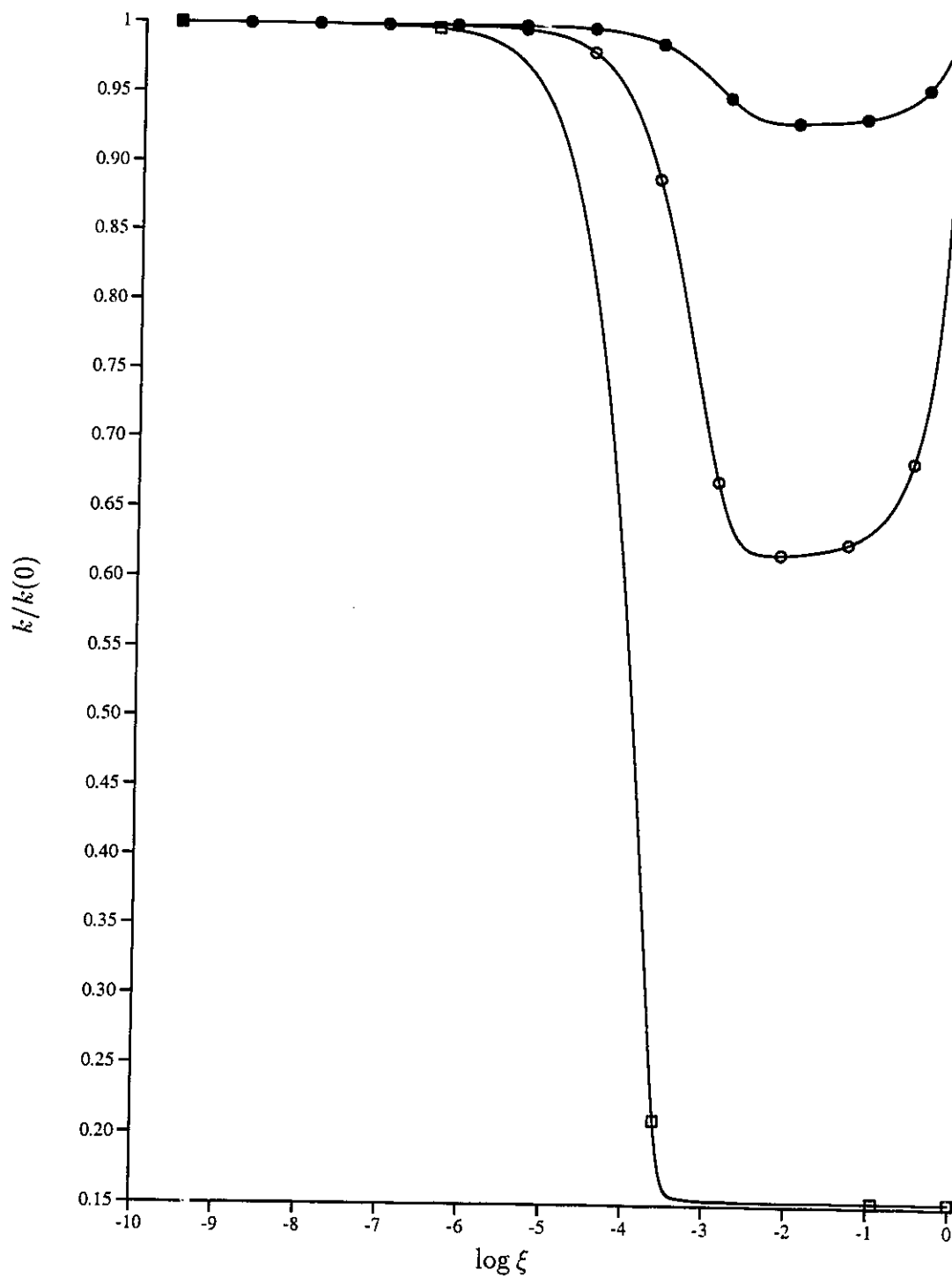


Figure B.8: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $R = \text{HCl}$ ,  $T = 1000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

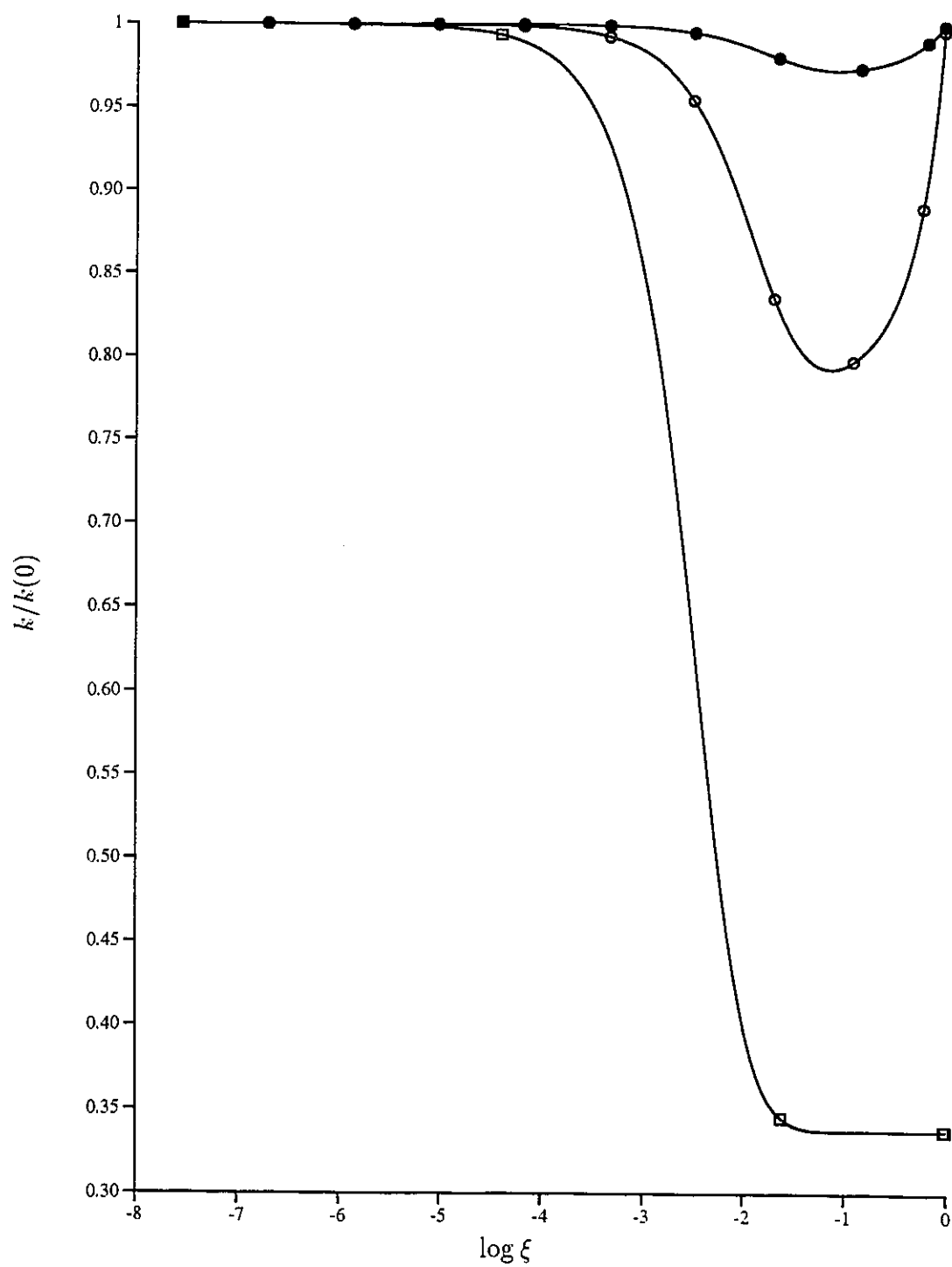


Figure B.9: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $\text{R} = \text{HCl}$ ,  $T = 2000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

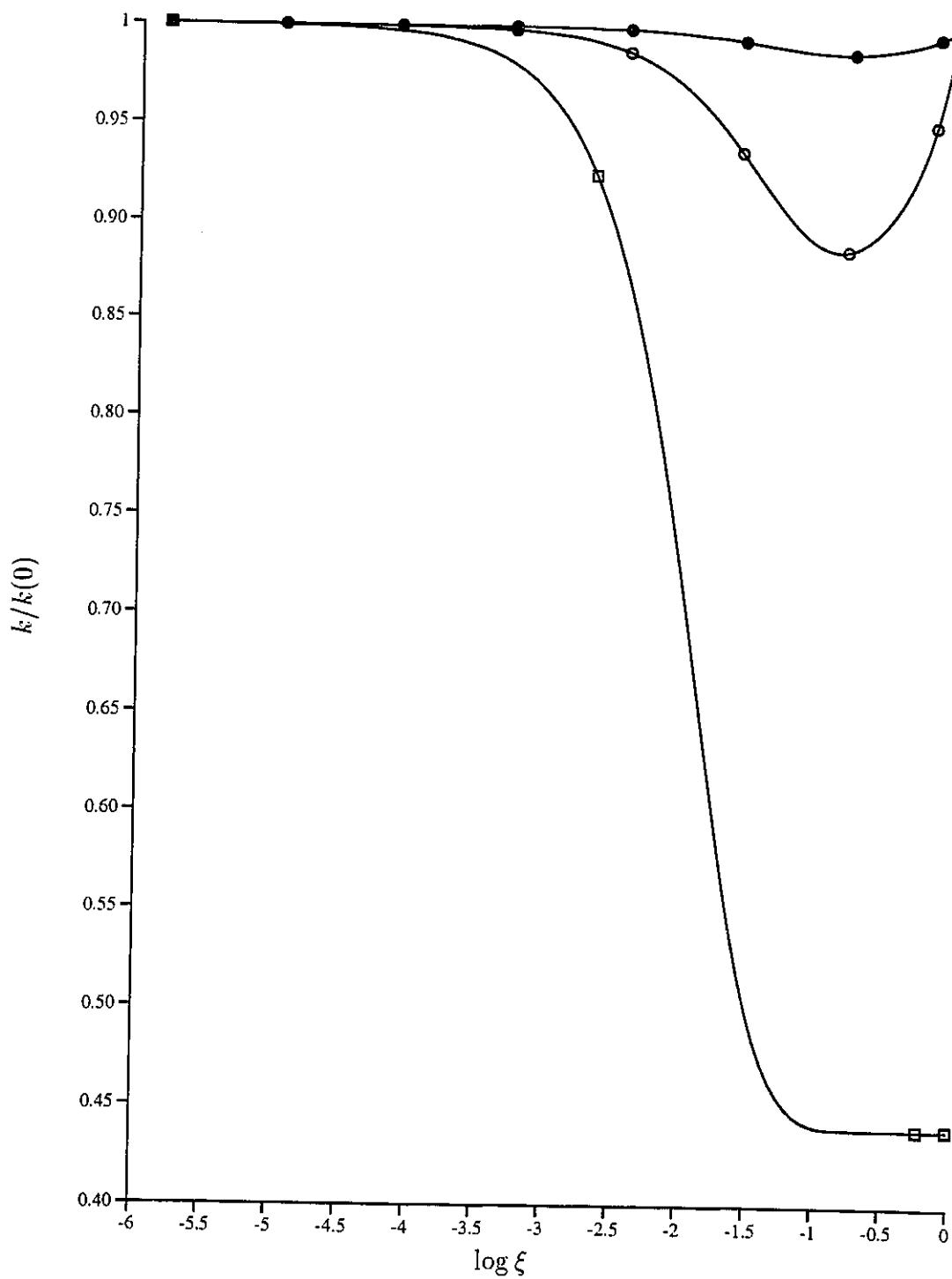


Figure B.10: The ratio  $k/k(0)$  vs.  $\log \xi$  for  $\text{R} = \text{HCl}$ ,  $T = 3000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

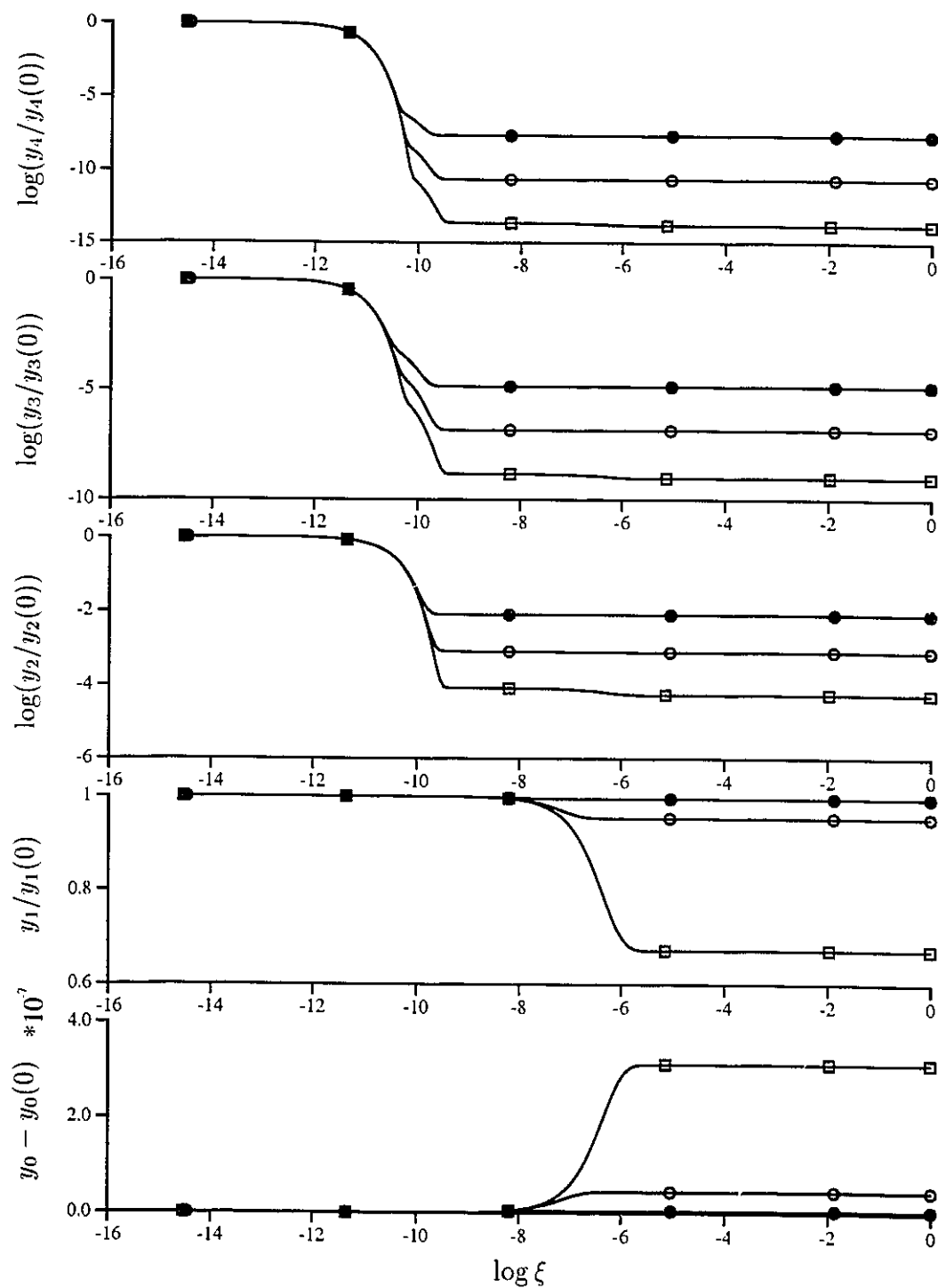


Figure B.11: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{He}$ ,  $T = 300 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

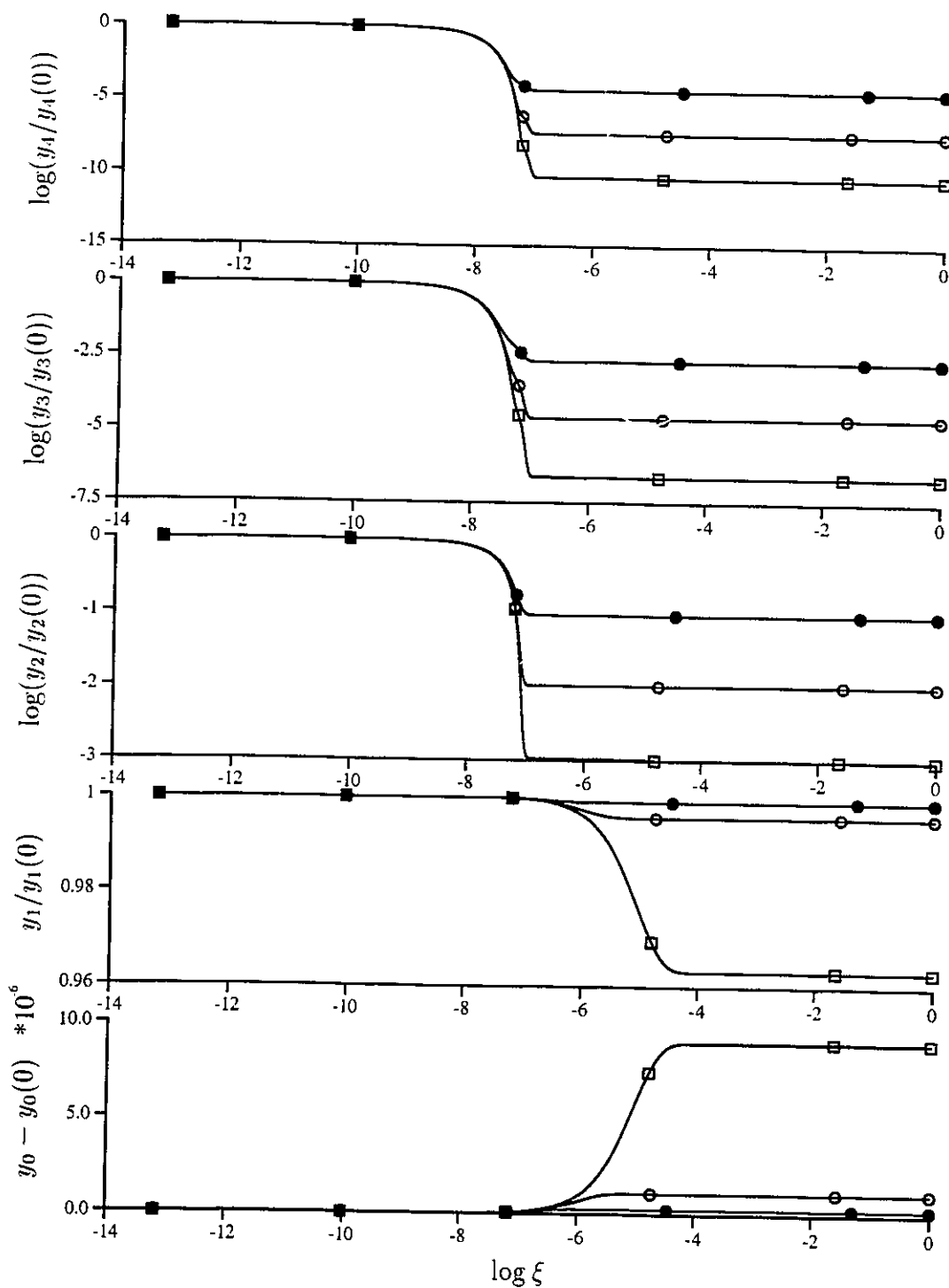


Figure B.12: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{He}$ ,  $T = 500 \text{ K}$  and various values of  $[Br]/[R]$ : (●) 0.01, (○) 0.1, (□) 1.0

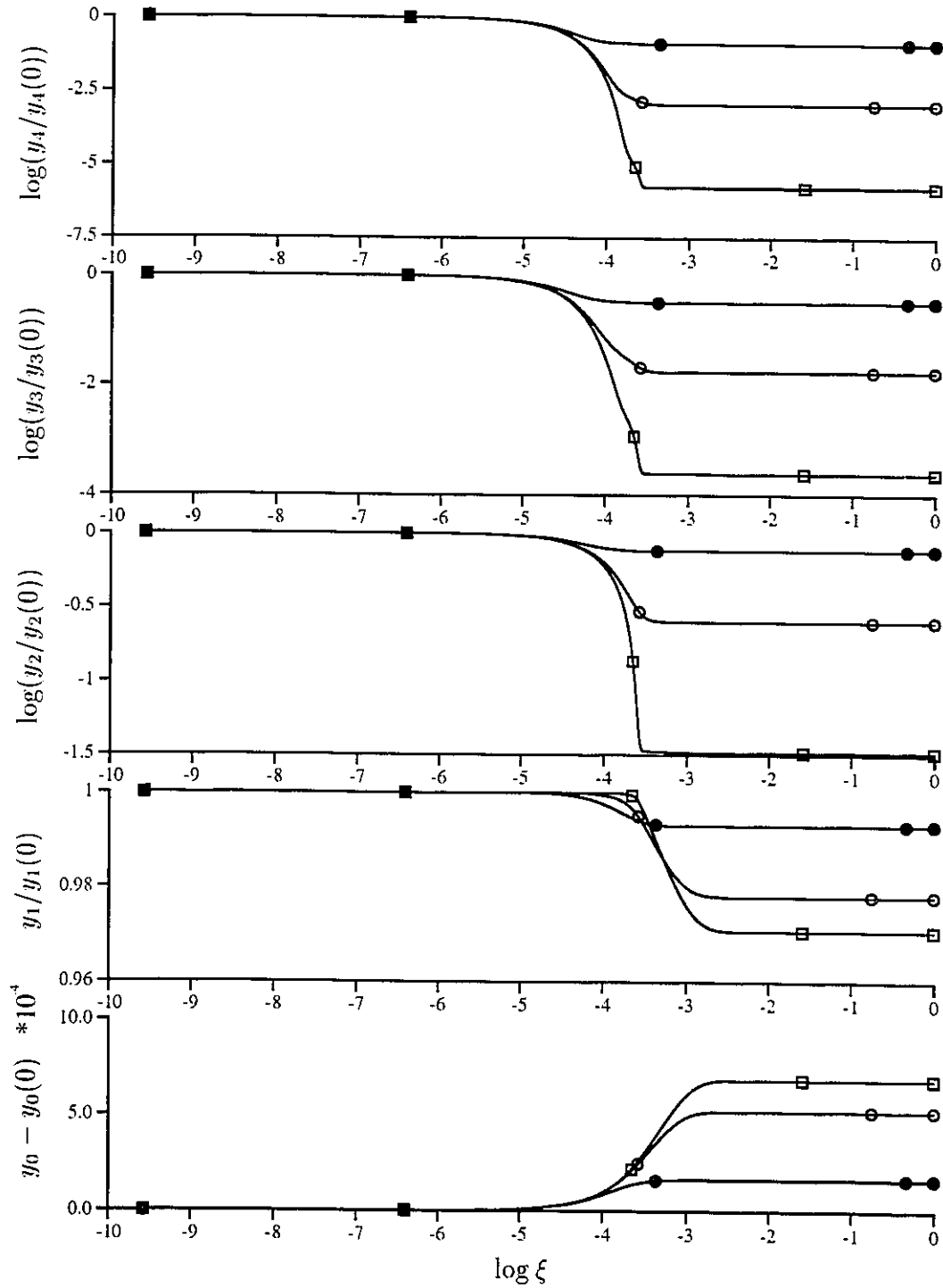


Figure B.13: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{He}$ ,  $T = 1000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : (●) 0.01, (○) 0.1, (□) 1.0

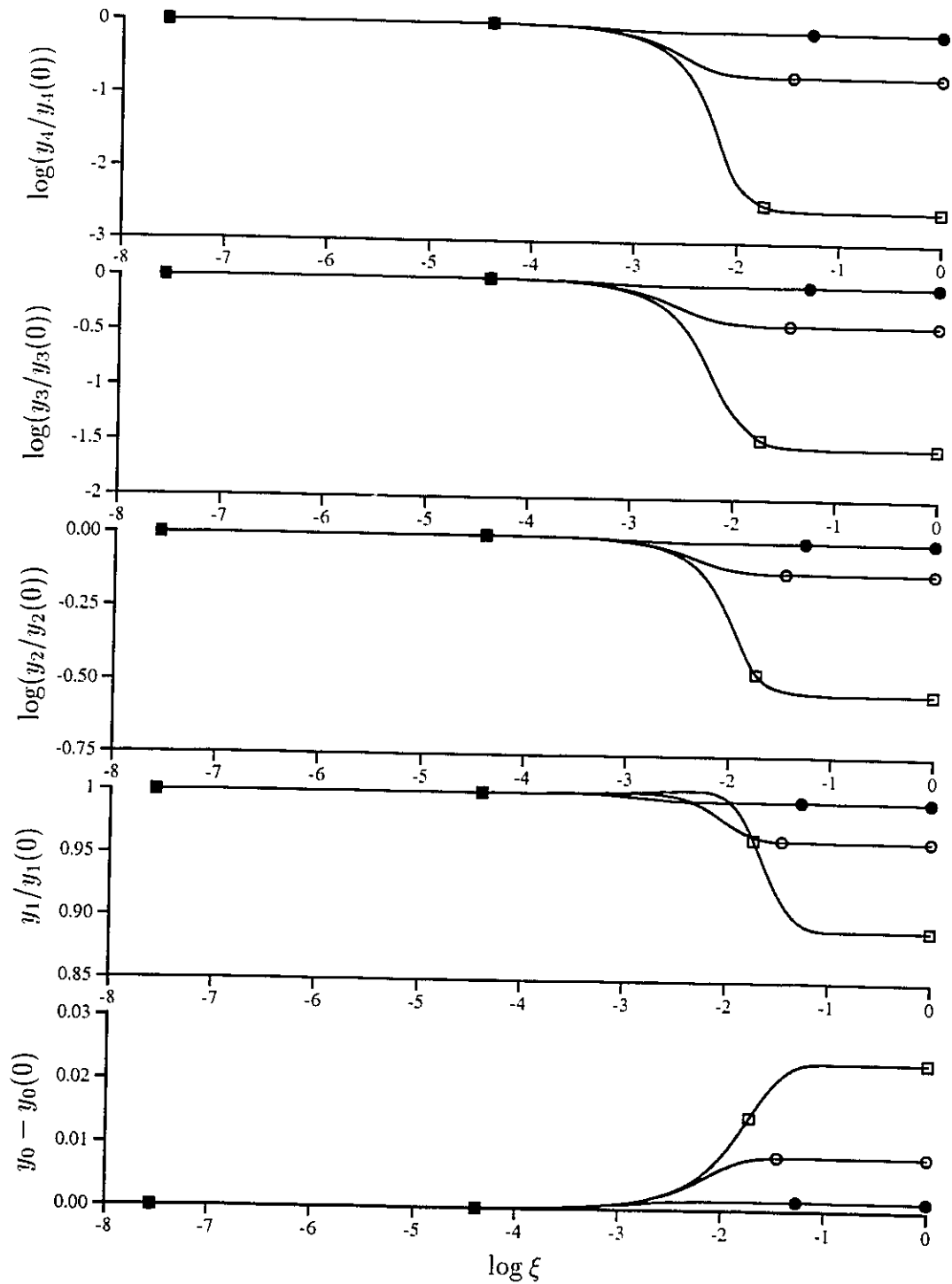


Figure B.14: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{He}$ ,  $T = 2000 \text{ K}$  and various values of  $[Br]/[R]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

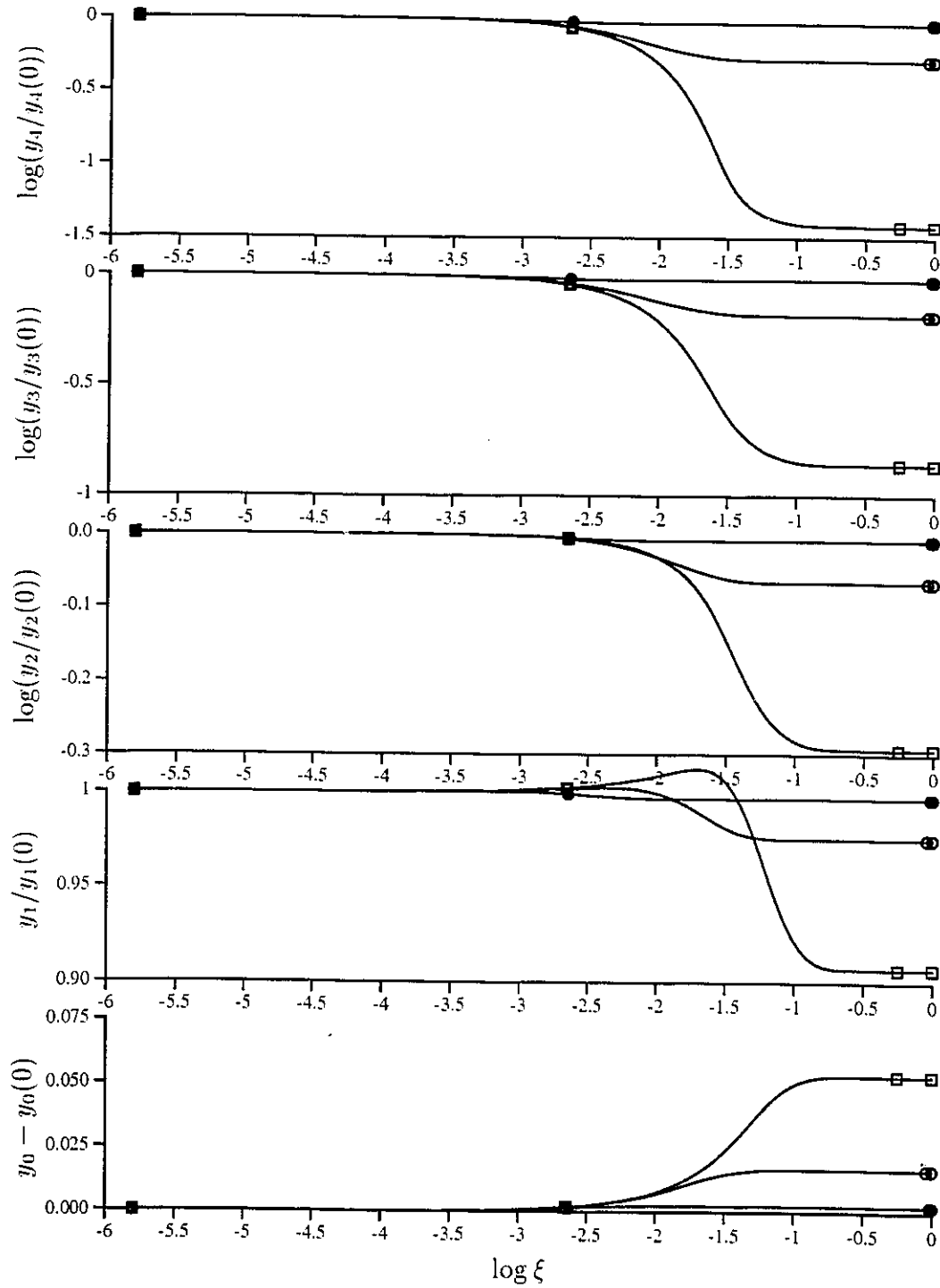


Figure B.15: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{He}$ ,  $T = 3000 \text{ K}$  and various values of  $[Br]/[R]$ : (●) 0.01, (○) 0.1, (□) 1.0

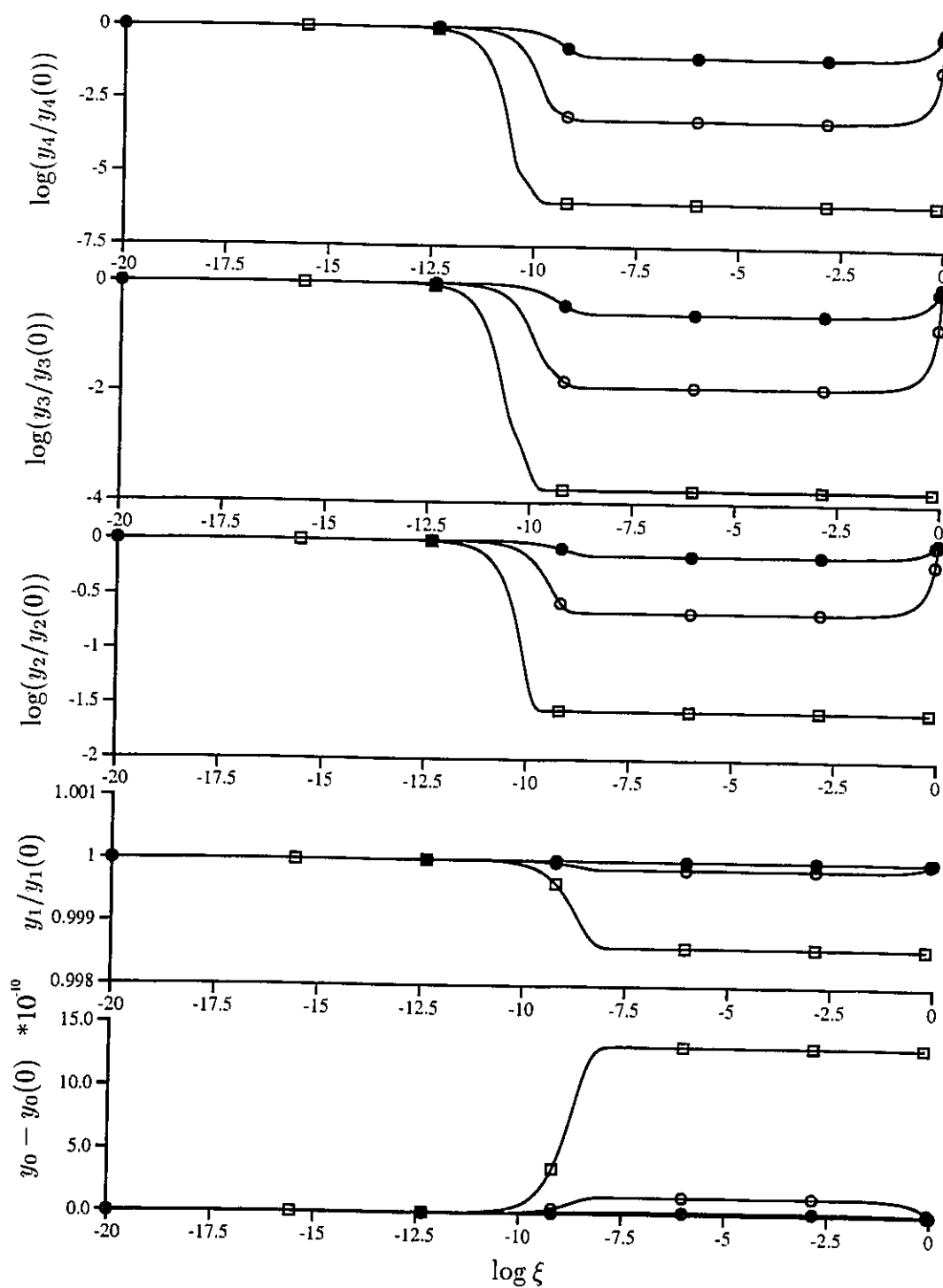


Figure B.16: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{HCl}$ ,  $T = 300 \text{ K}$  and various values of  $[Br]/[R]$ : ( $\bullet$ ) 0.01, ( $\circ$ ) 0.1, ( $\square$ ) 1.0

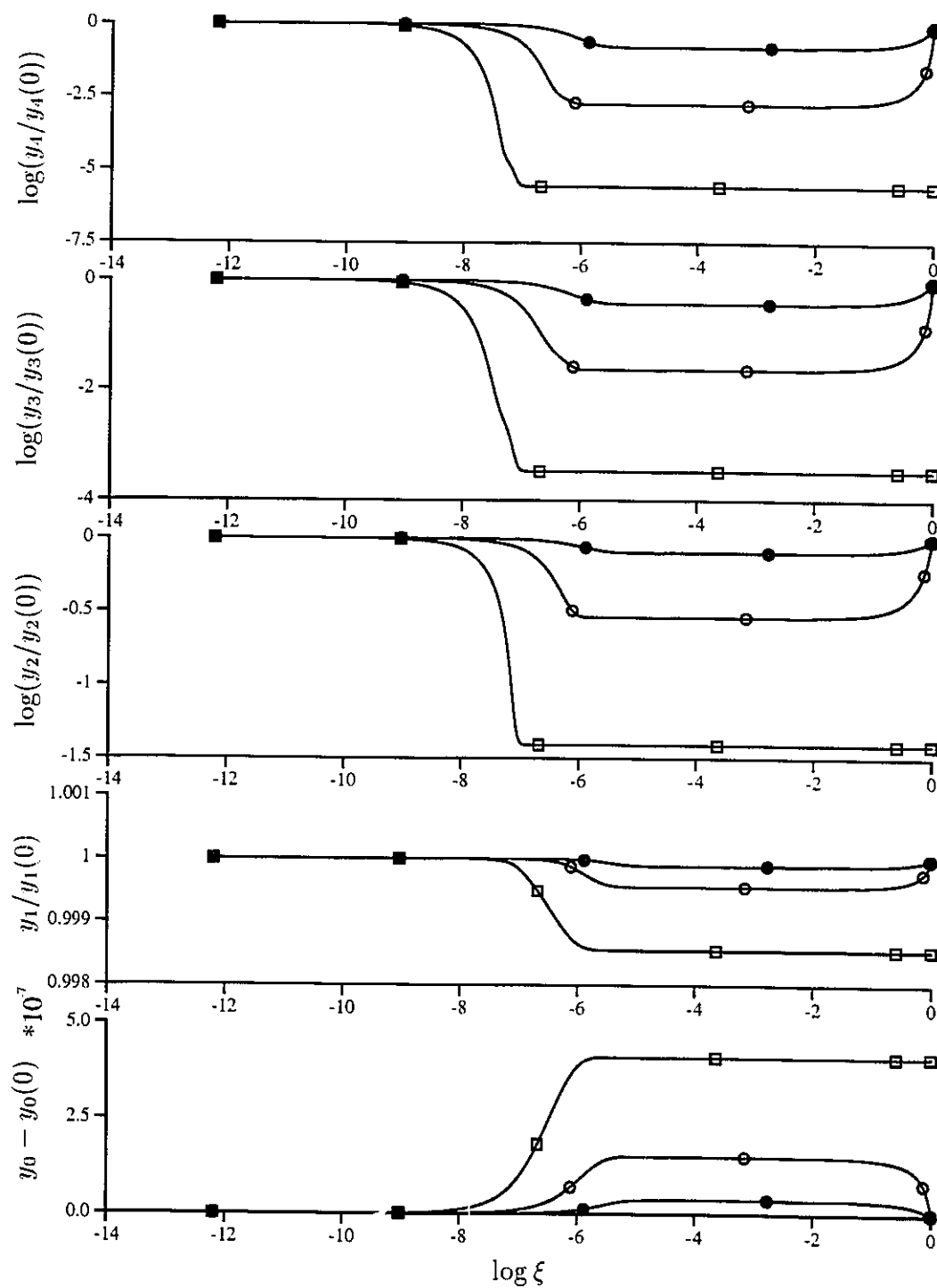


Figure B.17: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{HCl}$ ,  $T = 500 \text{ K}$  and various values of  $[Br]/[R]$ : (●) 0.01, (○) 0.1, (□) 1.0

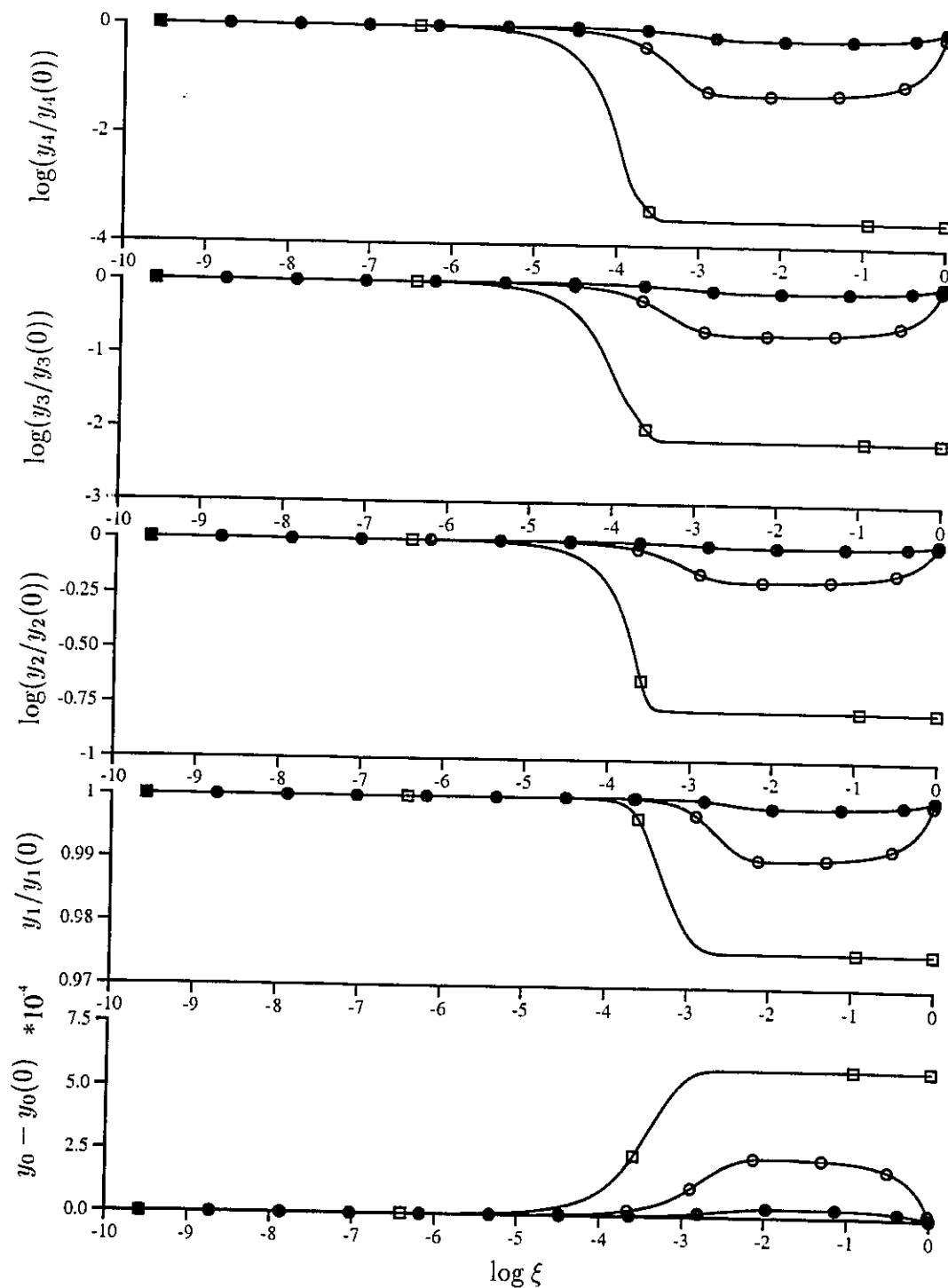


Figure B.18: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{HCl}$ ,  $T = 1000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : (●) 0.01, (○) 0.1, (□) 1.0

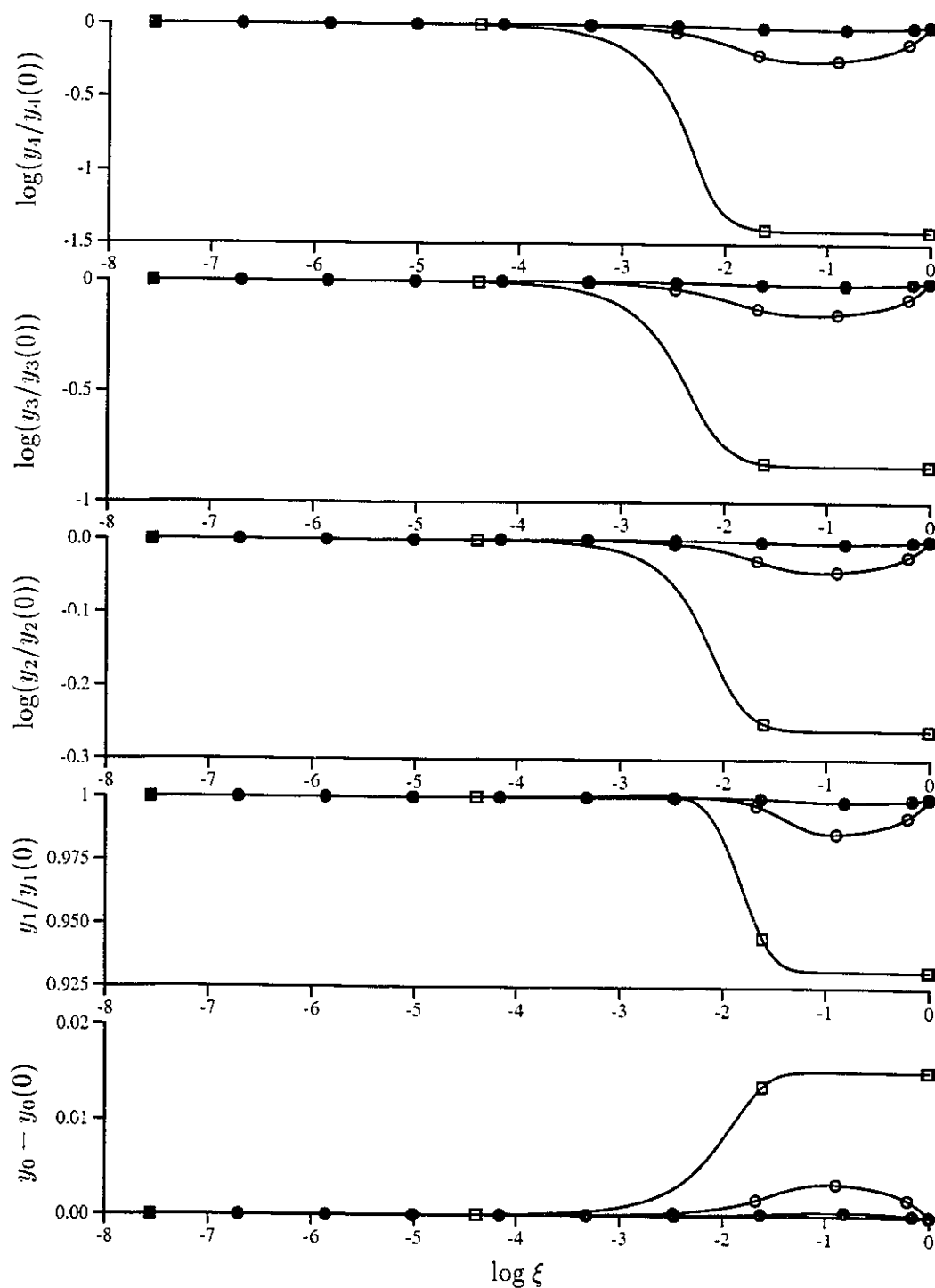


Figure B.19: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{HCl}$ ,  $T = 2000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : (●) 0.01, (○) 0.1, (□) 1.0

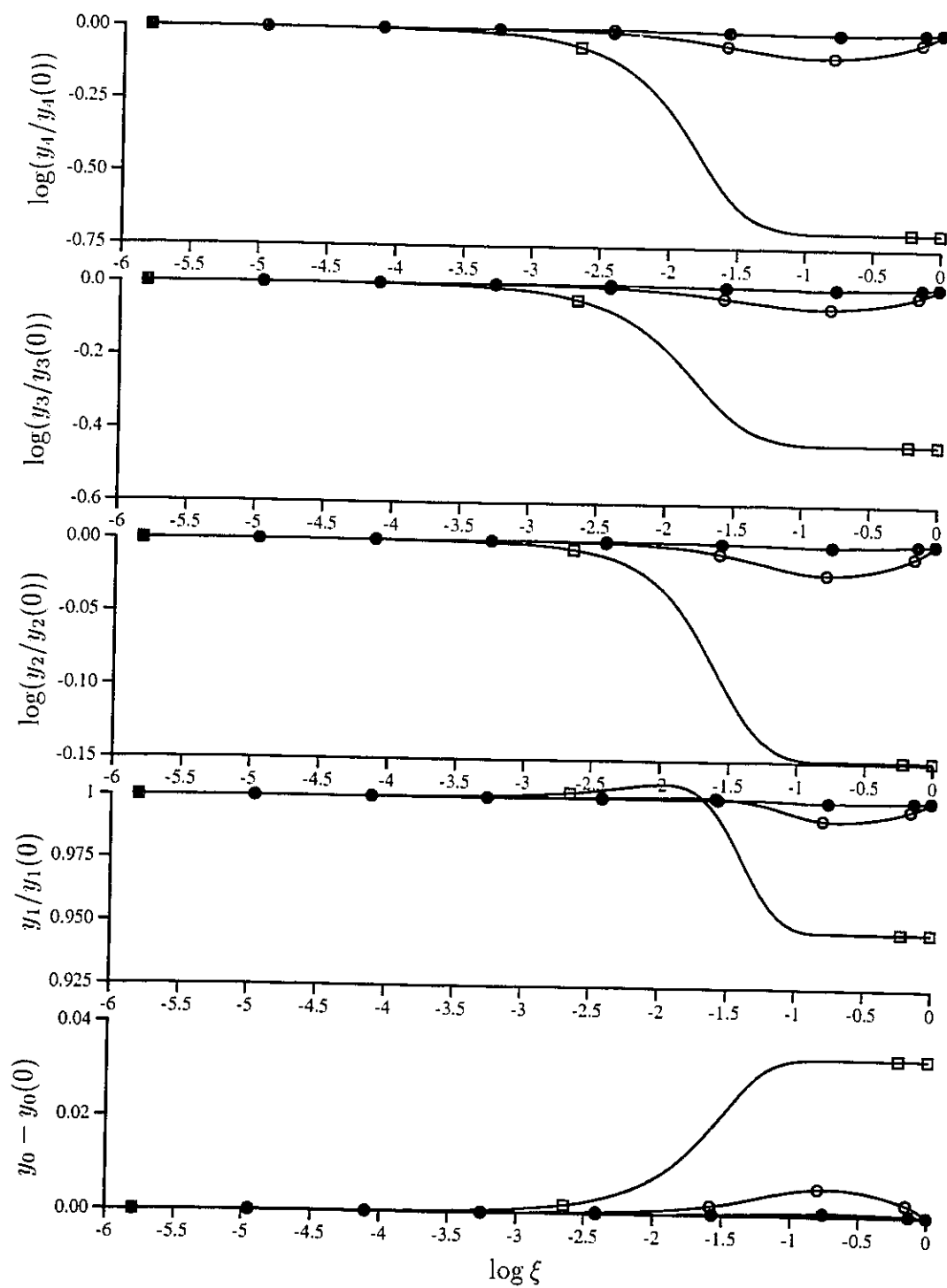


Figure B.20: Changes in  $y_v$  vs.  $\log \xi$  for  $v = 1-4$ ,  $R = \text{HCl}$ ,  $T = 3000 \text{ K}$  and various values of  $[\text{Br}]/[\text{R}]$ : (●) 0.01, (○) 0.1, (□) 1.0

## **Appendix C. Source Code**

The source code for BIMSIM is presented on the following pages. In order to conserve space and since the source is presented mainly for archival reasons a very small typeface and a double column layout is used. (A larger typeface would require single column layout and require four times as many pages.)



```

0 which exceed the collision frequency possible, according
0 to the temperature and pressure of the system.
0 Thus, the subroutine DIFFUS is called.
0 In this subroutine the collision frequency is calculated
0 and any MAX(V,J) which exceeds it is set equal to it.
0 If the user is modelling a unimolecular reaction, or an ideal
0 reaction which is not limited by the rate of diffusion, then
0 "no upper limit" in the subroutine DIFFUS should be specified.
0 IF (AMMETH.NE.3) THEN
0   CALL PIPPOPIANPOP
0   CALL DIFFUS
0   CALL SOLVER
0 ENDIF
0 IF (AMMETH.NE.1) THEN
0   CALL YINIT(10TR1)
0   CALL EIGEN(10TR1)
0 ENDIF
907 WRITE(10OUT) ' Type 1 if you want tab' read output. (default: 1)
0 READ(INP) * END=909: ANTAB
0 GO TO 906
908 IF (10OUT.EQ.ISUP) THEN
0   INP=5
0   10UR=6
0   GO TO 907
0 ELSE
0   CALL DFALTI(ANTAB)
0 ENDIF
906 WRITE(10AVE) * ANTAB
0 IF (ANTAB.EQ.1) CALL TABOUT
909 WRITE(10UT) ' Type 1 if you want output plotted (Def: 1)
0 READ(INP) * END=909: ANPLOT
0 GO TO 910
909 IF (10OUT.EQ.ISUP) THEN
0   INP=5
0   10UR=6
0   GO TO 908
0 ELSE
0   CALL DFALTI(ANPLOT)
0 ENDIF
910 WRITE(10AVE) * ANPLOT
0 IF (ANPLOT.EQ.1) CALL PLTOUT
0 STOP 1006
0 END

```

```

SUBROUTINE INIT
C This subprogram initializes the physical constants used.
C The constants are given for SI units.
C (Specify more significant figures for these constants)
C No document default changeability.
IMPLICIT DOUBLE PRECISION (A-H, Y-N, O-W-Z)
IMPLICIT INTEGER (A-I, J-L, M-V)
IMPLICIT CHARACTER*(*)
REAL*8 APAD, AMASS
DOUBLE PRECISION AVOGAD
CHARACTER*36 ISPEC(10)
COMMON/SPECIES/ISPEC(10)
COMMON/PHYSCT/PLANCK, FEOLTT, RGAS, SCKEPI, CLIGHT, PI, ISEVER
COMMON/IO/INP, 10OUT, ISAVE, ISUP, IDEBUG, INEAT, INCV, UUNIT, VUNIT(3)
COMMON/COUNT/MAXV, MAXU(10), MAXUN, INPTD, ICOLL
COMMON/POP/IN(10), 10, 10, 10, 10, 10, 10, 10, 10, 10, 10
COMMON/ELEV/ELEV(10), 0, 10, 10, 10, 10, 10, 10, 10, 10, 10
COMMON/MASS/AMASS, CMASS, AMASS, RECMAS, BRAD, CRAD, APAD, RE
PARAMETER (AVOGAD=6.02205E23)
EQUIVALENCE (ISPEC(1), ISPEC(1,1))
C The long strings which are specific for the various kinds of
C collisions are written into a *36 character string which is
C then effectively broken into 6 * 6 character strings by means
C of equivalencing with an array ISPEC of dimension 6.
C The string-length is chosen because CHARACTER*6 is conventionally
C used in other subroutines.
EFFECT is called because underflow errors may occur in the program.
ERRSET is an IBM system program. Other adaptations of this
program to different systems will require a call to a different
underflow-adjustment system routine. If no such subroutine is
available, the user may have to modify the program.
CALL EFFECT(1008, 0, -1, 1, 1)
MAXUN = 11
INP = 5
IDEBUG = 1
C ...debug toggled on for the time being
10OUT = 6
C 10OUT and IDEBUG must be set for $PRSUB to work
$SUB=$PRSUB( INIT )
C
C If this program had been run once, the answers will be saved
C on unit ISAVE so that a restart option is available.
C To avoid needless prompting on the terminal, the
C output will be suppressed by writing it to unit ISUP.
ISUP=99
ISAVE=99
IFLAG=0
WRITE(10OUT) * Defaults are explicitly noted or are
* indicated by a * on the forms.
WRITE(10OUT) * Defaults may be obtained by entering only a
* carriage return after the prompt.
WRITE(10OUT) *
904 WRITE(10OUT) * Is this a restart? (*Yes = 1, No = 0)
0 READ(INP) * END=911: ANRES
0 GO TO 911
911 CALL DFALTI(ANRES)
912 IF (ANRES.LT.0) OR (ANRES.GT.1) THEN
0 WRITE(10OUT) * Error in restart specification. (ANRES
0 * Choose 0 or 1)
0 IFLAG=IFLAG+1
0 IF (IFLAG.EQ.1) GO TO 904
0 CALL INTER(ANRES, ISUP)
0 ENDIF
0 IF (ANRES.EQ.1) THEN
0   INP=5
C ...to avoid colliding with filesets for v-levels
10OUT=ISUP
0 ENDIF
PLANCK=6.626E-34
FEOLTT=1.38065E-23
C k obtained from Rev. Mod. Phys. vol. 27, p. 363 (1955).
RGAS=8.31457 J/MOL-K
CLIGHT=3.0E+8
ISEVER=10000000
SCKEPI=1.050
PI=3.141592653589793
PMASS=1.67E-27
CMASS=0.0001
AMASS=6.000
RECMAS=6.000
BRAD=0.000
CRAD=0.000
APAD=0.000
RE=0.000
C
C The maximum number of output units is set by MAXUN. For
C v-levels ranging from 0 to 10, this means 11 output files,
C each of which will contain population information for
C appropriate J sub-levels. The rate-of-population-change
C information is contained on 11 more output files.
C 2 additional files will monitor total population
C and total rate-of-population-change.

```

	unit	VUNIT index
NO data for v=0...10	units 10...20	1-10
NO data summed (NLT)	21	11
NO data for v=0...10	units 40...50	10-10
NO data	unit 51	11

```

MAXV=0
DO 10 V=0, 10
  MAXU(V)=1
  INPT(V)=10+V
  UUNIT(V)=10+40+V
10 CONTINUE
C The matrix ELEV must have one extra level, so that the
C energy difference can be calculated in EDEL for a given level.
DO 10 V = 0, 10
  DO 10 J = 0, 20
    ELEV(V, J)=0.0
10 CONTINUE
C NV initialized value unimportant, normalized later.

```

# APPENDIX C. SOURCE CODE

```

NOV 01 * 1.0
NOTES ON DEFAULTS:
The default value is either explicitly denoted as such
or it is indicated with a * in front of the menu number.
If the input is from the terminal, a carriage return will
signify that the default value is desired.
If the input is from a file, then reset the input unit
to that of the terminal, so that the user can pick up
from where s/he left off...
The following default subroutines are called:
DFALTI - for double precision numbers
DFALTI - for integer numbers
DFALTI - for character input.
If the user wishes to change the default value, only the
prompting default reminder, and the second argument in the
DFALTI call must be changed.
900 WRITE(IOUT,*) Type a 1 if prompting for units is
* required. Any other number--no prompting. *Default: 0
* (The routine for prompting for units is NOVICE.)
READ(INP,*,END=901) INOV
GO TO 902
901 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 900
ELSE
  CALL DFALTI(INOV 0)
ENDIF
902 WRITE(ISAVE,*) INOV
JCOLL(1)=0
JCOLL(2)=0
JCOLL(3)=0
SSPGEN(1): For collisions between A and BC
SSPGEN(2): For collisions between BC and EC
SSPGEN(3): For collisions between M and BC
IF(INOV.NE.1) RETURN
WRITE(IOUT,*) This program accepts many kinds of units.
WRITE(IOUT,*) The user must know which two-letter codes refer
WRITE(IOUT,*) to which units. e.g. EV means electron volts.
WRITE(IOUT,*) The letter code for inverse units is identical to
WRITE(IOUT,*) that of the non-inverted units. The user must
WRITE(IOUT,*) input values and units with the units enclosed
1 in single quotes.
2 and separated by commas. e.g., "4.0 cm/s".
WRITE(IOUT,*)
RETURN
END

```

```

SUBROUTINE RANRYS
IMPLICIT CHARACTER*6 (Z)
IMPLICIT DOUBLE PRECISION (A-H,F,N,O,W,D)
IMPLICIT INTEGER (I-S,L-M,V)
CHARACTER*3 UNITS,UNIT
COMMON/IO/INP,IOUT,ISAVE,ISUP,ITERIG,INDAT,INOV,NOV,NOVICE,NO2,
COMMON/MASS/EMASS,CMASS,AMASS,REMGAS,RRAD,CRAD,ARAD,PR
DIMENSION XMASS(106), XRAD(106), ZMASS(1), ZRAD(1), ZD(1)
EQUIVALENCE (ZMASS(1),EMASS), (ZRAD(1),RRAD)
EQUIVALENCE (ZD(1),ZD), (ZD(2),ZD), (ZD(3),ZD)
The following is an admittedly schizophrenic combination
of the weights of the most prevalent isotopes (from
W.H. Flyvare 1978) and the averaged atomic weight
(from table from Chemical Services, Box 45124
Los Angeles, Ca. 90045). The s- and p-shell atoms
have the "most prevalent isotope" weights; the d- and
f-shell species are the averaged values.
Values are in g/mole.
DATA XMASS(1),L1,501/1.00783,4.00260,7.01601,9.01219,11.00931
1 12.00000,14.00307,15.99491,16.99940,19.99244
2 20.99744,22.98977,23.98504,26.98153,27.97683,30.97376
3 31.97207,34.96885,35.96338,38.96371,39.96249
4 44.9559,47.90,50.9415,51.996,54.9380
5 55.847,58.9332,58.70,63.546,65.38
6 68.9327,73.9212,74.9216,76.9163,78.9183
7 84.9115,84.9117,87.9056,88.9059,91.224
8 92.9064,95.94,97.0,101.07
9 106.4,107.868,112.41,114.9039,117.9018 /
DATA XMASS(L),L=51,106) /
1 120.9038,120.9062,126.9044,131.9043,132.9054,
2 137.9055,138.9055,140.12,140.9077,144.24
3 145.0,150.4,151.96,157.25,158.9254
4 163.50,164.9364,167.26,168.9342,173.04
5 174.967,178.49,180.9479,183.85,186.207
6 199.2,199.22,199.09,196.9665,200.59
7 204.9745,207.9766,208.9804,206.0,220.0
8 223.0,223.0,226.0254,227.0278,232.0381
9 231.0369,238.029,237.0462,244.0,244.0
1 247.0,247.0,251.0,254.0,257.0
2 258.0,259.0,260.0,261.0,262.0
3 263.0 /
C Values of atomic radii obtained from Chemical Principles
C by Masterton and Slowinski (W.B.Saunders 1969).
C Interpolated values for Fe, Mn, Sm, At, Fr and Ra.
C Values extend only up to U (element 92).
C Values are in angstrom.
DATA XRAD(L),L=1,92) /
1 0.77,0.70,0.66,0.64,1.12
2 1.66,1.60,1.43,1.17,1.10
3 1.06,0.99,1.54,1.31,1.27
4 1.60,1.46,1.31,1.35,1.29
5 1.26,1.25,1.34,1.38,1.33
6 1.22,1.22,1.21,1.17,1.14
7 1.69,1.44,1.15,1.60,1.57
8 1.43,1.36,1.34,1.33,1.34
9 1.38,1.44,1.42,1.62,1.40 /
DATA XRAD(L),L=93,106) /
1 1.41,1.37,1.33,1.50,1.62
2 1.27,1.27,1.22,1.21,1.20
3 1.0,1.0,1.04,1.19,1.27
4 1.77,1.78,1.75,1.74,1.73
5 1.74,1.57,1.43,1.37,1.37
6 1.34,1.35,1.36,1.44,1.37
7 1.71,1.75,1.46,1.63,1.6
8 2.1,2.1,2.0,2.0,1.8
9 1.8,1.4 /
C Values of atomic radii for atomic number > 92
C unavailable to us at the present time.
DATA XRAD(L),L=93,106) /
1 3.,3.,3.,3.,3.
2 3.,3.,3.,3.,3.
3 3.,3.,3.,3.,3. /
ISUP=DFALTI(IOUT,*)
900 WRITE(IOUT,*) This program simulates the reaction system
WRITE(IOUT,*) BC + A -> Products
WRITE(IOUT,*) Input the atomic numbers for B, C and A in that
4 order.
WRITE(IOUT,*) *Default: 1,17,35 (HCl-Br)
WRITE(IOUT,*) *Type 0,0 if you wish to use the built-in
1 periodic table.
READ(INP,*,END=901) ICB, ICC, ICA
901 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 900
ELSE
  CALL DFALTI(ICB,1)
  CALL DFALTI(ICC,17)
  CALL DFALTI(ICA,35)
ENDIF
902 WRITE(ISAVE,*) ICB, ICC, ICA
C Check to see whether the atomic numbers correspond to
C to a range available to known elements.
IF(ICB.LT.1 .OR. ICB.GT.106 .OR.
  ICC.LT.1 .OR. ICC.GT.106 .OR.
  ICA.LT.1 .OR. ICA.GT.106) THEN
  WRITE(IOUT,*) Atomic numbers range from 1 to 106.
  WRITE(IOUT,*) Choose from the following chart according
  4 to atomic number.
  CALL CHART(ICB, ICC, ICA)
ENDIF
WRITE(IOUT,*) These are the masses and covalent radii
WRITE(IOUT,*) corresponding to the atomic numbers chosen:
WRITE(IOUT,*) Atomic Number Mass Radius
WRITE(IOUT,*) (g/mole) (angst.)
WRITE(IOUT,*) ICB, XMASS(ICB), XRAD(ICB)
WRITE(IOUT,*) ICC, XMASS(ICC), XRAD(ICC)
WRITE(IOUT,*) ICA, XMASS(ICA), XRAD(ICA)
IFLAG=0
904 WRITE(IOUT,*) Do you wish to accept these values?

```

```

A      (*Yes = 1, No = 0)
READ(INP,*,END=905) IANACC
GO TO 906
905 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 904
ELSE
  CALL DPALTC(IANACC,1)
ENDIF
906 WRITE(ISAIVE,*) IANAC
  IF(IANACC.EQ.0) THEN
C prompt for new values, a mass/radius pair at a time
  DO 100 I=1,2
907   WRITE(IOUT,*) 'What is the mass and radius of element no.
     1',I,' including units?'
     WRITE(IOUT,*) '<Default: AMASS=12.011 GM'
     A      XRAD(I)=1.0
     IF(I.EQ.1.AND.INOV.EQ.1) THEN
C      only call novice once in the loop (novskill otherwise)
       CALL NOVICE(40)
       CALL NOVICE(30)
     ENDIF
     READ(INP,*,END=908) XMASS(I),UNITI,ERAD(I),UNITI
908   IF(IOUT.EQ.ISUP) THEN
     INP=5
     IOUT=6
     GO TO 907
   ELSE
     CALL DPALTC(XMASS(I),XMASS(I),1)
     CALL DPALTC(UNITI,GM)
     CALL DPALTC(ERAD(I),XRAD(I),1)
     CALL DPALTC(UNITI,AG)
   ENDIF
909   WRITE(ISAIVE,*) XMASS(I),UNITI,ERAD(I),UNITI
100  CONTINUE
  ELSEIF(IANACC.EQ.1) THEN
    XMASS=XMASS(I2B)
    XMASS=XMASS(I2C)
    AMASS=XMASS(I2A)
    BRAD=XRAD(I2B)
    CRAD=XRAD(I2C)
    ARAD=XRAD(I2A)
    UNITI=GM
    UNITS=AG
  ELSE
    WRITE(IOUT,*) Error in script specification IANACC,
      Choice 1 or 1
  A      IFLAG=IFLAG+1
      IF(IFLAG.LE.1) GO TO 904
      CALL XFORM(IANACC,$00B)
    ENDIF
    DO 200 I=1,3
      CALL SIU(40,XMASS(I),UNITI)
      CALL SIU(50,ERAD(I),UNITI)
200  CONTINUE
      REMASS=XMASS(XMASS)/XMASS - XMASS
      RETURN
    END

SUBROUTINE ASSUME
IMPLICIT CHARACTER*6 (A)
COMMON/IO/INP,IOUT,ISAIVE,ISUP,IDEBUG,INCAT,INOV,VMONLY,VMUNIT(0:13)
SUB=SPRUB('ASSUME')
WRITE(IOUT,*) This program is based on a number of assumptions
WRITE(IOUT,*) namely, v changes by +1, 0 or -1 and
WRITE(IOUT,*) c changes by +2, 0 or -1. No reverse reaction is
WRITE(IOUT,*) assumed to occur. The temperature is constant
WRITE(IOUT,*) during the course of the reaction. The microscopic
WRITE(IOUT,*) rate constant is assumed to be independent of the
WRITE(IOUT,*) initial product state. No v-v or P-R energy transfer
RETURN
END

SUBROUTINE TMGTIR
IMPLICIT DOUBLE PRECISION(A-H,J-Z)
IMPLICIT INTEGER (I)
IMPLICIT CHARACTER*6 (A)
CHARACTER*3 UNITI
INTEGER VMUNIT,VMONLY
COMMON/IO/INP,IOUT,ISAIVE,ISUP,IDEBUG,INCAT,INOV,VMONLY,VMUNIT(0:13)
COMMON/ENERGY/ELEV(0:10),G(20) TEMP
SUB=SPRUB('TMGTIR')
115 WRITE(IOUT,*) Input the temperature. <Default: 300 K>
  IF(INOV.EQ.1) CALL NOVICE(50)
  READ(INP,*,END=903) TEMP,UNITI
  GO TO 904
903 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 115
ELSE
  CALL DPALTC(TEMP,300.000)
  CALL DPALTC(UNITI,KE)
ENDIF
904 WRITE(ISAIVE,*) TEMP,UNITI
  CALL SIU(50,TEMP,UNITI)
  RETURN
  END

SUBROUTINE CONCN
IMPLICIT DOUBLE PRECISION(A-H,J-Z)
IMPLICIT INTEGER (I)
IMPLICIT CHARACTER*6 (A)
CHARACTER*3 UNITI
INTEGER VMUNIT,VMONLY
COMMON/IO/INP,IOUT,ISAIVE,ISUP,IDEBUG,INCAT,INOV,VMONLY,VMUNIT(0:13)
COMMON/POP/NV(0:10),G(20),NDUT,CONCA,CONCBC,CONCM,JCOLL(3)
SUB=SPRUB('CONCN')
110 WRITE(IOUT,*) What is the unit of concentration used?
  WRITE(IOUT,*) '<Default: mo/molar>'
  IF(INOV.EQ.1) CALL NOVICE(80)
  READ(INP,*,END=901) UNITI
  GO TO 901
901 IF(IOUT.EQ.IOUT) THEN
  INP=5
  IOUT=6
  GO TO 110
ELSE
  CALL DPALTC(UNITI,MO)
ENDIF
902 WRITE(ISAIVE,*) UNITI
C Dummy call to SIU, just to check if units for concentration
C are kosher.
  CALL SIU(80,1.000,UNITI)
111 WRITE(IOUT,*) What is the concentration of species A?
  WRITE(IOUT,*) '<Default: 1.0D-6>'
  READ(INP,*,END=902) CONCA
  GO TO 903
903 IF(IOUT.EQ.IOUT) THEN
  INP=5
  IOUT=6
112 What happened to IFLAG...?
  GO TO 111
  ELSE
    CALL DPALTC(CONCA,1.00D-6)
  ENDIF
903 WRITE(ISAIVE,*) CONCA
  CALL SIU(80,CONCA,UNITI)
112 WRITE(IOUT,*) What is the concentration of species BC?
  WRITE(IOUT,*) '<Default: 1.0D-6>'
  READ(INP,*,END=904) CONCBC
  GO TO 905
904 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 112
  ELSE
    CALL DPALTC(CONCBC,1.00D-6)
  ENDIF
905 WRITE(ISAIVE,*) CONCBC
  CALL SIU(80,CONCBC,UNITI)
113 WRITE(IOUT,*) What is the concentration of relaxant gas?
  WRITE(IOUT,*) '<Default: 1.0D-1>'
  READ(INP,*,END=906) CONCM
  GO TO 907
906 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 113
  ELSE
    CALL DPALTC(CONCM,1.00D-1)
  ENDIF
907 WRITE(ISAIVE,*) CONCM
  CALL SIU(80,CONCM,UNITI)
C The three gases present in the default concentrations give a
C combined pressure of about 60 torr.
  RETURN
  END

```

```

SUBROUTINE PIPROT(ANMETH)
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(5)
  COMMON/IO/INP,IOUT,ISAVE,ISUP,ICEBUS,INDAT,INCV,VONLY,VUNIT(0:23)
  SSUB=SPRUB('PIPRM')
  C The structure of the IF and GO TO is meant to simulate
  C a DOWHILE loop (do while IFLAG is less than 2).
  C
  IFLAG=0
108 WRITE(IOUT,*) ' Indicate the desired method of solution: '
  WRITE(IOUT,*) ' 1) Vibrational/Rotational Solution: '
  WRITE(IOUT,*) ' 2) eigenvalue solution (steady state
  WRITE(IOUT,*) ' 3) both. approximation: '
  WRITE(IOUT,*) ' 4) time integration. '
  WRITE(IOUT,*) ' 5) eigenvalue solution (steady state
  WRITE(IOUT,*) ' 6) both. approximation: '
  READ(INP,*) END=900) ANMETH
  GO TO 901
900 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 108
ELSE
  CALL DFALTI(ANMETH,6)
ENDIF
901 WRITE(ISAVE,*) ANMETH
IF(ANMETH.LT.1.OF(ANMETH,GT,6) THEN
  WRITE(IOUT,*) ' Error in method specification ANMETH
  Choose 1, 2, 3, 4, 5 or 6: '
  IFLAG=IFLAG+1
  IF(IFLAG.LE.1) GO TO 108
  CALL ISTERM(ANMETH,SSUB)
ENDIF
  C Check to see whether both vib/rotational solutions are desired
  C or only vibrational solutions. If only the latter are desired
  C flag is set so that the user does not receive prompts relating
  C to the rotational levels. (VONLY = 1 means vibrational only.)
  VONLY=0
  IF(ANMETH.GE.4.AND.ANMETH.LE.6) THEN
  ANMETH=ANMETH-3
  VONLY=1
ENDIF
  RETURN
END

SUBROUTINE PIPRNU(ANGNU)
  C Each subroutine for modelling the potential curve might find
  C rotational levels implicitly, or it will call a
  C further subroutine PIPROT, for determining the rotational
  C energy levels (unless
  C the user has specified the curve is obtained from a file of E(v, J)
  C being read in, or spectroscopically
  C so there is no need to determine how the rotational levels
  C are obtained since this is implicit in the spectroscopic eqn.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(5)
  COMMON/IO/INP,IOUT,ISAVE,ISUP,ICEBUS,INDAT,INCV,VONLY,VUNIT(0:23)
  SSUB=SPRUB('PIPRNU')
  IFLAG = 0
102 WRITE(IOUT,*) ' The potential curve is modelled on: '
  WRITE(IOUT,*) ' 1) Simple Harmonic Oscillator
  WRITE(IOUT,*) ' 2) 3-parameter Morse function
  WRITE(IOUT,*) ' 3) File of v, E(v) pairs
  WRITE(IOUT,*) ' 4) Partial spectroscopic fit
  IFLAG = 0
  IF(VONLY.EQ.0) THEN
  WRITE(IOUT,*) ' --no. waxes etc. '
  WRITE(IOUT,*) ' (vibrational constants only)
  WRITE(IOUT,*) ' 5) Spectroscopic fit.
  IF(VONLY.EQ.0) THEN
  WRITE(IOUT,*) ' --wv, waxes, etc., alphae, Be, etc.
  WRITE(IOUT,*) ' 6) File of v, J, E(v,J) pairs
  ENDIF
  READ(INP,*) END=900) ANGNU
  GO TO 901
900 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 102
ELSE
  CALL DFALTI(ANGNU,1)
ENDIF
901 WRITE(ISAVE,*) ANGNU
IF (ANGNU.EQ.1) THEN
  CALL SHO
  C Once MAXV is found in SHO, the user selects the method of
  C determining the rotational levels.
  IF(VONLY.EQ.0) CALL PIPROT
  ELSEIF (ANGNU.EQ.2) THEN
  CALL MORSE
  IF(VONLY.EQ.0) CALL PIPROT
  ELSEIF (ANGNU.EQ.3) THEN
  CALL READIN(11,ELEV)
  IF(VONLY.EQ.0) CALL PIPROT
  ELSEIF (ANGNU.EQ.4) THEN
  CALL SPEC(0)
  C The argument of SPEC is a flag to signify rotational constants
  C will also be used--hence, the appropriate spec. eqn. must be
  C used in the subroutine SPEC.
  C However, first we must find out the value of MAXJ.
  IF(VONLY.EQ.0) CALL JMAX
  CALL SPEC(0)
  IF(VONLY.EQ.0) CALL PIPROT
  ELSEIF (ANGNU.EQ.5.AND.VONLY.EQ.0) THEN
  CALL SPEC(1)
  ELSEIF (ANGNU.EQ.6.AND.VONLY.EQ.0) THEN
  CALL READV(12,ELEV)
ELSE
  WRITE(IOUT,*) ' Error in potential curve specification ANGNU
  IF(VONLY.EQ.0) THEN
  WRITE(IOUT,*) ' Specify 1, 2, 3, 4, 5 or 6.
  ELSE
  WRITE(IOUT,*) ' Specify 1, 2, 3 or 4.
  ENDIF
  IFLAG = IFLAG + 1
  IF (IFLAG.LE.1) GO TO 102
  CALL ISTERM(ANGNU,SSUB)
ENDIF
  CALL PRINTV(10,ELEV)
  C ...to print out ELEV
  RETURN
END

```

```

SUBROUTINE SHO
C Developed for the model of a truncated harmonic oscillator.
C SHO finds the (v,0) row of the ELEV array for v=0 to MAXV.
C the letter which is also determined in this subroutine.
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
C CHARACTER*2 UNITL, UNITR
COMMON/ENERGY/ELEV(0:10,0:20),TEMP
COMMON/TO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:13)
COMMON/CGNNT/MAXV, MAXN(0:15), MAXUN, INPTS, ICOLL
JSUB=SPRUB/SHO
C We have to know MAXV before we find the MAXV of each v-level.
IFLAG=0
101 WRITE(IOUT,*) 'What is the vibrational spacing'
* ' and its units? <default: 2900 cm >'
C
IF(INV.EQ.1) CALL NOVICE(10)
READ(INP,*) END=501; EDEL0,UNITL
GO TO 502
901 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 101
ELSE
CALL SPALTD(EDEL0,2901,000)
CALL SPALTC(UNITL, 'CM')
ENDIF
502 WRITE(ISAVE,*) EDEL0,UNITL
CALL SIU(10, EDEL0,UNITL)
903 WRITE(IOUT,*) 'What is the dissociation energy'
* ' and its units? <default: 4.62 eV >'
IF(INV.EQ.1) CALL NOVICE(10)
READ(INP,*) END=504; DE,UNITR
GO TO 505
904 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 903
ELSE
CALL SPALTD(DE,4.6200)
CALL SPALTC(UNITR, 'EV')
ENDIF
905 WRITE(ISAVE,*) DE,UNITR
CALL SIU(10, DE,UNITR)
MAXV= (DE/EDEL0) / 1/2
C ...this is based on the SHO equation we(v-1/2)h*nu.
701 FORMAT(' A De of ',F10.2,' eV/mole and a vibrational level',/,
* ' spacing of ',F9.2,' eV/mole corresponds to ',/,
* ' v-levels from 0 to ',I3,')
IF(MAXV.GT.(MAXUN-1)) THEN
702 FORMAT(' There are more than ',I3,' vibrational
* ' levels, so vmax is being set to ',I3,')
MAXV=MAXUN-1
ENDIF
DO 100 V=0,MAXV
ELEV(V,0)=DBLE(V)*EDEL0
100 CONTINUE
105 WRITE(IOUT,703) MAXUN-1
703 FORMAT(' Respectively v=999, De-type =1,/,
* -type 0,/, ' new MAXV value =Specify (between 1 and ',I3,')/,
* ' corresponding to the highest v-level you want. <default: ',I3,')
READ(INP,*) END=506) ANMAXV
GO TO 907
906 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 105
ELSE
CALL SPALTC(ANMAXV,0)
ENDIF
907 WRITE(ISAVE,*) ANMAXV
IF (ANMAXV.EQ. -1) GOTO 101
IF (ANMAXV.NE. 0) THEN
IF (ANMAXV.LT.0 .OR. ANMAXV.GT.MAXUN) THEN
WRITE(IOUT,*) 'Input error on maximum v-level ANMAXV'
IFLAG = IFLAG + 1
IF (IFLAG.LE.1) GOTO 105
STOP 1000
ENDIF
MAXV = ANMAXV
ENDIF
RETURN
END

SUBROUTINE PIKROT
C This subroutine determines how the J-levels for the given v-levels
C will be superimposed on the v-levels.
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT DOUBLE PRECISION (B-H,F,N-U,W-Z)
C IMPLICIT CHARACTER*6 (S)
COMMON/TO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:13)
JSUB=SPRUB/PIKROT
IFLAG = 0
WRITE(IOUT,*) 'The rotational energy levels are obtained through:
WRITE(IOUT,*) ' 1) A rigid rotor approximation.
WRITE(IOUT,*) ' 2) A rigid rotor approximation
* ' equation with B_e, D_e etc.
PEAC(INP,*) END=900) ANROT
GO TO 901
900 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 105
ELSE
CALL IFAULT(ANROT,1)
ENDIF
901 WRITE(ISAVE,*) ANROT
IF (ANROT.EQ.1) THEN
CALL JMAX
CALL RIROT
ELSEIF (ANROT.EQ.2) THEN
CALL JMAX
CALL SPERR
C Conceivably, another option for E(J) could have been specified:
C that of reading them in for each given v-level.
C At the moment, we will leave this to be treated as a subset of
C the READV program, where the energies of both the v and J
C levels are read in at once.
ELSE
WRITE(IOUT,*) 'Error in rotational E ' specification: ANROT
WRITE(IOUT,*) 'Specify 1, or 2.'
IFLAG = IFLAG + 1
IF (IFLAG.LE.1) GO TO 105
CALL ITERM(ANROT,JSUB)
ENDIF
RETURN
END

```

```

SUBROUTINE JMAX
C This subroutine determines the maximum J-level for each given
C v-level.
C ** Eventually, this program should allow one to specify
C ** a maximum energy, and then calculate all v,J levels
C ** available within the energy bounds.
IMPLICIT DOUBLE PRECISION (B-H,K-N,U,W-Z)
IMPLICIT INTEGER (A,I-O,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP,ICOUNT,ISAVE,ISUP,IDEREG,INEAT,INDY,ONLY,UNIT,MODE
COMMON/CONST/MAXV(MAXJ(0:10),MAXUN,INPTS,ISOLL)
C114 COMMON/MASS/EMASS,CMASS,AMASS,PEEMAS,BRAD,CRAD,ARAD,RE
      SSUB=SPRSUB/JMAX
C Check to see if JMAX has already been called. (This could happen
C if ANRMAX=1) If no then return immediately from this subroutine.
IF(MAXJ(0).NE.0 .OR. ONLY.EQ.1) RETURN
IFLAG = 0
107 WRITE(IGOUT,*) 'The maximum rotational level available:
      WRITE(IGOUT,*) ' 1) as a fixed number for all v-levels'
      WRITE(IGOUT,*) ' 2) differs for each v-level'
      READ(INP,*,END=906) ANRMAX
      IF(ANP.EQ.ISUP) THEN
        GO TO 901
      ELSE
        CALL DFALTI(ANRMAX,1)
      ENDIF
901 WRITE(ISAVE,*) ANRMAX
      IFLAG = IFLAG + 1
      IF(ANRMAX.EQ.1) THEN
630 WRITE(IGOUT,*) 'What is the maximum J level
          * for each v-level? (max. 201 <Default: 10>'
          READ(INP,*,END=906) MAX
          GO TO 904
      IF(IGOUT.EQ.ISUP) THEN
600 IF(IGOUT.EQ.ISUP) THEN
          INP=5
          IOUT=6
          GO TO 902
        ELSE
          CALL DFALTI(MAX,10)
        ENDIF
904 WRITE(ISAVE,*) MAX
          IF(MAX.GT.101 .OR. MAX.LT.1) THEN
            WRITE(IGOUT,*) 'Max. J value out of bounds..MAX
              IFLAG=IFLAG+1
              IF(IFLAG.LE.1) GO TO 903
              CALL ITERM(MAX,SSUB)
            ENDIF
            DO 100 I = 0,MAXV
              MAXJ(I) = MAX
            CONTINUE
100 ELSEIF (ANRMAX.EQ.1) THEN
605 WRITE(IGOUT,*) 'Input Jmax for each vibrational level.
          * (Input MAXV values: <Default: 10 for all v-levels>'
          READ(INP,*,END=906) (MAXJ(I), I=0,MAXV)
          GO TO 907
906 IF(IGOUT.EQ.ISUP) THEN
          INP=5
          IOUT=6
          GO TO 905
        ELSE
          DO 908 I=0,MAXV
            CALL DFALTI(MAXJ(I),10)
          CONTINUE
908 ENDIF
907 WRITE(ISAVE,*) (MAXJ(I), I=0,MAXV)
      ELSE
        WRITE(IGOUT,*) 'Error in checking Jmax specification. ANRMAX
          WRITE(IGOUT,*) ' Specify 1 or 2.
          IFLAG = IFLAG + 1
          IF (IFLAG.LE.1) GO TO 107
          CALL ITERM(ANRMAX,SSUB)
        ENDIF
        RETURN
      END

```

```

SUBROUTINE RIROT
C This subroutine will fill in the ELEV array up to ELEV(MAXV, MAXV)
C by determining the J-levels for each v-level.
C RIROT should only be called after JMAX is called and only
C when VONLY=0.
C A rigid rotor approximation assumes no coupling between
C V and J (the alpha_0 term in the spectroscopic equation)
C and no centrifugal distortion (alpha_2 term).
IMPLICIT DOUBLE PRECISION (B-H,F,M,H,W-Z)
IMPLICIT INTEGER (A,I-O,L-M,V)
IMPLICIT CHARACTER*6 (S)
REAL*8 ARAD,ANRMAX,AVOGAD
COMMON/ENERGY/ELEV(0:10,0:200) TEMP
COMMON/CONST/MAXV(MAXJ(0:10),MAXUN,INPTS,ISOLL)
COMMON/PHYSCT/PLANCK,FEOLTS,PGAS,SCIEPI,CLIGHT,PI,HAVER
COMMON/MASS/EMASS,CMASS,AMASS,PEEMAS,BRAD,CRAD,ARAD,RE
PARAMETER (AVOGAD=6.02205E23)
SSUB=SPRSUB/RIROT
IF (RE.EQ.0.000) CALL SONDIC
C Calculate E_L and then the rotational energy spacing...
C The reduced mass was found in RMNYS and is used to
C calculate XMOIN, the moment of inertia.
C Value for PLANCK -- pass through phys. cts. common.
C Formula from W. J. Moore, Physical Chemistry
C (1962) p.587.
XMOIN= REEMAS * RE * RE/AVOGAD
      divide by N_Av because the masses are per mole and the
      moment of inertia of a single molecule is desired.
SE= PLANCK/(8.000 * PI*PI*XMOIN)
DO 100 V=0,MAXV
  DO 100 J=0,MAXJ(V)
    INJ = SELE(J*(J+1))
    ELEV(V,J)=SE*PLANCK*DOU*AVOGAD + ELEV(V,0)
    here, we multiply by N_Av because the energy of
    the rotational level is only given for each molecule.
100 CONTINUE
      RETURN
      END

```

```

SUBROUTINE SPECRO
C This subroutine will fill in the ELEV array up to ELEV(MAXV,MAXN)
C by determining the J-levels for each v-level.
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,F,I,J,L-M,V)
IMPLICIT CHARACTER*8 (S)
DIMENSION RE(6),
COMMON/IO/INP, IOUT, ISAVE, ISUP, ISEBU, INEAT, INNV, VONLY, VUNIT(6,10)
COMMON/ENERGY/ELEV(6,10), ITEMP
COMMON/COUNT/MAXV, MAXN(6,10), INPTS, ITEMP
CHARACTER*8 UNIT
ISUB=IPRIBI/3 SPECRO
IFLAG=0
900 WRITE(IOUT,*) ' How many spectroscopic constants for the
      potential energy will you be inputting? (default: 3)
      READ(INP,*) END=901 IPE
      GO TO 902
901 IF(IOUT.EQ.ISEBU) THEN
      INP=5
      IOUT=6
      GO TO 900
      ELSE
      CALL DFALTI(IIRE, I)
      ENDDIF
902 WRITE(ISAVE,*) IRE
      IF(IFLAG.LE.1) THEN
      WRITE(IOUT,*) ' Number of spectroscopic constants specified
      is out of bounds. IPE Please re-specify.'
      IFLAG=IFLAG+1
      GO TO 900
      ELSE
      CALL ITERM(IIRE, ISUB)
      ENDDIF
903 WRITE(IOUT,*) ' What is the unit used for the constants?
      READ(INP,*) END=904 UNIT
      GO TO 905
904 IF(IOUT.EQ.ISEBU) THEN
      INP=5
      IOUT=6
      GO TO 903
      ELSE
      CALL DFALTO(UNIT, CM)
      ENDDIF
905 WRITE(ISAVE,*) UNIT
906 WRITE(IOUT,*) ' Input the spectroscopic constants
      alpha_s, B, D, e
      s, separated by commas. IPE constants?
      IF(INNV.EQ.1) CALL NCVTE(10)
      READ(INP,*) END=907 (RE(1), I=1, IPE)
      GO TO 908
907 IF(IOUT.EQ.ISEBU) THEN
      INP=5
      IOUT=6
      GO TO 906
      ELSE
      CALL DFALTO(RE(1), 0.307161)
      CALL DFALTO(RE(2), 10.593416)
      ... values for HCl: Park et al. J Mol Spectr 17 (1965) 111.
      ENDDIF
908 WRITE(ISAVE,*) IRE(1), I=1, IRE
      GO TO 909 I=1, IRE
      CALL SIU(10, RE(1), UNIT)
909 CONTINUE
      WRITE(IOUT,*) ' v J E(v, J)
      DO 600 J=0, MAXN
      DO 700 J=1, MAXN(J)
      DV=CHL(IV)
      DV=CHL(IV)
      C The first constant read in is, by convention, alpha_s
      ELEV(V, J)=ELEV(V, J)- RE(1)*J*(J+1)*IV*0.5
      GO 800 I=2, IRE
      ELEV(V, J)=ELEV(V, J)- RE(1)*J*(J+1)*IV**2
800 CONTINUE
700 CONTINUE
600 CONTINUE
RETURN
END

```

```

SUBROUTINE MORSE
MORSE will find the (v, 0) row of the ELEV array for v=0 to MAXV,
which is also determined in this subroutine.
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,F,I,J,L-M,V)
IMPLICIT CHARACTER*8 (S)
REAL*8 AV0, AD, ARAD, AMASS
CHARACTER*8 UNIT, UNITI
COMMON/IO/INP, IOUT, ISAVE, ISUP, ISEBU, INEAT, INNV, VONLY, VUNIT(6,10)
COMMON/ENERGY/ELEV(6,10), ITEMP
COMMON/PHYSIC/PLANG, PEOLE, PGAD, SKEPI, CLIGHT, PI, IANV
COMMON/COUNT/MAXV, MAXN(6,10), MAXN1, INPTS, ITEMP
COMMON/MASS/AMASS, CMAD, AMASS, REIMAG, REAT, SPAD, ARAD, PE
PARAMETER (AV0=0.06, AD=0.0015003,
           ISEBU=IPRIBI/3, MURSE=1)
100 WRITE(IOUT,*) ' Input the dissociation energy and its
      units. (default: 4.62 eV)
      IF(INNV.EQ.1) CALL NCVTE(10)
      READ(INP,*) END=901 DE, UNITI
      GO TO 901
901 IF(IOUT.EQ.ISEBU) THEN
      INP=5
      IOUT=6
      GO TO 100
      ELSE
      CALL DFALTO(DE, 4.62DE)
      CALL DFALTO(UNITI, 'EV')
      ENDDIF
902 WRITE(ISAVE,*) DE, UNITI
      CALL SIU(10, DE, UNITI)
      IF (RE(2).EQ.0) CALL BONDIS
903 WRITE(IOUT,*) ' Input beta and the units of distance
      s (default: 1.0051 bohr)
      IF(INNV.EQ.1) CALL NCVTE(10)
      READ(INP,*) END=904 BETA, UNITI
      GO TO 907
904 IF (IOUT.EQ.ISEBU) THEN
      INP=5
      IOUT=6
      GO TO 904
      ELSE
      CALL DFALTO(BETA, 1.0051DE)
      ... from Wright and Barclay Chem Phys (1984)
      *** get full reference
      CALL DFALTO(UNITI, 'BO')
      ENDDIF
907 WRITE(ISAVE,*) BETA, UNITI
      DE=DE*DE
      C The units for beta are input as distance; the inverse conversion
      factor is needed, so the value of BETA is divided by the
      distance conversion factor (XKTOME)
      XKTOME=1.0
      CALL SIU(10, XKTOME, UNITI)
      BETA = BETA/XKTOME
      C Formula from P.W.Atkins, Physical Chemistry, Freeman & Co. (1964)
      HBARBE = AV0*AD*BETA*PLANG/(2.0DE*PI)
      WE = 2.0DE*HBARBE*DE*RTICE/(2.0DE*PE*MASS)
      WENE = HBARBE*HBARBE/(2.0DE*PE*MASS)
      C Solve quadratic equation
      wxwv=1/2**2 - wv(1/2) - DE = 0
      CALL QUAD(WENE, -WE, DE, WX1, WX2)
      C Both roots will be positive. We must decide which root is
      C physically reasonable (i.e. which is smallest since during
      C vibrational excitation to dissociation, the smallest v value
      C will be reached first and dissociation will occur.
      XX=MIN(WX1, WX2)
      HBARV = WX-0.5
      C Check to make sure the value for MAXV is positive.
      IF(MAXV.LE.0) CALL ITERM(MAXV, ISUB)
      WRITE(IOUT,*) ' A De of DE (euler/mole) an Be of PE
      A meters and a Beta of BETA (meters) have MAXN1
      vibrational levels.
      IF(MAXV.GT. MAXN-1) THEN
      WRITE(IOUT,*) ' There are more than MAXN vibrational
      levels so the max v is being set to MAXN-1
      MAXV=MAXN-1
      ENDDIF
101 WRITE(IOUT,*) MAXN-1
102 FORMAT (/, 'Respectively De, Be, Beta (units with each) /
      & --type 1 -- / accept max. v value--type 0 -- / specify new
      & max. v value--type 1 to 10 corresponding to the highest v
      & level you want. (Default: maxv=3)
      READ(INP,*) END=908 ANMAXV
      GO TO 909
908 IF(IOUT.EQ.ISEBU) THEN
      INP=5
      IOUT=6
      GO TO 103
      ELSE
      CALL DFALTI(ANMAXV, 3)
      ENDDIF
909 WRITE(ISAVE,*) ANMAXV
      IF(ANMAXV.EQ.-1) GO TO 102
      IF(ANMAXV.EQ.0) THEN
      WRITE(IOUT,*) ' MAXV = MAXV
      ELSE IF(ANMAXV.GE.1 AND ANMAXV.LE.MAXN) THEN
      MAXV=ANMAXV
      ELSE
      IFLAG=IFLAG+1
      WRITE(IOUT,*) ' ANMAXV is out of range. Re-specify.
      IF(FLAG.LE.1) GO TO 103
      CALL ITERM(ANMAXV, ISUB)
      ENDDIF
      DO 200 J=0, MAXV
      DV=CHL(IV)
      ELEV(V, J)=WE*(IV+0.5)+WENE*(IV+0.5)**2
100 CONTINUE
RETURN
END

```

```

SUBROUTINE BONDIS
C To determine the equilibrium bond length of species BC.
IMPLICIT DOUBLE PRECISION (B-H,F,N-O,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
REAL*8 ARAD,AMASS
CHARACTER*2 UNIT2
COMMON/IO/INP,IOUT,ISAVE,ISUP,IDESUB,INDAT,INOV,WONLY,VUNIT(0:13)
COMMON/MASS/EMASS,CMASS,AMASS,PCDMAS,BRAD,CRAD,ARAD,PE
ISUB=$PREBU('BONDIS')
BCRAD=BRAD+CRAD
C
C ...default bond length is sum from RWNEYS.
900 WRITE(IOUT,*) Input the equilibrium bond distance
  4 and its units. (Default: BCRAD, bo
  IF(INOV.EQ.1) CALL NOVICE(30)
  READ(INP,*,END=902) RE,UNIT2
  GO TO 904
903 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 903
ELSE
  CALL DPALTD(IRE,BCRAD)
  CALL DPALTD(UNIT2,'BO')
  ENDDIF
904 WRITE(ISAVE,*) RE,UNIT2
  CALL SIU(30,RE,UNIT2)
  RETURN
END

```

```

SUBROUTINE CHART(INDEX1, INDEX2, INDEX3)
C CHART displays the periodic table, and acts as a look-up for the
  relative atomic masses.
C CHART returns via its arguments, the atomic numbers of A, B and C.
IMPLICIT DOUBLE PRECISION(A-H,O-Z)
IMPLICIT INTEGER(I-N)
IMPLICIT CHARACTER*6 (S)
INTEGER*4 VUNIT, WONLY
COMMON/IO/INP,IOUT,ISAVE,ISUP,IDESUB,INDAT,INOV,WONLY,VUNIT(0:13)
ISUB=$PREBU('CHART')
IFLAG=0
WRITE(IOUT,*)
WRITE(IOUT,10)
10 FORMAT(1X,'PERIODIC CHART OF THE ELEMENTS')
WRITE(IOUT,11)
11 FORMAT(1X,' H He Li Be B C N O F Ne ')
WRITE(IOUT,12)
12 FORMAT(1X,' Na Mg Al Si P S Cl Ar ')
WRITE(IOUT,13)
13 FORMAT(1X,' K Ca Sc Ti V Cr Mn Fe Co Ni Cu ')
WRITE(IOUT,14)
14 FORMAT(1X,' Zn Ga Ge As Se Br Kr ')
WRITE(IOUT,15)
15 FORMAT(1X,' Rb Sr Y Zr Nb Mo Tc Ru Rh Pd Ag ')
WRITE(IOUT,16)
16 FORMAT(1X,' Cd In Sn Sb Te I Xe ')
WRITE(IOUT,17)
17 FORMAT(1X,' Ba La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er ')
WRITE(IOUT,18)
18 FORMAT(1X,' Tm Yb Lu ')
WRITE(IOUT,19)
19 FORMAT(1X,' Hf Ta W Re Os Ir Pt Au ')
WRITE(IOUT,20)
20 FORMAT(1X,' Hg Tl Pb Bi Po At Rn ')
WRITE(IOUT,21)
21 FORMAT(1X,' Fr Ra Ac Th Pa U Np Pu Am Cm Bk Cf Ee Fm ')
WRITE(IOUT,22)
22 FORMAT(1X,' Md No Lr ')
WRITE(IOUT,23)
23 FORMAT(1X,' 101 102 103 ')
101 WRITE(IOUT,*) Select the number of three elements from the
  4 periodic table shown above.
  READ(INP,*) INDEX1,INDEX2,INDEX3
  IF(INDEX1.LT.1.OR.INDEX1.GT.106) THEN
  IFLAG=IFLAG+1
  WRITE(IOUT,*) 'Incorrect selection of first element number.'
  WRITE(IOUT,*) 'Number should lie between 1 and 106.'
  IF(IFLAG.LE.1) GO TO 101
  CALL ITERM(INDEX1,$SUB)
  ENDDIF
  IF(INDEX2.LT.1.OR.INDEX2.GT.106) THEN
  IFLAG=IFLAG+1
  WRITE(IOUT,*) 'Incorrect selection of second element number.'
  WRITE(IOUT,*) 'Number should lie between 1 and 106.'
  IF(IFLAG.LE.1) GO TO 101
  CALL ITERM(INDEX2,$SUB)
  ENDDIF
  IF(INDEX3.LT.1.OR.INDEX3.GT.106) THEN
  IFLAG=IFLAG+1
  WRITE(IOUT,*) 'Incorrect selection of third element number.'
  WRITE(IOUT,*) 'Number should lie between 1 and 106.'
  IF(IFLAG.LE.1) GO TO 101
  CALL ITERM(INDEX3,$SUB)
  ENDDIF
  RETURN
END

```

```

SUBROUTINE QUAD(AA,BB,CC,XX1,XX2)
C To find the roots of  $AX^2 + BX + C = 0$ .
IMPLICIT DOUBLE PRECISION (A-H,O-Z)
IMPLICIT CHARACTER*6 IS
INTEGER*4 VUNIT,VOONLY
COMMON/IO/INP,IOUT,ISAVE,ISUP,ISERES,INDAT,INOV,VOONLY,VUNIT(6),DS
SSUB=SPRUB('QUAD ')
C Set value of threshold so that calculation can be checked.
THRESH=1.0E-06
DTERM=BB*BB-4.0*AA*CC
IF(DTERM.LT.0.0) THEN
WRITE(IOUT,*) ' Error in QUAD--determinant < 0'
CALL ABTERM(DTERM,SSUB)
ENDIF
DTERM=DSQRT(DTERM)
XX1=(-BB-DTERM)/(2.0*AA)
XX2=(-BB+DTERM)/(2.0*AA)
C Check the parameters determined.
RESULT=AA*XX1+BB*XX2+CC
IF(RESULT.LT.-THRESH .OR. RESULT.GT.THRESH) THEN
WRITE(IOUT,*) ' Error in QUAD for XX1 calculation.'
CALL ABTERM(XX1,SSUB)
ENDIF
RESULT=AA*XX2+BB*XX1+CC
IF(RESULT.LT.-THRESH .OR. RESULT.GT.THRESH) THEN
WRITE(IOUT,*) ' Error in QUAD for XX2 calculation.'
CALL ABTERM(XX2,SSUB)
ENDIF
RETURN
END

SUBROUTINE SPEC(IROT)
C If the argument in the call to this subroutine is 0, the
C rotational levels will be calculated using spectroscopic parameters.
IMPLICIT DOUBLE PRECISION (B-H,K,N,O,W-Z)
IMPLICIT INTEGER (A,I,J,L,M,V)
IMPLICIT CHARACTER*6 IS
DIMENSION WE(6),RE(6)
COMMON/IO/INP,IOUT,ISAVE,ISUP,ISERES,INDAT,INOV,VOONLY,VUNIT(6),DS
COMMON/ENERGY/ELEV(6),LE(6),TEMP
COMMON/COUNT/MAXV,MAXU(6),MAXUN,INPTR,ICOLL
CHARACTER*6 IWE
ISUB=ISPRUB('SPEC ')
DO 200 I=1,6
200 WE(I)=0.0
101 WRITE(IOUT,*) ' What is the dissociation energy?'
* <Default: 4.6175 eV >
IF(INOV.EQ.1) CALL NVOICE(10)
READ(INP,*) END=900, DE,UNIT
GO TO 901
900 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 101
ELSE
CALL DFAULT(DE,4.6175D0)
...default values for HCl from Huber and Herzberg (1979).
CALL DFAULT(UNIT, 'eV')
ENDIF
901 WRITE(ISAVE,*) DE,UNIT
CALL SIU(10,DE,UNIT)
902 WRITE(IOUT,*) ' How many spectroscopic constants for the
* vibrational energy will you be inputting? <Default: 3>'
READ(INP,*) END=903, IWE
GO TO 904
903 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 902
ELSE
CALL DFAULT(IWE,3)
ENDIF
904 WRITE(ISAVE,*) IWE
905 WRITE(IOUT,*) ' Input the spectroscopic constants  $W_e$ ,  $W_{e1}$ ,
*  $W_{e2}$ , etc. separated by commas (between 1 and 6 constants).
* and at the end specify the unit to be used
* i.e. 'cm-1' means cm-1.
WRITE(IOUT,*) '<Default: 1990.9463, 52.8155, 0.3244, cm-1>'
IF(INOV.EQ.1) CALL NVOICE(10)
READ(INP,*) END=906, WE(1),IWE,UNIT
GO TO 907
906 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 905
ELSE
CALL DFAULT(WE(1),1990.9463D0)
CALL DFAULT(WE(2),52.8155D0)
CALL DFAULT(WE(3),0.3244D0)
...values for HCl: Rank et al. J Mol Spectr 17 (1965) 133.
CALL DFAULT(UNIT, 'cm')
ENDIF
907 WRITE(ISAVE,*) WE(1),IWE,UNIT
DO 200 I=2,6 IWE
CALL SIU(10,WE(I),UNIT)
200 CONTINUE
C The Newton-Raphson algorithm is used to determine the maximum v-level
C to a tolerance of 0.01 (relative).
C The initial guess is based on a second order quadratic.
*  $W_{e1}v - 1/2 W_{e1}^2 v^2 - W_{e2}v^3 = 0$ 
C
CALL QUAD(WE(3),-WE(2),DE,XX1,XX2)
KOLD=MIN(X1,XX2)
C Both roots should be positive; check to make sure this is true.
IF(KOLD.LT.0) CALL ABTERM(KOLD,SSUB)
CALL NEWTR(KOLD,FX,WE,IWE)
MAXV=KOLD-0.5)
WRITE(IOUT,*) ' At  $W_e$  of WE(1) and an  $W_{e1}$  of WE(2)
* 1000s/molec gave MAXV=1
* vibrational levels
IF(MAXV.GT.MAXUN-1) THEN
WRITE(IOUT,*) ' There are more than MAXUN vibrational
* levels so the max v is being set to MAXUN-1
17 FORMAT('14')
MAXV=MAXUN-1
ENDIF
909 WRITE(IOUT,*) MAXV-1
101 FORMAT(' Respecify  $D_e$  and spectroscopic constants'
* --type -1 / accept max. v value--type 0: specify new max. v
* value--type 1 to 3 / corresponding to the highest v
* level you want. <Default: 3>'
READ(INP,*) END=910, ANMAXV
GO TO 911
910 IF(IOUT.EQ.ISUP) THEN
INP=5
IOUT=6
GO TO 909
ELSE
CALL DFAULT(ANMAXV,3)
ENDIF
911 WRITE(ISAVE,*) ANMAXV
IF(ANMAXV.EQ.-1) GO TO 101
IF(ANMAXV.EQ.0) THEN
WRITE(IOUT,*) ' MAXV = MAXV
ELSEIF(ANMAXV.GE.1.AND.ANMAXV.LE.MAXUN) THEN
MAXV=ANMAXV
ELSE
IFLAG=IFLAG+1
IF(IFLAG.GE.1) THEN
WRITE(IOUT,*) ' Answer is out of bounds; respecify ANMAXV
GO TO 909
ENDIF

```

```

STOP 1011
ENDIF
IF (IRROT.EQ.0) THEN
912 WRITE (IOUT,*) ' How many spectroscopic constants for the
rotational energy will you be inputting? <default: 2>'
1 WRITE (IOUT,*) ' (You can specify up to 6, including v-J
coupling terms.)'
4 READ (INP,*,END=913) IRE
GO TO 914
913 IF (IOUT.EQ.ISUP) THEN
INP=6
IOUT=6
GO TO 910
ELSE
CALL DFALTI(IRE,3)
ENDIF
914 WRITE (ISAVE,*) IRE
IF (IRE.LT.1.OR.IRE.GT.6) THEN
IF (IFLAG.LE.1) THEN
WRITE (IOUT,*) ' Number of rotational constants specified
is out of bounds. IRE Please re-specify.'
GO TO 912
ELSE
CALL ISTERM(IRE,ISUB)
ENDIF
ENDIF
915 WRITE (IOUT,*) ' Input the spectroscopic constants
A 'alpha_e, B_e, D_e,
& etc. separated by comma',
& and at the end, specify the unit to be used.'
READ (INP,*,END=916) (RE(I),I=1,IRE),UNIT
IF (INOV.EQ.1) CALL NOVICE(10)
GO TO 917
916 IF (IOUT.EQ.ISUP) THEN
INP=6
IOUT=6
GO TO 915
ELSE
CALL DFALTD(RE(1),0.1D0)
CALL DFALTD(RE(2),1.0D0)
CALL DFALTD(UNIT,'CM')
ENDIF
917 WRITE (ISAVE,*) (RE(I),I=1,IRE),UNIT
DO 303 I=1,IRE
CALL SIU(30 RE(I),UNIT)
CONTINUE
203 WRITE (IOUT,*) ' v J E(v,J)
ELSE
WRITE (IOUT,*) ' v E(v,0)
ENDIF
DO 500 V=0,MAXV+1
DO 500 J=1,IWE
ELEV(V,0)=WE(1)+(.DBLE(V)+0.5)**2 * ELEV(V,0)
500 CONTINUE
IF (IRROT.EQ.0) THEN
DO 700 J=0,MAXJ(V)+1
C Sign convention for this summation is according to Furplus &
C Porter p.483. Note: Check signs for higher rotational terms (e.g.
C H_e) and higher order vibrational terms (e.g., w_e'y_3)
C The first constant read in is, by convention, alpha_w
DJ=DBLE(J)
ELEV(V,J)= -RE(1)* (DBLE(V)+.5)* (DJ*(DJ+1))
DO 800 I=1,IRE
ELEV(V,J)=ELEV(V,J)+ RE(I)/(DJ*(DJ+1)**((I-1)*(I-1)**I)
600 CONTINUE
WRITE (IOUT,*) V,J,ELEV(V,J)
CONTINUE
700 ELSE
WRITE (IOUT,*) V,ELEV(V,0)
ENDIF
ENDIF
800 CONTINUE
C If the rotational constants have not been specified then
C some other method of calculating rotational energy levels
C must be found.
IF (IRROT.NE.0) CALL PIKROT
RETURN
END

```

```

SUBROUTINE NEWTRA(XOLD,OLDFX WE IWE)
C The Newton-Raphson algorithm is used to determine the maximum v-level
C to a tolerance of 0.01 (relative)
C The initial guess is based on a second order quadratic.
C We(v-1/2)**2 - we(v-1/2) + De = 0
C
IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (C)
COMMON/IO/INP IOUT
DIMENSION WE(6)
SUBEPPSUB(NEWTRA)
NPTOL=0.01
DFDX=0.0
ICOUNT=0
400 FX=OLDFX
C Calculation of function (polynomial) and its derivative.
DO 300 I=1,IWE
FX=WE(I)**(XOLD**I) + FX
DFDX=WE(I)*DBLE(I)**(XOLD**I-1) + DFDX
300 CONTINUE
XNEW=XOLD-FX/DFDX
IF (ICOUNT.GT.ISEVER) THEN
WRITE (IOUT,*) ' Newton Raphson not converging'
CALL ASTERM(XNEW,ISUB)
ENDIF
IF (ICABS(XNEW-XOLD)/XNEW.GT.NPTOL) THEN
ICOUNT=ICOUNT+1
XOLD=XNEW
GO TO 400
ENDIF
RETURN
END

```

```

FUNCTION EXIST(VV,JJ,VV1,JJ1)
C This function simply verifies that, for the v,J -> v1,J1
C transition, the energy levels exist as defined in the potential
C energy curve construction.
IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
LOGICAL*1 EXIST
COMMON/ENERGY/ELEV(0:10,0:30) TEMP
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN, INPTS, ICOLL
C To see whether a transition exists we check whether
C each value lies within the range of existing values calculated
C or specified earlier in the program.
IF (VV1.GE.0 .AND. VV.LE.MAXV .AND.
& VV1.LE.0 .AND. VV.LE.MAXV ) THEN
IF (JJ1.GE.0 .AND. JJ.LE.MAXJ(VV) .AND.
& JJ1.LE.0 .AND. JJ1.LE.MAXJ(VV1) ) THEN
EXIST=.TRUE.
ELSE
EXIST=.FALSE.
ENDIF
ELSE
EXIST=.FALSE.
ENDIF
C We must also ascertain that the initial level is unique
C from the final level.
IF (VV.EQ.VV1 .AND. JJ.EQ.JJ1) EXIST=.FALSE.
RETURN
END

```

```

SUBROUTINE PIRKOU(ANKOUP)
C Select method for obtaining KOUF vector for each v level...
C Where KOUF(v,m) means k(v,m)-->k(v,m+2)
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*6 (S)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, WONLY, VUNIT(0:23)
  COMMON/FVIB/FUPUP(0:10,0:20), FUPD(0:10,0:20), FUP0(0:10,0:20)
  1 FDOWN(0:10,0:20), FDOWN0(0:10,0:20), FDOWN0(0:10,0:20)
  2 FUP(0:10,0:20), FDOWN(0:10,0:20)
  COMMON/SPCIES/ISPEC $SPEC(6,3)
  $SUB=$PRSUB: PIRKOU)
  IFLAG=0
  WRITE(IOUT,701) ($SPEC(1,ISPEC), L=1,6)
  701 FORMAT(1X,6A6)
  140 WRITE(IOUT,*) Rate constants between rotational
  & levels J and J+2 are:
  & * Calculated by scaling according to
  & the rotational energy gap
  WRITE(IOUT,*) ' ' * J) Input from data file
  READ(INP,*) END=900) ANFOUP
  GO TO 901
  900 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 140
  ELSE
    CALL DFALTI(ANFOUP,1)
  ENDF
  901 WRITE(ISAVE,*) ANFOUP
  IF (ANFOUP.EQ.1) THEN
    CALL FROUF
  ELSEIF (ANFOUP.EQ.2) THEN
    CALL READIN(2) KOUF)
  ELSE
    WRITE(IOUT,*) Error in specification of KOUF-->J+2)
  &
  & WRITE(IOUT,*) Answer should be 1, or 2)
  IFLAG = IFLAG + 1
  IF (IFLAG.LE.1) GO TO 140
  CALL ITERM(ANFOUP, $SUB)
  ENDF
  IF(IDEBUG.EQ.1) CALL PRINTV(23,FUOUP)
  RETURN
END

```

```

SUBROUTINE FROUFP
C Calculation of k(v,J-->J+2) rate constant.
C This subroutine assigns KHALT (which determines the
C rate coefficients for vibrational levels based on
C the energy gap). FROUFP determines the rate
C coefficients for rotational levels when there is
C no change in vibrational levels.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*6 (S)
  CHARACTER*3 UNIT
  LOGICAL*1 EXIST
  COMMON/SPCIES/ISPEC $SPEC(6,3)
  COMMON/FVIB/FUPUP(0:10,0:20), FUPD(0:10,0:20), FUP0(0:10,0:20)
  1 FDOWN(0:10,0:20), FDOWN0(0:10,0:20), FDOWN0(0:10,0:20)
  2 FUP(0:10,0:20), FDOWN(0:10,0:20)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, WONLY, VUNIT(0:23)
  COMMON/PHYCCT/PLANCE, FOLITE, RGAS, SCEXPL, CLIGHT, PI, ISEVER
  COMMON/ENERGY/ELEV(0:10,0:20), TEMP
  COMMON/COUNT/MAXV, MAXJ(0:10), MAXUP, INPTS, IROLL
  $SUB=$PRSUB: FROUFP)
  WRITE(IOUT,701) ($SPEC(1,ISPEC), L=1,6)
  701 FORMAT(1X,6A6)
  900 WRITE(IOUT,*) Input k for v=0 J=0 --> v=0 J=2 transition
  & sig to unit: <default: 0.0 "e"/s>
  READ(INP,*) END=901) FOUF(0,0) UNIT
  GO TO 902
  901 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 900
  ELSE
    CALL DFALTD(KOUP(0) 0) 80,000)
    CALL DFALTC(UNIT SE)
  ENDF
  902 WRITE(ISAVE,*) KOUP(0,0) UNIT
  CALL SIU(20,FOUF(0,0) UNIT)
  WRITE(IOUT,*) $$ Rotational Transition Rate Coefficients $$
  WRITE(IOUT,*) ' THE k(v,J-->J+2) CONSTANTS ARE :
  WRITE(IOUT,*) ' J k(v,J-->J+2)
  C user can use value of constant (recip) for alternate scaling.
  C or change function altogether.
  RECIP=1.0
  C Should be rgas to make energy*recip unitless (delz/RT)
  RECIP=RECIP/(RGAS*TEMP)
  C pre-exponential factor needed to return constant k(v,0-->2)
  PREEXP = DEXP(-RECIP/EDEL(0,0,0,2))
  PREEXP = PREEXP*KOUP(0,0)
  DO 10 J=0,MAXJ(0)
    IF (EXIST(0,J,0,J+2))
      KOUP(0,J)=PREEXP*DEXP(RECIP/EDEL(0,J,0,J+2))
  10 CONTINUE
  RETURN
END

```

```

SUBROUTINE FVIBL(ANKO1)
C In the calculation of the rate coefficient for v-->v+1 scaling
C the value from which all other values are scaled must first be
C obtained. This routine prompts for k(0,0-->1,0)
C and if rotational levels exist (i.e., Wonly=0) it prompts
C for k(0,0-->1,2) and k(0,2-->1,0).
C When ANKO1 is 1, Landau-Teller scaling is desired.
C If the argument ANKO1 is 2 use an alternate form of scaling.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*6 (S)
  CHARACTER*3 UNIT
  COMMON/SPCIES/ISPEC $SPEC(6,3)
  COMMON/FVIB/FUPUP(0:10,0:20), FUPD(0:10,0:20), FUP0(0:10,0:20)
  1 FDOWN(0:10,0:20), FDOWN0(0:10,0:20), FDOWN0(0:10,0:20)
  2 FUP(0:10,0:20), FDOWN(0:10,0:20)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, WONLY, VUNIT(0:23)
  COMMON/PHYCCT/PLANCE, FOLITE, RGAS, SCEXPL, CLIGHT, PI, ISEVER
  COMMON/COUNT/MAXV, MAXJ(0:10), MAXUP, INPTS, IROLL
  $SUB=$PRSUB: FVIBL)
  WRITE(IOUT,701) ($SPEC(1,ISPEC), L=1,6)
  701 FORMAT(1X,6A6)
  900 WRITE(IOUT,*) Input the units to be used for v-->v+1
  & rate constants: <default: "e"/s>
  IF(INOV.EQ.1) CALL NOVICE(20)
  READ(INP,*) END=901) UNIT
  GO TO 902
  901 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 900
  ELSE
    CALL DFALTC(UNIT SE)
  ENDF
  902 WRITE(ISAVE,*) UNIT
  ... dummy call to SIU checks for valid unit
  CALL SIU(20,ISUP,UNIT)
  C Originally we prompted for k(0,1) but this turned out to be inconvenient
  C (the literature usually cites k(0,1) so now we prompt for k(0) and
  C immediately convert to k(0,1) so that the remainder of the code is not
  C affected. Users may wish to prompt for k(0,1)
  903 WRITE(IOUT,*) Input k for v=1 J=0 --> v=0 J=0 transition.
  & <default: 1.44D7>
  READ(INP,*) END=904) F1000
  GO TO 905
  904 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 903
  ELSE
    CALL DFALTD(F1000 1.44D7)
  ENDF
  905 WRITE(ISAVE,*) F1000
  CONST=1.0/(RGAS*TEMP)
  FUPD(0,0) = F1000*DEXP(EDEL(0,0,1,0)*CONST)
  IF (ISPEC.EQ.1) WRITE(I,*) Fup0: FUP0(0,0)
  CALL SIU(20,FUPD(0,0) UNIT)
  IF (WONLY.EQ.0) THEN
    906 WRITE(IOUT,*) Input k for v=1 J=2 --> v=0 J=2 transition:
    WRITE(IOUT,*) <default: 1.8D7>
    READ(INP,*) END=907) F1000
    GO TO 908
  907 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 906
  ELSE
    CALL DFALTD(F1000 1.8D7)
  ENDF
  908 WRITE(ISAVE,*) F1000
  FUPUP(0,0) = F1000*DEXP(EDEL(0,0,1,2)*CONST)
  CALL SIU(20,FUPUP(0,0) UNIT)
  909 WRITE(IOUT,*) Input k for v=1 J=0 --> v=0 J=2 transition:
  & <default: 1.2D7>
  READ(INP,*) END=910) K1000
  GO TO 911
  910 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 909
  ELSE
    CALL DFALTD(F1000 1.2D7)
  ENDF
  911 WRITE(ISAVE,*) K1000
  FUPD(0,0) = K1000*TEMP*EDEL(0,0,1,0)*CONST
  CALL SIU(20,FUPD(0,0) UNIT)
  C Must find FUPD(0,0) all J, and FUPD(0) all J, and FUPD(0) all J)
  C --this will be done using a modified FROUFP routine.
  CALL KROTI
  ENDF
  IF(ANKO1.EQ.1) THEN
    CALL KHALT
  ELSEIF (ANKO1.EQ.2) THEN
    CALL KHALT
  ELSE
    CALL ITERM(ANKO1, $SUB)
  ENDF
  RETURN
END

```

```

SUBROUTINE K01LT
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 ($)
LOGICAL*1 EXIST
COMMON/KVIB/KUPUP(0:10,0:20),KUPD(0:10,0:20),KUP0(0:10,0:20)
1 KDOWN(0:10,0:20),KDOWN0(0:10,0:20),KDOWN00(0:10,0:20)
2 K0UP(0:10,0:20),K0DOWN(0:10,0:20)
COMMON/IO/INP,IOUT,ISAVE,ISUP,IDEBUG,INDAT,INOV,VONLY,VUNIT(0:23)
COMMON/KVIB/KUPUP(0:10,0:20),KUPD(0:10,0:20),KUP0(0:10,0:20)
1 KDOWN(0:10,0:20),KDOWN0(0:10,0:20),KDOWN00(0:10,0:20)
2 K0UP(0:10,0:20),K0DOWN(0:10,0:20)
COMMON/ENERGY/ELEV(0:10,0:20),TEMP
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN, INPTS, ICOLL
SSUB=$PRSUB('K01LT ')
DO 100 V = 0, MAXV
DVI=DBLE(V+1)
DO 100 J = 0, MAXJ(V)
IF(EXIST(V,J,V+1,J-2)) KUPD(V,J)=DVI*KUPD(0,J)
IF(EXIST(V,J,V+1,J-1)) KUPD(V,J)=DVI*KUPD(0,J)
IF(EXIST(V,J,V+1,J-2)) K0UP(V,J)=DVI*K0UP(0,J)
IF(EXIST(V,J,V+1,J-2)) KUPUP(V,J)=DVI*KUPUP(0,J)
100 CONTINUE
RETURN
END
    
```

```

SUBROUTINE KROT2
C This subroutine mimics KROTUP, except that KROT2 determines the
C rate coefficients for rotational levels when there is
C a change of  $\pm 1$  in vibrational levels.
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 ($)
LOGICAL*1 EXIST
COMMON/KVIB/KUPUP(0:10,0:20),KUPD(0:10,0:20),KUP0(0:10,0:20)
1 KDOWN(0:10,0:20),KDOWN0(0:10,0:20),KDOWN00(0:10,0:20)
2 K0UP(0:10,0:20),K0DOWN(0:10,0:20)
COMMON/IO/INP,IOUT,ISAVE,ISUP,IDEBUG,INDAT,INOV,VONLY,VUNIT(0:23)
COMMON/PHYSCT/PLANCK,KBOLTZ,RGAS,SCEXP1,CLIGHT,PI,ISVEVER
COMMON/ENERGY/ELEV(0:10,0:20),TEMP
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN, INPTS, ICOLL
SSUB=$PRSUB('KROT2 ')
C k for v=0, J=2 --> v=1, J=2 (KUPUP(0,0)) transition is known.
C k for v=0, J=2 --> v=1, J=0 (KUPD(0,0)) transition is known.
C Now we will just build KUPUP(0.all J) and KUPD(0.all J) vectors.
WRITE(IOUT,*) 'Rotational Transition Rate Coefficients: KUPUP'
WRITE(IOUT,*) ' THE k(0,J-->1,J-2) constants are : '
WRITE(IOUT,*) ' J k(0,J-->1,J-2) '
C user can set value of constant (recip) for alternate scaling.
C or change function altogether.
RECIP=1.0
C Should be Rgas to make energy*recip unitless (dale/RT)
RECIP=RECIP/(RGAS*TEMP)
C pre-exponential factor needed to return consistent k(0,0-->1,2)
PREEXP = DEXP(-RECIP*EDEL(0,0,1,2))
PREEXP = PREEXP*KUPUP(0,0)
DO 100 J=0,MAXJ(0)-2
WITH the pre-exponential factor as defined, using (0,J,1,J+2)
will amount to taking the difference between the lengths
of the 2 transition vectors.
IF (EXIST(0,J,0,J-2))
4 KUPUP(0,J)=PREEXP*DEXP(RECIP*EDEL(0,J,1,J-2))
WRITE(IOUT,*) J,KUPUP(0,J)
100 CONTINUE
WRITE(IOUT,*) ' '
WRITE(IOUT,*) 'Rotational Transition Rate Coefficients:KUPD'
WRITE(IOUT,*) ' THE k(0,J-->1,J-2) constants are : '
WRITE(IOUT,*) ' J k(0,J-->1,J-2) '
C pre-exponential factor needed to return consistent k(0,2-->1,0)
PREEXP = DEXP(-RECIP*EDEL(0,2,1,0))
PREEXP = PREEXP*KUPD(0,2)
DO 100 J=2,MAXJ(0)
IF (EXIST(0,J,1,J-2))
4 KUPD(0,J)=PREEXP*DEXP(RECIP*EDEL(0,J,1,J-2))
WRITE(IOUT,*) J,KUPD(0,J)
200 CONTINUE
WRITE(IOUT,*) ' '
WRITE(IOUT,*) 'Rotational Transition Rate Coefficients:KUP0'
WRITE(IOUT,*) ' THE k(0,J-->1,J) constants are : '
WRITE(IOUT,*) ' J k(0,J-->1,J) '
PREEXP=DEXP(-RECIP*EDEL(0,0,1,0))
PREEXP=PREEXP*KUP0(0,0)
DO 300 J=0,MAXJ(0)
IF (EXIST(0,J,1,J)) KUP0(0,J)=PREEXP*DEXP(RECIP*EDEL(0,J,1,J))
WRITE(IOUT,*) J,KUP0(0,J)
300 CONTINUE
WRITE(IOUT,*) ' '
RETURN
END
    
```

```

SUBROUTINE PIKKG1(ANK01)
C Selection of method to calculate rate coefficients for adjacent
C vibrational level transitions.
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 ($)
COMMON/IO/INP,IOUT,ISAVE,ISUP,IDEBUG,INDAT,INOV,VONLY,VUNIT(0:23)
COMMON/KVIB/KUPUP(0:10,0:20),KUPD(0:10,0:20),KUP0(0:10,0:20)
1 KDOWN(0:10,0:20),KDOWN0(0:10,0:20),KDOWN00(0:10,0:20)
2 K0UP(0:10,0:20),K0DOWN(0:10,0:20)
SSUB=$PRSUB('PIKKG1')
101 WRITE(IOUT,*) 'Rate constants between adjacent v-levels are:'
WRITE(IOUT,*) ' 1) Calculated by Landau-Teller scaling'
WRITE(IOUT,*) ' 2) Calculated by alternate scaling'
WRITE(IOUT,*) ' 3) Input from data file'
READ(INP,*) END=900) ANK01
GO TO 901
IF (IOUT.EQ.ISUP) THEN
INP=5
IOUT=5
GO TO 101
ELSE
CALL DFALTI(ANK01,1)
ENDIF
901 WRITE(ISAVE,*) ANK01
IF (ANK01.EQ.1 OR ANK01.EQ.2) THEN
CALL K0ISCL(ANK01)
ELSEIF (ANK01.EQ.3) THEN
CALL READIN(21,K0UP)
ELSE
WRITE(IOUT,*) 'Error in selection of rate constant'
WRITE(IOUT,*) ' calculation method.' ANK01
4 WRITE(IOUT,*) 'Answer should be 1, 2 or 3.'
IFLAG = IFLAG + 1
IF (IFLAG.LE.1) GO TO 101
CALL ITERM(ANK01,SSUB)
ENDIF
IF (IDEBUG.EQ.1) CALL PRINTVJ(23,K0UP)
RETURN
END
    
```

```

SUBROUTINE K0IALT
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 ($)
LOGICAL*1 EXIST
COMMON/KVIB/KUPUP(0:10,0:20),KUPD(0:10,0:20),KUP0(0:10,0:20)
1 KDOWN(0:10,0:20),KDOWN0(0:10,0:20),KDOWN00(0:10,0:20)
2 K0UP(0:10,0:20),K0DOWN(0:10,0:20)
COMMON/PHYSCT/PLANCK,KBOLTZ,RGAS,SCEXP1,CLIGHT,PI,ISVEVER
COMMON/ENERGY/ELEV(0:10,0:20),TEMP
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN, INPTS, ICOLL
SSUB=$PRSUB('K0IALT')
CONST=1.0D0
PPEEXP=1.0D0
...user can change values of exponential and
pre-exponential constants.
CONST=CONST/(PLANCK*TEMP)
DO 100 V = 0, MAXV
DO 100 J = 0, MAXJ(V)
IF (EXIST(V,J,V+1,J-2))
4 KUPD(V,J)=PPEEXP*DEXP(CONST*EDEL(V,J,V+1,J-2))*KUPD(0,J)
IF (EXIST(V,J,V+1,J-1))
4 KUPD(V,J)=PPEEXP*DEXP(CONST*EDEL(V,J,V+1,J-1))*KUPD(0,J)
IF (EXIST(V,J,V+1,J-2))
4 K0UP(V,J)=PPEEXP*DEXP(CONST*EDEL(V,J,V+1,J-2))*K0UP(0,J)
IF (EXIST(V,J,V+1,J-2))
4 KUPUP(V,J)=PPEEXP*DEXP(CONST*EDEL(V,J,V+1,J-2))*KUPUP(0,J)
100 CONTINUE
RETURN
END
    
```

```

SUBROUTINE CHECKR
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I,J,L-M,V)
  LOGICAL CHARACTER*6 (S)
  LOGICAL*1 EXIST
  COMMON/RVIB/PUPUP(0:10,0:20),KUPD(0:10,0:20),FUPD(0:10,0:20)
  1 FDOWN(0:10,0:20),FEDOWN(0:10,0:20),FDOWN(0:10,0:20)
  2 KUP(0:10,0:20),KEDOWN(0:10,0:20)
  COMMON/IO/INP,IOUT,ISAVE,ISUP,ISEREG,INAT,INOV,VONLY,VUNIT(0:21)
  COMMON/ENERGY/ELEV(0:10,0:20),TEMP
  COMMON/COUNT/MAKV,MAKJ(0:10),MAKUN,INPTS,ICOLL
  SSUB=IPRPSUB/CHECKR/
  SBLANK =
  C This routine checks whether the arrays used by MICREV are not 0 in
  C the positions 0 0 -> MAKV, MAKJ(MAKV)
  C If 0 then a warning is generated and the V, J position is flagged.
  DO 100 V = 0, MAKV
    DO 100 J = 0, MAKJ(V)
      IF(EXIST(V,J,V+1,J+2)) THEN
        C N.B. --if transition does not exist, then we do not want to
        C look up the rate coefficient in the array since the index
        C might be out of bounds.
        IF (KUPUP(V,J).EQ.0.D0) WRITE(IOUT,706) V,J
      ENDF
      IF(EXIST(V,J,V,J+2)) THEN
        IF (KUP(V,J).EQ.0.D0) WRITE(IOUT,707) V,J
      ENDF
      IF(EXIST(V,J,V-1,J)) THEN
        IF (KUPD(V,J).EQ.0.D0) WRITE(IOUT,708) V,J
      ENDF
      IF(EXIST(V,J,V-1,J-2)) THEN
        IF (KUPD(V,J).EQ.0.D0) WRITE(IOUT,709) V,J
      ENDF
    100 CONTINUE
  706 FORMAT(' Array Kup.up has a 0 in position ',2I4)
  707 FORMAT(' Array K 0.up has a 0 in position ',2I4)
  708 FORMAT(' Array Kup.0 has a 0 in position ',2I4)
  709 FORMAT(' Array Kup.down has a 0 in position ',2I4)
  C Values for the Kup,x (x=up,down,0) and K0,up are tabulated.
  WRITE(IOUT,705)
  705 FORMAT(2X,' J',2X,' J',2X,' K(V,J--sv-1,J+2)',2X,' F(V,J--sv-1,
  & J+2)',2X,' K(V,J--sv-1,J)',2X,' K(V,J--sv-1,J-2)')
  DO 200 V = 0, MAKV
    DO 300 J = 0, MAKJ(V)
      IF(EXIST(V,J,V-1,J-2)) THEN
        WRITE(IOUT,711) V,J,KUPUP(V,J)
      ELSE
        WRITE(IOUT,712) V,J,SBLANK
      ENDF
      IF(EXIST(V,J,V,J-2)) THEN
        WRITE(IOUT,713) KUP(V,J)
      ELSE
        WRITE(IOUT,714) SBLANK
      ENDF
      IF(EXIST(V,J,V-1,J)) THEN
        WRITE(IOUT,715) KUPD(V,J)
      ELSE
        WRITE(IOUT,716) SBLANK
      ENDF
      IF(EXIST(V,J,V-1,J-2)) THEN
        WRITE(IOUT,717) FUPD(V,J)
      ELSE
        WRITE(IOUT,718) SBLANK
      ENDF
    300 CONTINUE
  200 CONTINUE
  711 FORMAT(2I3,D12.4,3X)
  712 FORMAT(2I3,3X,A6)
  713 FORMAT(2X,D12.4)
  714 FORMAT(2X,3X,A6)
  715 FORMAT(3X,D12.4)
  716 FORMAT(3X,3X,A6)
  717 FORMAT(5X,D12.4)
  718 FORMAT(5X,3X,A6)
  RETURN
  END

```

```

SUBROUTINE MICREV
  C The J--J-2 transition rate coefficients can be determined
  C by microscopic reversibility (along with v--sv-1 rates).
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  LOGICAL CHARACTER*6 (S)
  LOGICAL*1 EXIST
  COMMON/RVIB/PUPUP(0:10,0:20),FUPD(0:10,0:20),KUPD(0:10,0:20)
  1 FDOWN(0:10,0:20),FEDOWN(0:10,0:20),FDOWN(0:10,0:20)
  2 KUP(0:10,0:20),KEDOWN(0:10,0:20)
  COMMON/IO/INP,IOUT,ISAVE,ISUP,ISEREG,INAT,INOV,VONLY,VUNIT(0:21)
  COMMON/PHYSCT/PLANCE,FROLTE,PGAS,SCEKPI,CLIGHT,PI,IAEVER
  COMMON/ENERGY/ELEV(0:10,0:20),TEMP
  COMMON/COUNT/MAKV,MAKJ(0:10),MAKUN,INPTS,ICOLL
  SSUB=IPRPSUB/MICREV/
  SBLANK =
  C This constant should be negative, so that kdwn=>kup
  CONST=-1.0/(PGAS*TEMP)
  WRITE(IOUT,' ***** F DOWN (USING MICROSCOPIC REVERSIB
  & ILITY) *****')
  C Values for the Fdown,x (x=up,down,0) and K0,down are tabulated.
  WRITE(IOUT,705)
  705 FORMAT(2X,' V',2X,' J',2X,' K(V,J--sv-1,J-2)',2X,' K(V,J--sv-1,
  & J-2)',2X,' K(V,J--sv-1,J)',2X,' K(V,J--sv-1,J+2)')
  DO 200 V = 0, MAKV
    DO 300 J = 0, MAKJ(V)
      IF(EXIST(V,J,V-1,J-2)) THEN
        FDOWN(V,J)=KUPUP(V-1,J-2)*DEXP(EDEL(V,J,V-1,J-2)*CONST)
        WRITE(IOUT,711) V,J,FDOWN(V,J)
      ELSE
        WRITE(IOUT,712) V,J,SBLANK
      ENDF
      IF(EXIST(V,J,V,J-2)) THEN
        FDOWN(V,J)=KUP(V,J-2)*DEXP(EDEL(V,J,V,J-2)*CONST)
        WRITE(IOUT,713) FDOWN(V,J)
      ELSE
        WRITE(IOUT,714) SBLANK
      ENDF
      IF(EXIST(V,J,V-1,J)) THEN
        FDOWN(V,J)=KUPD(V-1,J)*DEXP(EDEL(V,J,V-1,J)*CONST)
        WRITE(IOUT,715) FDOWN(V,J)
      ELSE
        WRITE(IOUT,716) SBLANK
      ENDF
      IF(EXIST(V,J,V-1,J-2)) THEN
        FDOWN(V,J)=KUPD(V-1,J-2)*DEXP(EDEL(V,J,V-1,J-2)*CONST)
        WRITE(IOUT,717) FDOWN(V,J)
      ELSE
        WRITE(IOUT,718) SBLANK
      ENDF
    300 CONTINUE
  200 CONTINUE
  711 FORMAT(2I3,D12.4,3X)
  712 FORMAT(2I3,3X,A6)
  713 FORMAT(2X,D12.4)
  714 FORMAT(2X,3X,A6)
  715 FORMAT(3X,D12.4)
  716 FORMAT(3X,3X,A6)
  717 FORMAT(5X,D12.4)
  718 FORMAT(5X,3X,A6)
  RETURN
  END

```

```

SUBROUTINE PIPKRX(ANMXR)
C Scaling can be done as a function of the energy of the rovibrational
C state in question or directly as a function of the vibrational and
C rotational quantum numbers involved.
C For option 1) we chose scaling
C as a function of energy since this broad-brush approach seems to be
C more appropriate for global scaling. Thus
C in BMSCL the v,J levels are viewed as unique energy levels.
C The microscopic rate constants Kbin(v,J) are then mapped as
C Kbin(E). Choosing option 1) on this menu therefore
C implies that vibrational and rotational excitation can be
C "pooled" in a system where only their relative energies are
C seen as significant.
C For example, Ebin(v,J) happens to be accidentally
C degenerate with Ebin(v',J'), then their Kbin's will be equal.
C The energy-scaled approach has another characteristic
C that harmonic and anharmonic oscillators will yield different
C scaling since the Ebin changes for the anharmonic oscillator.
C For options 2) and 4)
C where a more detailed (and therefore more closely tied to the
C quantum number involved) description of the system is given
C scaling as a function of quantum number seems more appropriate.
C It could alternatively be argued that since vibrational motion
C is fundamentally different from rotational motion the
C function Kbin should be fitted separately to v, and then
C separately to the J values. For example,
C one could intuitively picture a system whose probability of
C reacting increases as its vibrations grow stronger,
C although the rotation of the system has little or no effect
C on whether it will react. Such a system should then be
C modelled by selecting options 2) or 4) from the menu.
C This approach 2) also provides a way of bypassing the
C energy-scaled approach of 1) while still achieving the global
C scaling provided for in 1).
C IMPLICIT DOUBLE PRECISION (B-H,F,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
C CHARACTER*2 UNIT
C COMMON/IO/INP,ICUT,ISAVE,ISUP,IDEBUG,INDAT,INOV,VOONLY,VUNIT(0:10)
C COMMON/MASTR/KBIN(0:10,0:10)
C SSUB=SPRSUB('PIPKRX')
C IFLAG = 0
103 WRITE(IOUT,*) 'The microscopic Kbin rate constants for
C the reaction A -> BC ->> pdt. are:'
C IF (VOONLY.EQ.0) THEN
C WRITE(IOUT,*) '1) Scaled by v,J energy from Fxn(v,J).
C WRITE(IOUT,*) '2) Read from a file of Fxn(v,J)-v,J pairs.
C (Unit 3.)
C WRITE(IOUT,*) '3) Scaled from datafile of Fxn(v,J)-v,J pairs.
C (Unit 3.)
C WRITE(IOUT,*) '4) Scaled separately for the v-levels
C and subsequently for the J-levels.
C This last option has been added, to account for cases where,
C for example, the microscopic rate coefficient increases
C monotonically w.r. to v-level, but has a distribution
C peaked around an optimum J-level.
C ELSE
C WRITE(IOUT,*) '1) Scaled by v-level energy from Fxn(v,J).
C WRITE(IOUT,*) '2) Read from a file of Fxn(v,J)-v,J pairs.
C (Unit 3.)
C ENDIF
C READ(INP,'END=900) ANMXR
C GO TO 901
900 IF (ICUT.EQ.ISUP) THEN
C INP=5
C IOUT=6
C GO TO 103
C ELSE
C CALL DFALTI(ANMXR,1)
C ENDIF
901 WRITE(ISAVE,*) ANMXR
C IF (ANMXR.EQ.1) THEN
C CALL BMSCL
C ELSEIF (ANMXR.EQ.2 .AND. VOONLY.EQ.0) THEN
C CALL READVJ(25,KBIN)
C ELSEIF (ANMXR.EQ.2 .AND. VOONLY.EQ.1) THEN
C CALL READIN(24,KBIN)
C ELSEIF (ANMXR.EQ.3 .AND. VOONLY.EQ.0) THEN
C CALL READIN(24,KBIN)
C CALL BMSCLJ(UNIT)
C In BMSCLJ, for a given v-level, the Kbin(J) can be made to fit a
C predefined function (e.g. gaussian values w.r. to J-value).
C ELSEIF (ANMXR.EQ.4 .AND. VOONLY.EQ.0) THEN
C CALL BMSCLV(UNIT)
C BMSCLV is analogous to BMSCLJ, with J set at 0, the Kbin(v)
C is made to fit a predefined function (e.g. exponential increase
C over v.)
C CALL BMSCLV(UNIT)
C ELSE
C WRITE(IOUT,*) 'Error in Fxn(v) specification - ANMXR
C IF (VOONLY.EQ.0) WRITE(IOUT,*) 'Input : 1, 2, 3 or 4.'
C IF (VOONLY.EQ.1) WRITE(IOUT,*) 'Input 1 or 3.'
C IFLAG = IFLAG + 1
C IF (IFLAG.LE.1) GOTO 103
C CALL ITERM(ANMXR,SSUB)
C ENDIF
C IF (IDEBUG.EQ.1) CALL PRINTVJ(24,KBIN)
C RETURN
C END

```

```

SUBROUTINE BMSCL
C This subroutine allows the user to select how Kbin(v,J) will be
C represented as a function of E.
C In each form of scaling 3 guidelines are followed:
C 1) the energy gap will be the independent variable (occurring
C on the RHS).
C 2) the equation must return the FWHM(v,J) value for the
C v,J=0 level.
C 3) the dimensions of the rate coefficient must be preserved.
C IMPLICIT DOUBLE PRECISION (B-H,F,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
C CHARACTER*2 UNIT
C COMMON/IO/INP,ICUT,ISAVE,ISUP,IDEBUG,INDAT,INOV,VOONLY,VUNIT(0:10)
C COMMON/CONST/MAV,MAV(0:10),MANIN,INPTS,ICOLL
C COMMON/ENERGY/ELEV(0:10,0:10),TEMP
C DIMENSION SCENP(10)
C COMMON/MASTR/VJBIN(0:10,0:10)
C SSUB=SPRSUB('BMSCL')
C CALL FWHM0
C FWHM=FWHM0(0)
C IFLAG = 0
101 WRITE(IOUT,*) 'Select the form of the bimolecular rate
C vs. E curve according to number:
C but a call to notice here to write a message about defaults
C and time**(-1) for intercept, so that the rate coefficient
C will have overall units of time**(-1)
C WRITE(IOUT,*) '1) Linear
C WRITE(IOUT,*) '2) Simple exponential
C WRITE(IOUT,*) '3) Cusp (two piecewise-continuous
C exponentials)
C WRITE(IOUT,*) '4) Sum of exponentials
C READ(INP,'END=903) ANSCL
C GOTO 904
903 IF (ICUT.EQ.ISUP) THEN
C INP = 5
C IOUT = 6
C GOTO 102
C ELSE
C CALL DFALTI(ANSCL,1)
C ENDIF
904 WRITE(ISAVE,*) ANSCL
C IF (ANSCL.EQ.1) THEN
C CALL SCLN1(FWMA)
C ELSEIF (ANSCL.EQ.2) THEN
C CALL SCLN2(FWMA)
C ELSEIF (ANSCL.EQ.3 .OR. ANSCL.EQ.4) THEN
C CALL SCLN3(FWMA,ANSCL)
C ELSE
C WRITE(IOUT,*) 'Error in potential curve specification ANSCL
C WRITE(IOUT,*) 'Specify 1, 2, 3 or 4.'
C IFLAG = IFLAG + 1
C IF (IFLAG.LE.1) GOTO 102
C CALL ITERM(ANSCL,SSUB)
C ENDIF
C WRITE(IOUT,*) '#### K MICROSCOPIC BIMOLECULAR ####
C WRITE(IOUT,*) ' THE KIV-->RXN) CONSTANTS ARE :
C WRITE(IOUT,*) ' V KIV-->RXN)
C DO 10 V=0,MAV
C WRITE(IOUT,*) V,VJBIN(V,0)
10 CONTINUE
C RETURN
C END

```

```

SUBROUTINE FWHM0
C IMPLICIT DOUBLE PRECISION (B-H,F,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
C CHARACTER*2 UNIT
C COMMON/IO/INP,ICUT,ISAVE,ISUP,IDEBUG,INDAT,INOV,VOONLY,VUNIT(0:10)
C COMMON/CONST/MAV,MAV(0:10),MANIN,INPTS,ICOLL
C COMMON/ENERGY/ELEV(0:10,0:10),TEMP
C DIMENSION SCENP(10)
C COMMON/MASTR/VJBIN(0:10,0:10)
C SSUB=SPRSUB('FWHM0')
900 WRITE(IOUT,*) 'Input the microscopic rate constant (and its
C units) for the v=0,J=0 level
C WRITE(IOUT,*) 'Default: 3.0E-2 (sec^-1)
C IF (INOV.EQ.1) CALL NOTICE(10)
C READ(INP,'END=901) FWHMA,UNIT
C GOTO 902
901 IF (ICUT.EQ.ISUP) THEN
C INP = 5
C IOUT = 6
C GOTO 900
C ELSE
C CALL DFALTI(FWMA,3.0E-2)
C CALL DFALTI(UNIT,'SE')
C ENDIF
902 WRITE(ISAVE,*) FWHMA,UNIT
C CALL SIU(10,FWMA,UNIT)
C FWHM(0,0) = FWHMA
C RETURN
C END

```

```

SUBROUTINE SCLM1(FBIMA)
C SCLM1 is one of three scaling routines called by SIMSCL.
C It scales linearly the elements of the
C FVUBIM(V) vector from FVUBIM(0) according to the size of
C the energy gap between levels.
C IMPLICIT DOUBLE PRECISION (B-H F-N-U-W-Z)
C IMPLICIT INTEGER (A I-J L-M V)
C IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP IOUT ISAVE ISUP ISEBEN INEAT INOV VONLY VUNIT(10)
COMMON/COUNT/MAXV MAXJ(10) MAXUN INPTS ISOLL
COMMON/ENERGY/ELEV(0:10 0:10) TEMP
COMMON/MASTR/FVUBIM(0:10 0:10)
ISSUB=IPRSUB+SCLM1
905 WRITE(IOUT,*) 'By what factor does Fvbm(1.0) increase
      & relative to Fvbm(0)?'
      & WRITE(IOUT,*) 'This will not be the actual slope since
      & a further energy
      & scaling will be applied: <Default: 1.0>'
      & READ(INP,*) END=906: SLOPE
      & GOTO 907
906 IF(IOUT.EQ.ISUP) THEN
      & INP = 5
      & IOUT = 6
      & GOTO 905
      & ELSE
      & CALL DFAULT(SLOPE 1.000)
      & ENDDIF
907 WRITE(ISAVE,*) SLOPE
      & DO 100 J = 0, MAXV
      & DO 100 I = 0, MAXJ(I)
      & FVUBIM(V(I)) = SLOPE*EDEL(V(I)) + FBIMA/EDEL(I) 0 0 0:FBIMA
100 CONTINUE
      & RETURN
      & END

```

```

SUBROUTINE SCLM2(FBIMA)
C SCLM2 is one of three scaling routines called by SIMSCL.
C It scales exponentially the elements of the
C FVUBIM(V) vector from FVUBIM(0).
C IMPLICIT DOUBLE PRECISION (B-H F-N-U-W-Z)
C IMPLICIT INTEGER (A I-J L-M V)
C IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP IOUT ISAVE ISUP ISEBEN INEAT INOV VONLY VUNIT(10)
COMMON/COUNT/MAXV MAXJ(10) MAXUN INPTS ISOLL
COMMON/ENERGY/ELEV(0:10 0:10) TEMP
COMMON/MASTR/FVUBIM(0:10 0:10)
ISSUB = SCLM2
908 WRITE(IOUT,*) 'What factor should be applied to the
      & argument of the exponential?
      & The sign of the constant is positive so that reactivity
      & will increase as v level increases.
      & READ(INP,*) END=909: CONST1
      & GOTO 910
909 IF(IOUT.EQ.ISUP) THEN
      & INP = 5
      & IOUT = 6
      & GOTO 908
      & ELSE
      & CALL DFAULT(CONST1 1.00-3)
      & ENDDIF
910 WRITE(ISAVE,*) CONST1
      & DO 100 V = 0, MAXV
      & DO 100 J = 0, MAXJ(V)
      & FVUBIM(V(J)) = FBIMA*DEXP(EDEL(V(J)) 0/ELEV(0) 0) *CONST1
100 CONTINUE
      & RETURN
      & END

```

```

SUBROUTINE SCLM3(FBIMA ANSCL)
C SCLM3 is one of three scaling routines called by SIMSCL.
C It scales exponentially (using more than one exponential) the
C elements of the FVUBIM(V) vector from FVUBIM(0).
C Note that choice 4 is not a set of piecewise-continuous exponentials
C but a true sum.
C IMPLICIT DOUBLE PRECISION (B-H F-N-U-W-Z)
C IMPLICIT INTEGER (A I-J L-M V)
C IMPLICIT CHARACTER*6 (S)
C CHARACTER*3 UNIT
COMMON/IO/INP IOUT ISAVE ISUP ISEBEN INEAT INOV VONLY VUNIT(10)
COMMON/COUNT/MAXV MAXJ(10) MAXUN INPTS ISOLL
COMMON/ENERGY/ELEV(0:10 0:10) TEMP
COMMON/MASTR/FVUBIM(0:10 0:10)
COMMON/SCXFP/SCXFP(10)
ISSUB=IPRSUB+SCLM3
C ...for option 4
IF(ANSCL.EQ.4) THEN
913 WRITE(IOUT,*) 'How many exponentials will be
      & summed? <default: 2>'
      & READ(INP,*) END=914: NEXP
      & GOTO 915
914 IF(IOUT.EQ.ISUP) THEN
      & INP = 5
      & IOUT = 6
      & GOTO 913
      & ELSE
      & CALL DFAULT(N 2)
      & ENDDIF
915 WRITE(ISAVE,*) NEXP
      & ENDDIF
C For a cusp the first value should be positive and the second negative.
C This is the sign that the calculation expects.
916 WRITE(IOUT,*) 'Input the constants for the exponentials
      & (N.B. for a cusp the first constant is positive
      & and the second is negative).
      & READ(INP,*) END=916: (SCXFP(1) 1-1 NEXP) UNIT
      & GOTO 917
917 IF(IOUT.EQ.ISUP) THEN
      & INP = 5
      & IOUT = 6
      & GOTO 916
      & ELSE
      & DO 100 I=1,NEXP
      & CALL DFAULT(SCXFP(I) 10.000**I-1)
100 CONTINUE
C No special significance to this formula: just wanted constants
C to be 10^1, 1/10 etc.
      & ENDDIF
918 WRITE(ISAVE,*) (SCXFP(I) 1-1 NEXP)
C Since constants are in inverse energy units the reciprocal units will
C be found using DUMMY.
      & DUMMY=1.0
      & CALL SUB(10 DUMMY UNIT)
      & DO 100 I = 1,NEXP
      & SCXFP(I) = SCXFP(I)/DUMMY
100 CONTINUE
IF(ANSCL.EQ.3) THEN
C The first exponential may be discontinuous from the second
C exponential at E = Ecrit so they are forced to be piecewise
C continuous by a simple ratio applied to the second exponential.
C This seems to be the physically reasonable thing to do.
917 WRITE(IOUT,*) 'What is the energy Ecrit (and its units) at
      & which the maximum of the cusp occurs?
      & IF(INOV.EQ.1) CALL NOTICE(10)
      & READ(INP,*) END=918: CRITE UNIT
      & GOTO 919
918 IF(IOUT.EQ.ISUP) THEN
      & INP = 5
      & IOUT = 6
      & GOTO 917
      & ELSE
      & CALL DFAULT(CRITE 1.000)
      & CALL DFAULT(UNIT '0')
      & ENDDIF
919 WRITE(ISAVE,*) CRITE UNIT
      & CALL SUB(10 CRITE UNIT)
C crite = edlv(0) is analogous to edlv(1) 0.01
      & V1 = FBIMA*DEXP(CRITE - ELEV(0)/ELEV(0) *SCXFP(1))
      & V2 = FBIMA*DEXP(CRITE - ELEV(0)/ELEV(0) *SCXFP(2))
      & FBIMA = V1/V2 + FBIMA
      & DO 100 J=0,MAXV
      & DO 100 I=0,MAXJ(I)
      & FVUBIM(V(I)) = FBIMA*DEXP(EDEL(V(I)) 0/ELEV(0) 0) *SCXFP(1)
      & FVUBIM(V(I)) = FBIMA*DEXP(EDEL(V(I)) 0/ELEV(0) 0) *SCXFP(2)
100 CONTINUE
      & ENDDIF
      & ENDDIF
      & DO 100 V = 1, MAXV
      & FVUBIM(V) = SCXFP
      & DO 100 I = 1, NEXP
      & FVUBIM(V(I)) = FVUBIM(V(I))
      & FBIMA*DEXP(EDEL(V(I)) 0/ELEV(0) 0) *SCXFP(I)
100 CONTINUE
      & CONTINUE
      & ENDDIF
      & RETURN
      & END

```

```

SUBROUTINE BMSCLJ(UNIT)
C The scaling of the microscopic bimolecular reaction rate
C constants could have a variety of J-dependencies.
C (i) The simplest model assumes no predictable
C J-dependence---but this could
C mean one of two things: that krxn varies randomly
C according to J or that it is a constant value over
C all values of J.
C The random number generated will be bounded by 0 and 2
C so that the average is 1) and
C for a given v,J level, will multiply the value for kvjbin(v, J)
C to produce a value for kvjbin(v, J).
C (ii) krxn might increase monotonically over J so that the
C reaction coefficient becomes larger as the rotational
C excitation increases. (This could also occur as a function
C simply of the energy.)
C (iii) krxn might "decrease" monotonically over J although this
C does not seem entirely defensible (i.e., a small amount of
C rotation would enable the species to align in the most
C reactive position--see also Bethynarthy's review 1983).
C N.B. options ii) and iii) can be handled together since the same
C function with different input parameters can give the desired
C behavior (e.g. for a linear function, negative or positive slope).
C (iv) Alternatively, krxn might show a J-dependence which is
C peaked (strongly or broadly) about a non-zero J.
C In a very flexible program, the peak-J should be able to
C be varied according to v-level.
C
C (ii) consider the use of UNIT in this section--should it be parsed?
C
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C CHARACTER*3 UNIT
C COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INEAT, INOV, VONLY, VUNIT(0:23)
C COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
C COMMON/MASTR/KVJBIN(0:10, 0:20)
C SSUB=SPSUB('BMSCLJ')
C IFLAG = 0
102 WRITE(IOUT,*) 'Select the scaling required for Frxn:'
C WRITE(IOUT,*) '1) constant with respect to J'
C WRITE(IOUT,*) '2) varies randomly between
C 0 and 2 times the Krxn(v,0) value.
C 3) increases or decreases
C 4) monotonically over the energy of the J-levels'
C WRITE(IOUT,*) '4) Gaussian peaked at a given J value'
C READ(INP,*,END=903) ANSCL
C GOTO 904
903 IF(IOUT.EQ.ISUP) THEN
C INP = 5
C IOUT = 6
C GOTO 103
ELSE
C CALL DFALTD(ANSCL,1)
ENDIF
904 WRITE(ISAVE,*) ANSCL
C IF (ANSCL.EQ.1) THEN
C CALL SCLJ1
ELSEIF (ANSCL.EQ.2) THEN
C CALL SCLJ2
ELSEIF (ANSCL.EQ.3) THEN
C CALL SCLJ3
ELSEIF (ANSCL.EQ.4) THEN
C CALL SCLJ4
ELSE
C WRITE(IOUT,*) 'Error in potential curve specification .ANSCL'
C WRITE(IOUT,*) 'Specify 1, 2, 3 or 4.'
C IFLAG = IFLAG + 1
C IF (IFLAG.LE.1) GOTO 103
C CALL ITERM(ANSCL, SSUB)
ENDIF
C WRITE(IOUT,*) '***** F MICROSCOPIC, BIMOLECULAR *****'
C WRITE(IOUT,*) 'THE K(v,J-->RXN) CONSTRAINTS ARE:'
C WRITE(IOUT,*) 'V J F(v,J-->RXN)'
C DO 10 V=0, MAXV
C DO 10 J=0, MAXJ(V)
C WRITE(IOUT,*) V, J, KVJBIN(V, J)
10 CONTINUE
C IF (IDEBUG.EQ.1) CALL PRINTV(20, KVJBIN)
C RETURN
C END

SUBROUTINE SCLJ1
C SCLJ1 is one of four scaling routines called by BMSCLJ.
C It maps KVJBIN(V,0) values into all remaining KVJBIN(V,J) positions
C for a given v.
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C CHARACTER*3 UNIT
C COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INEAT, INOV, VONLY, VUNIT(0:23)
C COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
C COMMON/MASTR/KVJBIN(0:10, 0:20)
C SSUB=SPSUB('SCLJ1')
C DO 200 V = 0, MAXV
C DO 200 J = 1, MAXJ(V)
C KVJBIN(V, J) = KVJBIN(V, 0)
200 CONTINUE
C RETURN
C END

SUBROUTINE SCLJ2
C SCLJ2 is one of four scaling routines called by BMSCLJ.
C It generates random values for KVJBIN(V, J) centered on KVJBIN(V, 0)
C bounded by 0 and 2*KVJBIN(V, 0) for a given v.
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C CHARACTER*3 UNIT
C COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INEAT, INOV, VONLY, VUNIT(0:23)
C COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
C COMMON/MASTR/KVJBIN(0:10, 0:20)
C SSUB = SCLJ1
908 WRITE(IOUT,*) 'Input seed for random number generator'
C (v=0) number between 1.0E0 and 1.0E10.
C READ(INP,*,END=909) DSEED
C GOTO 910
909 IF(IOUT.EQ.ISUP) THEN
C INP = 5
C IOUT = 6
C GOTO 908
ELSE
C CALL DFALTD/DSEED 14285750.
ENDIF
910 WRITE(ISAVE,*) DSEED
C IF (DSEED.LT.1.0D0 .OR. DSEED.GT.1.147485647E10) THEN
C WRITE(IOUT,*) 'Seed is out of range.'
C IFLAG = IFLAG + 1
C IF (IFLAG.LE.1) GOTO 908
C CALL ITERM(DSEED, SSUB)
ENDIF
C We generate all the random numbers we need with a single call to the
C IMSL routine, DGOBS. Note that the numbers 11 and 101 are dependent
C on the size of the vibrational and rotational arrays.
C NR = 11*101
C CALL DGOBS(DSEED, NR, R)
C DO 300 V = 0, MAXV
C DO 300 J = 1, MAXJ(V)
C KVJBIN(V, J) = KVJBIN(V, 0) + 2.0D0*R(V*101, J-1)
300 CONTINUE
C RETURN
C END

SUBROUTINE SCLJ3
C SCLJ3 is one of four scaling routines called by BMSCLJ.
C It scales linearly by the energy gap the elements
C of the KVJBIN(V, J) array from KVJBIN(V, 0).
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C CHARACTER*3 UNIT
C COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INEAT, INOV, VONLY, VUNIT(0:23)
C COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
C COMMON/ENERGY/ELEV(0:10, 0:20), TEMP
C COMMON/MASTR/KVJBIN(0:10, 0:20)
C SSUB=SPSUB('SCLJ3')
905 WRITE(IOUT,*) 'Enter the ratio krxn(v=0, j=1)/krxn(v=0, j=0):'
C READ(INP,*,END=906) SLOPE
C GOTO 907
906 IF(IOUT.EQ.ISUP) THEN
C INP = 5
C IOUT = 6
C GOTO 905
ELSE
C CALL DFALTD(SLOPE, 1.0D0)
ENDIF
907 WRITE(ISAVE,*) SLOPE
C DO 200 V = 0, MAXV
C DO 200 J = 1, MAXJ(V)
C The unitless approach is easy to implement and values
C on the order of 1.0 seem physically reasonable.
C Note for scaling subroutines--the energy values should not be
C based on ELEV(0,0)=0 because numerous zero-divide errors can
C occur--especially wherever scaling is done (put a check for
C this in REACH.
C KVJBIN(V, J) = SLOPE*EDEL(V, J, 0, 0) + KVJBIN(V, 0) / EDEL(1, 0, 0, 0)
200 CONTINUE
C RETURN
C END

```

```

SUBROUTINE SCLJ4
C SCLJ4 is one of four scaling routines called by EMSCLV.
C It scales using a gaussian centered at a given J level.
C IMPLICIT DOUBLE PRECISION (B-H,K-N,O,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP,IOU,ISAVE,ISUP,IEBEG,INDET,INOV,VONLY,VUNIT(0:10)
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN,INPTS,ICOLL
COMMON/ENERGY/ELEV(0:10),G(0:10),TEMP
COMMON/MASTR/KVJBM(0:10),G(0:10)
SSUB=SPRSUB/SCLJ4
C Unlike the other J-scaling subroutines called by EMSCLV
C this routine will have to prompt for input explicitly
C at each of the allowed v-levels.
C
C DO 100 V=0,MAXV
C WRITE(OUT,*) '*** Scaling for J-values at the v level =',V
C WRITE(OUT,*) 'What is the J-value at which the k-bin is'
C 'at its peak? <default: Jpeak = Jmax/2>'
C 'Please input a decimal number: '
C READ(INP,*) END=902) JPEAK
C GOTO 903
C IF(OUT.EQ.ISUP) THEN
C INP = 5
C IOU = 6
C GOTO 901
C ELSE
C CALL DFALTD(PEAK,DBLE(MAXJ(V)/2))
C ENDF
C WRITE(ISAVE,*) JPEAK
C
C How wide is the Gaussian? Since chemists are already
C quite familiar with the idea of "full width at half maximum"
C the question is tailored to this concept.
C
C WRITE(OUT,*) 'What is a J-value at which the k-bin is
C 'one-half of its max. height?
C 'Default (arbitrary): Jpeak = 1/2*Jmax'
C READ(INP,*) END=906) FWHM
C GOTO 907
C IF(OUT.EQ.ISUP) THEN
C INP = 5
C IOU = 6
C GOTO 905
C ELSE
C FWHM=PEAK*DBLE(MAXJ(V)/2)
C CALL DFALTD(FWHM,FWHM)
C ENDF
C WRITE(ISAVE,*) FWHM
C
C Note that it doesn't matter if a J-value to the right
C or to the left of Jpeak is specified, since the formula for
C a Gaussian squares the difference.
C
C The formula of interest for the dimensionless case is:
C f(x) = const*exp(-a*(x-xm)**2)
C where xm is the x value where
C f(x) is at a maximum.
C f(x) = -const*2*a*(x-xm)*exp(-a*(x-xm)**2)
C which is zero when x = xm.
C
C The value of kvjbm at J=0 is used to find the value of
C the pre-exponential constant:
C f(0) = const*exp(-a*xm**2)
C Thus const = f(0)*exp(a*xm**2)
C
C The full-width at half-maximum is used to find a.
C Note that f(xh) = 1/2*f(xm)
C where xm is the x value at which
C the peak is one-half its max.
C
C Therefore,
C a = ln 2
C / (xh-xm)**2
C
C In this subroutine PEAK = xm and FWHM = xh-ALPHA = a.
C
C ALPHA = ELOG(2.00001)/(PEAK*FWHM)**2)
C CONST=KVJBM(V,0)*DEXP(ALPHA*PEAK*PEAK)
C DO 100 J = 1,MAXJ(V)
C KVJBM(V,J) = CONST*DEXP(-ALPHA*(DBLE(J)-PEAK)**2)
C CONTINUE
C RETURN
C END
    
```

```

SUBROUTINE EMSCLV
C The scaling of the microscopic bimolecular reaction rate
C constants could have a variety of v-dependencies.
C (i) The simplest model assumes no predictable
C v-dependence--but this could
C mean one of two things: that kvbin varies randomly
C according to v, or that it is a constant value over
C all values of v.
C The random number generated will be bounded by 0 and 1
C so that the average is 1) and
C will multiply the value for kvbin(0,0)
C to produce a value for kvbin(v,0).
C (ii) kvbin might increase monotonically over v, so that the
C reaction coefficient becomes larger as the vibrational
C excitation increases. (This could also occur as a function
C simply of the energy.)
C (iii) kvbin might "decrease" monotonically over v.
C N.B. options (i) and (iii) can be handled together, since the same
C function with different input parameters can give the desired
C behavior (e.g. for a linear function, negative or positive slope
C or for an exponential function a negative or positive exponent).
C
C IMPLICIT DOUBLE PRECISION (B-H,K-N,O,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP,IOU,ISAVE,ISUP,IEBEG,INDET,INOV,VONLY,VUNIT(0:10)
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN,INPTS,ICOLL
COMMON/MASTR/KVJBM(0:10),G(0:10)
SSUB=SPRSUB/EMSCLV
IFLAG = 0
C
C 100 WRITE(OUT,*) 'Select the scaling required for Fvkm:
C WRITE(OUT,*) '1) constant with respect to v
C WRITE(OUT,*) '2) varies randomly between
C '0 and 1 times the Fvkm(v=0) value'
C WRITE(OUT,*) '3) increases or decreases
C 'linearly over energy of v-level'
C WRITE(OUT,*) '4) increases or decreases
C 'exponentially over v
C READ(INP,*) END=903) ANSCL
C GOTO 904
C IF(OUT.EQ.ISUP) THEN
C INP = 5
C IOU = 6
C GOTO 101
C ELSE
C CALL DFALTD(ANSCL,1)
C ENDF
C 904 WRITE(ISAVE,*) ANSCL
C Prompt for value of KVJBM(0,0) and its units.
C CALL PRIM00
C IF (ANSCL.EQ.1) THEN
C CALL SCLV1
C ELSEIF (ANSCL.EQ.2) THEN
C CALL SCLV2
C ELSEIF (ANSCL.EQ.3) THEN
C CALL SCLV3
C ELSEIF (ANSCL.EQ.4) THEN
C CALL SCLV4
C ELSE
C WRITE(OUT,*) 'Error in potential curve specification ANSCL
C WRITE(OUT,*) 'Specify 1, 2, 3 or 4.'
C IFLAG = IFLAG + 1
C IF (IFLAG.LE.1) GOTO 102
C CALL TERMR(ANSCL,SSUB)
C ENDF
C WRITE(OUT,*) '***** MICROSCOPIC BIMOLECULAR *****'
C WRITE(OUT,*) 'THE F(V,J)-F(V,K) CONSTANTS ARE:'
C WRITE(OUT,*) 'V J F(V,J)-F(V,K)'
C DO 100 V=0,MAXV
C WRITE(OUT,*) 'V FVJBM(V,0)
C CONTINUE
C IF(IEBEG.EQ.1) CALL PRINT(0) FVJBM
C RETURN
C END
    
```

```

SUBROUTINE SCLV1
C SCLV1 is one of four scaling routines called by EMSCLV.
C It maps FVJBM(0,0) values into all remaining FVJBM(v,0) positions
C IMPLICIT DOUBLE PRECISION (B-H,K-N,O,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
COMMON/COUNT/MAXV,MAXJ(0:10),MAXUN,INPTS,ICOLL
COMMON/MASTR/KVJBM(0:10),G(0:10)
SSUB=SPRSUB/SCLV1
DO 100 V = 0,MAXV
C KVJBM(V,0) = FVJBM(0,0)
C ...the value of FVJBM(0,0) is obtained during READIN.
C READVJ or FVJMO.
C CONTINUE
C RETURN
C END
    
```

```

SUBROUTINE SCLV2
C SCLV2 is one of four scaling routines called by BMSCLV.
C It generates random values for KVBIM(V,0) centered on KVBIM(0,0)
C bounded by 0 and 2*KVBIM(0,0)
C *** may want to merge with SCLV3 into a generic random number
C generator.
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
C DOUBLE PRECISION R(11)
COMMON/IO/INP, IOUT, ISAVE, ISUP, IEEBEG, INDAT, INOV, VONLV, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
COMMON/MASTR/KVBIM(0:10), 0:20)
SSUB = SCLV3
908 WRITE(IOUT,*) Input 'Seed' for random number generator
      (real number between 1.0E0 and 2.0E10)
      READ(INP,*,END=909) DSEED
      GOTO 916
909 IF(IOUT.EQ.ISUP) THEN
      INP = 5
      IOUT = 6
      GOTO 908
      ELSE
      CALL DFALTD(DSEED,142857D0)
      ENDF
910 WRITE(ISAVE,*) DSEED
      IF (DSEED.LT.1.0D0 .OR. DSEED.GT.2.147483647D0) THEN
      WRITE(IOUT,*) 'Seed is out of range.'
      IFLAG = IFLAG + 1
      IF (IFLAG.LE.1) GOTO 908
      CALL ABTERM(DSEED, SSUB)
      ENDF
C We generate all the random numbers we need with a single call to the
C DMSL routine GSUBS. Note that the number 11 is dependent
C on the size of the vibrational and rotational arrays.
      NR = 11
      CALL GSUBS(DSEED, NR, R)
      DO 300 V = 0, MAXV
      KVBIM(V,0) = KVBIM(0,0)*2.0D0*(R(V)-1)
300 CONTINUE
      RETURN
      END

```

```

SUBROUTINE SCLV3
C SCLV3 is one of four scaling routines called by BMSCLV.
C It scales linearly by the energy gap the KVBIM(V,0) elements
C of the KVBIM(V,0) array from KVBIM(0,0).
C Eqm to use is kvjbm(v,0) = kvjbm(0,0)*v/slope + kvjbm(0,0)
C where slope is input by user
C Rather than have slope input directly program asks for ratio
C of first rate constant to zeroth: ratio = kvjbm(1,0)/kvjbm(0,0)
C then from first eqm (dividing by kvjbm(0,0)) for v=1:
C ratio = 1+slope
C therefore slope = ratio - 1
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP, IOUT, ISAVE, ISUP, IEEBEG, INDAT, INOV, VONLV, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
COMMON/ENERGY/ELEV(0:10,0:20), TEMP
COMMON/MASTR/KVBIM(0:10), 0:20)
SSUB=SPRSUB('SCLV3')
905 WRITE(IOUT,*) Enter the ratio k_xn(v=1,j=0)/k_xn(v=0,j=0)
      READ(INP,*,END=906) RATIO
      GOTO 907
906 IF(IOUT.EQ.ISUP) THEN
      INP = 5
      IOUT = 6
      GOTO 905
      ELSE
      CALL DFALTD(RATIO,1.5D0)
      ENDF
907 WRITE(ISAVE,*) RATIO
      SLOPE = RATIO - 1.0D0
      DO 200 V = 1, MAXV
C The unitless approach is easy to implement and values of the
C on the order of 1.0 seem physically reasonable.
      KVBIM(V,0)=SLOPE*EDEL(V,0,0,0)+KVBIM(0,0)/EDEL(1,0,0,0)
      * KVBIM(0,0)
200 CONTINUE
      RETURN
      END

```

```

SUBROUTINE SCLV4
C SCLV4 is one of four scaling routines called by BMSCLV.
C It scales exponentially the KVBIM(V,0) elements
C of the KVBIM(V,0) array from KVBIM(0,0).
C Eqm to use is kvjbm(v,0) = kvjbm(0,0)*exp(alpha*v)
C where alpha is input by user
C Rather than have alpha input directly program asks for ratio
C of first rate constant to zeroth: ratio = kvjbm(1,0)/kvjbm(0,0)
C then from first eqm (rearranging for v=1: ratio = exp(alpha*1)
C therefore alpha = ln(ratio)
C IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
C IMPLICIT INTEGER (A,I-J,L-M,V)
C IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP, IOUT, ISAVE, ISUP, IEEBEG, INDAT, INOV, VONLV, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
COMMON/ENERGY/ELEV(0:10,0:20), TEMP
COMMON/MASTR/KVBIM(0:10), 0:20)
SSUB=SPRSUB('SCLV4')
905 WRITE(IOUT,*) Enter the ratio k_xn(v=1,j=0)/k_xn(v=0,j=0)
      READ(INP,*,END=906) RATIO
      GOTO 907
906 IF(IOUT.EQ.ISUP) THEN
      INP = 5
      IOUT = 6
      GOTO 905
      ELSE
      CALL DFALTD(RATIO,1.5D0)
      ENDF
907 WRITE(ISAVE,*) RATIO
      ALPHA = DLOG(RATIO)
      DO 200 V = 0, MAXV
C The unitless approach is easy to implement and values of the
C on the order of 1.5 seem physically reasonable.
      KVBIM(V,0) = KVBIM(0,0)*DEXP(ALPHA*V)
200 CONTINUE
      RETURN
      END

```

```

SUBROUTINE COLLECT
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
COMMON/IVIB/RUPUP(0:10,0:20), RUPD(0:10,0:20), RUP0(0:10,0:20)
1 KDOWN(0:10,0:20), KDOWND(0:10,0:20), KDOWN0(0:10,0:20)
2 FUP(0:10,0:20), FDOWN(0:10,0:20)
COMMON/FVIB/FU(0:10,0:20), FUD(0:10,0:20)
1 FUD3(0:10,0:20), FUD33(0:10,0:20), FUD333(0:10,0:20)
2 FUD333(0:10,0:20), FUD3333(0:10,0:20), FUD3333(0:10,0:20)
SSUB=SPRSUB('COLLECT')
WRITE(6,*) ICOLL & ICOLL
C ICOLL denotes which collision partner is involved: A, M or B
C Scheck
DO 100 V=0,MAXV
DO 100 J=0,MAXJ(V)
FUD3(ICOLL,V,J)=RUPUP(V,J)
FUD3(ICOLL,V,J)=RUPD(V,J)
FUD3(ICOLL,V,J)=RUP0(V,J)
FUD3(ICOLL,V,J)=KDOWN(V,J)
FUD3(ICOLL,V,J)=KDOWND(V,J)
FUD3(ICOLL,V,J)=KDOWN0(V,J)
FUD3(ICOLL,V,J)=FUP(V,J)
FUD3(ICOLL,V,J)=FDOWN(V,J)
FUD3(ICOLL,V,J)=FU(V,J)
FUD3(ICOLL,V,J)=FUD(V,J)
100 CONTINUE
C Scheck
RETURN
END

```

```

SUBROUTINE DIFFUS
  The krxn(i,j) should be examined to see which values are
  not reasonable for the system of interest.
  For a bimolecular reaction, this would imply that those krxn
  which exceed the collision frequency possible according
  to the temperature and pressure of the system
  should have an upper limit which is equal to or less
  than the rate of diffusion. The rate-limiting step is
  no longer the speed of chemical conversion, but the
  speed at which the molecules can meet another molecule.
  In this subroutine, the collision frequency is calculated
  and any krxn(i,j) which exceeds it is set equal to it.
  If the user is modeling a unimolecular reaction, or an ideal
  reaction which is not limited by the rate of diffusion, then
  no upper limit in the subroutine DIFFUS should be specified.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  REAL*8 AVOGAD, AMASS, APAD
  PARAMETER (AVOGAD=6.02205E23)
  COMMON/ISUP/ IOUT, ISAVE, ISUP, IDEBUG, INEAT, INOV, VONLY, VUNIT(10)
  COMMON/COU/ COU, MAXV, MAXJ(10), MAXIN, INPTS, ICOLL
  COMMON/MASTP/ MASTP, MASTBIM(10), MASTC
  COMMON/ENERGY/ELEV(10), E(10), TEMP
  COMMON/PHYSCT/PLANGF, FBOLTE, RGAS, SCENPI, CLIGHT, PI, IAEVER
  COMMON/MASS/MASS, CMASS, AMASS, BCMASS, PRAD, CRAD, APAD, RE
  COMMON/PCP/PCP(10), PC(10), NTOT, CONCA, CONCB, CONCM, JC(10)
  JSUB=SPRNB('DIFFUS')
  IFLAG = 0
102 WRITE(IOUT,*) 'Should the microscopic rate constants
  & have an upper limit equal to that of the maximum collision
  & frequency possible?'
  WRITE(IOUT,*) '1) Yes, establish this as an upper limit
  WRITE(IOUT,*) '2) The upper limit is proportional to
  & the collision frequency'
  WRITE(IOUT,*) '3) No upper limit (e.g. unimolecular rxn.)
  READ(INP,*,END=903) ANDIFF
  GOTO 904
903 IF(IOUT.EQ.ISUP) THEN
  INP = 5
  IOUT = 6
  GOTO 102
ELSE
  CALL DPALTI(ANDIFF,1)
  ENDF
904 WRITE(IGAVE,*) ANDIFF
  IF (ANDIFF.EQ.3) THEN
  GO TO 104
  ELSEIF (ANDIFF.NE.1.AND.ANDIFF.NE.2) THEN
  WRITE(IOUT,*) 'Error in upper limit specification. ANDIFF
  WRITE(IOUT,*) Specify 1, 2 or 3.
  IFLAG = IFLAG + 1
  IF (IFLAG.LE.1) GOTO 102
  CALL INTERM(ANDIFF,JSUB)
  ENDF
  The rate of collision, Z, may be expressed in a form analogous to
  the rate of reaction with a diffusion coefficient, k_diff:
  Z = k_diff [A] [B]
  where [A] and [B] are the concentrations of the species involved
  in collision.
  For each vj level
  we want to compare
  k_diff with the k_vjBim coefficient input to the
  program. Since only collisional processes are considered in this
  simulation, no k_vjBim should exceed k_diff.
  The formula for Z is from J.N. Levine, Physical Chemistry
  (McGraw-Hill, 1978), p. 421:
  Z = pi^1/2 * (r_A + r_B)^2 * (8 * RT / pi)^1/2 * (1/M_A + 1/M_B)^1/2 *
  * N_A * V^1/2 * N_B * V^1/2
  Therefore, k_diff = Z / ([A] * [B] * N_A * V)
  where the additional N_A V in the denominator comes in because we are
  considering the number of "moles" of collisions rather than the
  number of collisions.
  or, k_diff = pi^1/2 * (r_A + r_B)^2 * (8 * RT / pi)^1/2 *
  * RECIPROCAL-REDUCED-MASS^1/2 * N_A * V
  This will be used to estimate the upper limit for
  the bimolecular rate coefficient.
  The molecular diameter is estimated as the sum of atomic
  radii, whose values are passed through common block MASS.
  The atomic masses and radii are found in subroutines RNDYS.
  Since we are concerned only with reactive collisions (A-B)
  we ignore M-BC collisions in determining k_diff.
  REC = (REAO+CRAD)
  BCMASS=BMASS+CMASS
  RECRM = (1.000/AMASS + 1.000/BCMASS)*AVOGAD
  CMASS and BCMASS come in on a 'per mole' basis...we need them
  on a 'per molecule' basis, so we divide the mass by N_A V, which
  is equivalent to multiplying the reciprocal by N_A V as above.
  KDIFF=PI*(REAO+CRAD)**2*(8.000*RGAS/TEMP/PI)*RECRM*AVOGAD
  WRITE(IOUT,700) KDIFF
700 FORMAT(' Collision coefficient (k_diff) is .D12.4 sec-1 )
  IF(ANDIFF.EQ.2) THEN
105 WRITE(IOUT,700) 'What is the ratio of the upper limit to the
  & collision coefficient? (Default: 1.0)'
  READ(INP,*,END=908) RATIO
  GOTO 909
908 IF(IOUT.EQ.ISUP) THEN
  INP = 5
  IOUT = 6
  GOTO 105
ELSE
  CALL DPALTD(RATIO,1.000)
  ENDF
909 KDIFF=RATIO*KDIFF
  ENDF
  DO 100 V=0, MAXV
  DO 100 J=0, MAXJ(V)
  IF(VJUBIM(V,J).GT.KDIFF) THEN
  WRITE(IOUT,701) V,J,VJUBIM(V,J),KDIFF
  VJUBIM(V,J)=KDIFF
701 FORMAT(' VjBim(.,.D12.) exceeds the collision

```

```

& coefficient: .D12.4 > .D12.4 / Value reset.)
100 CONTINUE
104 RETURN
END

SUBROUTINE PIPPOP(ANPOP)
  Selection of the form of population distribution.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  COMMON/PCP/PCP(10), PC(10), NTOT, CONCA, CONCB, CONCM, JC(10)
  COMMON/PHYSCT/PLANGF, FBOLTE, RGAS, SCENPI, CLIGHT, PI, IAEVER
  COMMON/COU/ COU, MAXV, MAXJ(10), MAXIN, INPTS, ICOLL
  COMMON/ENERGY/ELEV(10), E(10), TEMP
  COMMON/COU/ COU, MAXV, MAXJ(10), MAXIN, INPTS, ICOLL
  JSUB=SPRNB('PIPPOP')
  IFLAG = 0
104 WRITE(IOUT,*) 'Is the population distributed according to
  WRITE(IOUT,*) '1) The Boltzmann equation?
  WRITE(IOUT,*) '2) Laser excitation?
  WRITE(IOUT,*) '3) Delta function?
  WRITE(IOUT,*) '4) A database of IV/V J, v, J pairs?
  IF (VONLY.EQ.0) WRITE(IOUT,*) '5) A database of IV/V J, v, J tripler?
  READ(INP,*,END=900) ANPOP
  GO TO 901
  IF(IOUT.EQ.ISUP) THEN
  INP = 5
  IOUT = 6
  GO TO 104
  ELSE
  CALL DPALTI(ANPOP,1)
  ENDF
901 WRITE(IGAVE,*) ANPOP
  IF (ANPOP.EQ.1) THEN
  CALL BOLTZ
  ELSEIF (ANPOP.EQ.2) THEN
  CALL LASER
  ELSEIF (ANPOP.EQ.3) THEN
  CALL DELTA
  ELSEIF (ANPOP.EQ.4) THEN
  CALL READIN(61,NV)
  IF (VONLY.EQ.0) CALL JPOLTC
  ELSEIF (ANPOP.EQ.5.AND.VONLY.EQ.0) THEN
  CALL READIN(62,NV)
  ELSE
  WRITE(IOUT,*) 'Error in choice of vibrational population.
  IF (VONLY.EQ.1) THEN
  WRITE(IOUT,*) ANPOP Specify 1, 2, 3 or 4.
  ELSE
  WRITE(IOUT,*) ANPOP Specify 1, 2, 3, 4 or 5.
  ENDF
  IFLAG = IFLAG + 1
  IF (IFLAG.LE.1) GO TO 104
  CALL INTERM(ANPOP,JSUB)
  ENDF
  The populations of each level (no matter how obtained) must be
  added up to get a total population.
  NTOT=0.0D0
  For accuracy's sake the smallest numbers should be added together
  first, and the largest numbers at the end.
  We will not do a size-sort however, but instead go backwards
  through the loop to approximate a smaller-to-larger addition.
  e.g. 1991.026, what we are actually doing is calculating
  the partition function so that we can divide by it to get true
  N_v's.
  DO 404 V=MAXV,0,-1
  DO 404 J=MAXJ(V),0,-1
  NTOT=NTOT+CONCB(V,J)
404 CONTINUE
  WRITE(IOUT,*) 'The initial unnormalized population is
  & NTOT
  CONCM=CONCBC/NTOT
  DO 405 V=0,MAXV
  DO 405 J=0,MAXJ(V)
  N_VJ=J*NTOT*CONCM
  WRITE(IOUT,*) ' V J ELEV(V,J) N_VJ J'
405 CONTINUE
  adjust CONCBC over so slightly to avoid small numerical errors
  later on...e.g. 1991.626. Then adjust conca and concm
  to keep them in sync.
  CALD = CONCBC
  CONCBC=0.0D0
  DO 406 V=MAXV,0,-1
  DO 406 J=MAXJ(V),0,-1
  CONCB = CONCBC+CONCB(V,J)
406 CONTINUE
  CONCA = CONCBC/CALD*CONCA
  CONCM = CONCBC/CALD*CONCM
  CONCB = CONCBC/CALD*CONCB
  WRITE(IOUT,*) 'The populations have been adjusted to an
  & initial total value of CONCM
  & using a scaling factor of CONCMST
  IF(IDEBUG.EQ.1) CALL PRINTU(10,NV)
  RETURN
  END

```

# APPENDIX C. SOURCE CODE

```

SUBROUTINE BOLTZ
  C Calculation of population based on the Boltzmann distribution.
  C degeneracy factor for the population level is from Farplus & Porter
  C page 474.
  C Since Elev is per-molecule, the Boltzmann constant is used.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, VONLY, VUNIT(0:23)
  COMMON/PHYSCT/PLANCK, PBOLTZ, RGAS, SCEXP1, CLIGHT, PI, I4EVER
  COMMON/ENERGY/ELEV(0:10,0:20), TEMP
  COMMON/COU/NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  COMMON/POP/NV(0:10,0:20), NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  ISUB=SPRUB('BOLTZ')
  BCONST=-1.0D0/(RGAS*TEMP)
  WRITE(IOUT,*) ***** INITIAL BOLZMANN DISTRIBUTION (j0/mol)
  7 FORMAT(1X, 'v', 11X, 'J', 12X, 'Elev', 15X, 'N(v,j)')
  NV0 = NV(0,0)
  DO 100 V=0, MAXV
    DO 100 J=0, MAXJ(V)
      BATERM = BCONST*EDEL(0.0,V,J)
      NV(V,J) = DBLE(2*J + 1)*NV0*DEXP(BATERM)
      IF(NV(V,J).GT.0.0D0) WRITE(IOUT,*) V,J,ELEV(V,J),NV(V,J)
  100 CONTINUE
  RETURN
  END

```

```

SUBROUTINE LASER
  C This routine calculates the population distribution to simulate a laser
  C excitation experiment in which molecules in a particular v, J level
  C make a transition to another level. Thus the number of molecules in
  C the initial level decreases while the number of molecules in the final
  C level increases.
  Parameters: None.
  Calling routine: PIKPOP.
  Called routines: Boltz.
  Variables:
  VLO, JLO, VHI, JHI---these are the indices for the lower (VLO,
  JLO) and upper (VHI, JHI) levels corresponding to the transition
  selected.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  LOGICAL*1 EXIST
  COMMON/COU/NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, VONLY, VUNIT(0:23)
  COMMON/ENERGY/ELEV(0:10,0:20), TEMP
  COMMON/POP/NV(0:10,0:20), NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  ISUB=SPRUB('LASER')
  200 WRITE(IOUT,*) 'What transition is involved in the laser?'
  WRITE(IOUT,*) 'excitation? (Input 4 integers, separated by a'
  WRITE(IOUT,*) 'comma corresponding to v j --> v' j' E.g.,'
  WRITE(IOUT,*) 'v=0 J=0 --> v=3, J=2 would be input as 0 0 3 2)'
  WRITE(IOUT,*) 'Default is 0,0 --> 1,0 transition.'
  READ(INP,*,END=900) VLO,JLO, VHI,JHI
  GO TO 501
  900 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 200
  ELSE
    CALL DFALTI(VLO,0)
    CALL DFALTI(JLO,0)
    CALL DFALTI(VHI,1)
    CALL DFALTI(JHI,0)
  ENDDIF
  901 WRITE(ISAVE,*) VLO,JLO,VHI,JHI
  IFLAG=0
  C Check to see that VLO and VHI are within the proper range.
  IF(.NOT.EXI.(VLO,JLO,VHI,JHI)) THEN
    WRITE(IOUT,*) 'Transition levels out of bounds VLO,JLO'
    6 VHI,JHI
    IFLAG = IFLAG + 1
    IF (IFLAG.LE.1) GO TO 200
    WRITE(IOUT,*) 'Terminating in LASER due to v or J bounds.'
    CALL ISTERM(VHI,ISUB)
  ENDDIF
  101 WRITE(IOUT,*) 'What fraction of the initial population'
  WRITE(IOUT,*) 'in the lower level has been excited to the upper'
  WRITE(IOUT,*) 'level?'
  READ(INP,*,END=903) FRACEX
  GO TO 903
  902 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 101
  ELSE
    CALL DFALTI(FRACEX,0.5D0)
  ENDDIF
  903 WRITE(ISAVE,*) FRACEX
  C Check to see that fraction is valid.
  IF(FRACEX.LE.0.0 OR FRACEX.GT.1.0) THEN
    WRITE(IOUT,*) 'Fraction', FRACEX, 'must be between 0 and 1.'
    WRITE(IOUT,*) 'Respecify fraction within bounds.'
    IFLAG = IFLAG + 1
    IF(IFLAG.LE.1) GO TO 101
    WRITE(IOUT,*) 'Program Terminating in LASER.'
    CALL ISTERM(FRACEX,ISUB)
  ELSE
    CALL BOLTZ
    NVEXC = NV(VLO,JLO)*FRACEX
    NV(VHI,JHI) = NV(VHI,JHI) + NVEXC
    NV(VLO,JLO) = NV(VLO,JLO) - NVEXC
  ENDDIF
  RETURN
  END

```

```

SUBROUTINE DELTA
  C This routine is an idealized representation of a very restrictive
  C nascent population. It will set
  C all populations to zero, and the fraction in the desired level
  C will be equal to the total population.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  LOGICAL*1 EXIST
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, VONLY, VUNIT(0:23)
  COMMON/POP/NV(0:10,0:20), NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  COMMON/COU/NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  ISUB=SPRUB('DELTA')
  IFLAG=0
  200 WRITE(IOUT,*) 'Situates the delta function at which v,j level?'
  WRITE(IOUT,*) '<Default: 1 1>'
  READ(INP,*,END=900) ANV,ANJ
  GO TO 501
  900 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 200
  ELSE
    CALL DFALTI(ANV,1)
    CALL DFALTI(ANJ,1)
  ENDDIF
  901 WRITE(ISAVE,*) ANV,ANJ
  C Check to see that this level is within the proper range.
  C (To use EXIST, function, look for 0,0->anv,anj transition.)
  IF(.NOT.EXIST(0,0,ANV,ANJ)) THEN
    WRITE(IOUT,*) 'Specified level out of bounds ANV,ANJ.'
    4 Value must be between 0 and MAXV
    IFLAG=IFLAG+1
    IF (IFLAG.LE.1) GO TO 200
    WRITE(IOUT,*) 'Terminating in DELTA due to v bounds.'
    CALL ISTERM(ANV,ISUB)
  ENDDIF
  DO 100 V=0, MAXV
    DO 100 J=0, MAXJ(V)
      NV(V,J)=0.0D0
  100 CONTINUE
  NV(ANV,ANJ)=1.0D0
  RETURN
  END

```

```

SUBROUTINE JBOLTZ
  C Given an (V,NV) vector, this routine will calculate values for
  C the J-levels for each vibrational level, assuming a Boltzmann
  C distribution of the J-levels. For the moment we are assuming
  C the rotational temperature will be coincident with the
  C experimental temperature.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INOV, VONLY, VUNIT(0:23)
  COMMON/PHYSCT/PLANCK, PBOLTZ, RGAS, SCEXP1, CLIGHT, PI, I4EVER
  COMMON/ENERGY/ELEV(0:10,0:20), TEMP
  COMMON/COU/NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  COMMON/POP/NV(0:10,0:20), NTOT, CONCA, CONCBC, CONCNC, JCOLL(3)
  ISUB=SPRUB('JBOLTZ')
  BCONST = -1.0D0/(KBOLTZ*TEMP)
  WRITE(IOUT,*) ***** J-level BOLZMANN DISTRIBUTIONS *****
  WRITE(IOUT,*)
  7 FORMAT(1X, 'v', 10X, 'J', 12X, 'Elev', 15X, 'N(v,j)')
  DO 100 V=0, MAXV
    NV0 = NV(V,0)
    DO 100 J=0, MAXJ(V)
      C degeneracy factor for the population level is from Farplus & Porter
      C page 474.
      C Since EDEL is per-molecule, the Boltzmann constant is used.
      NV(V,J) = DBLE(2*J + 1)*NV0*DEXP(BCONST*EDEL(V,0,V,J))
      IF(NV(V,J).GT.0.0D0) WRITE(IOUT,*) V,J,ELEV(V,J),NV(V,J)
  100 CONTINUE
  RETURN
  END

```



```

905 WRITE(IOUT,*) ' Input the preferred time step size.
WRITE(IOUT,*) '<Carriage return: time integrator routine decides.>
READ(INP,END=906) H
INDEX=0
GO TO 907
906 IF(IOUT.EQ.ISUP) THEN
  INP=5
  IOUT=6
  GO TO 905
ELSE
  C Dummy call to DFALTD to make sure we can read from INP the next time
  CALL DFALTD(H)
ENDIF
907 WRITE(ISAUT,*) H
DO 100 I=COUNT,1,14EVER
  TEND = I*H*(COUNT)/0.0000100
  TEND = T * H
Warning: Index=3 causes DGEAR to work much more slowly. Don't use
unless you really want to see the result after each step.
  INDEX=3
  ...will make DGEAR return after one step...
  CALL DGEAR(IVLVL,MASTER,JACOBI,INVPAC,TEND,TUL,METH,MITER
  A INDEX,WF,WF,IER)
  IF(IER.GT.128) GO TO 30
  IF ((COUNT.LT.LASTI .AND. MOD(COUNT,INDEX).EQ.0) THEN
  IF ((COUNT.LT.LASTI .AND. MOD(COUNT,INDEX).EQ.0) .OR.
  A ((COUNT.LT.LASTI .AND. MOD(COUNT,INDEX).EQ.0) .OR.
  C ((COUNT.GE.LASTI .AND. MOD(COUNT,INDEX).EQ.0) THEN
  A CALL DUMP(IVLVL,T,INVPAC,TEND)
  IF((CONCNC-NTOT)/NEND.GE.PRACT) GO TO 210
  ENDF
200 CONTINUE
210 WRITE(IOUT,*) ' SOLVER ends after ',COUNT,' iterations.
WRITE(IOUT,*) ' Max. no. of iterations allowed = ',14EVER
WRITE(IOUT,*) ' Suggested TEND value=??'
RETURN
30 CONTINUE
WRITE(IOUT,*) ' Problems in DGEAR. IER = ',IER
RETURN
END
  
```

```

SUBROUTINE MASTER (I,VLVL,T,INVPAC,INVPAC)
C The arguments of MASTER are set by IMSL specifications.
C (see DGEAR.) MASTER evaluates the c.d.e.'s to be integrated
C by DGEAR.
IMPLICIT DOUBLE PRECISION (B-H,F,N,Q,W,Z)
IMPLICIT INTEGER (A,I-C,L-M,V)
IMPLICIT CHARACTER*6 (S)
LOCALS = ENDF
COMMON/INVPAC/FUN3(3,0:10,0:10),FUN3(3,1:1,0:10)
1 FUN3(3,0:10,0:10),FUN3(3,1:1,0:10),FUN3(3,1:1,1:10)
2 FUN3(3,0:10,0:10),FUN3(3,1:1,0:10),FUN3(3,1:1,1:10)
3 FUN3(3,0:10,0:10),FUN3(3,1:1,0:10),FUN3(3,1:1,1:10)
COMMON/MASTER/FUN3IM(0:10,0:10)
COMMON/PP/IN(V,1:3,0:10),NTOT,CONCA,CONCB,CONCM,CONCL,3,
COMMON/CONST/MAXI,MAXJ(1:10),MAXK,INPTS,ICOLL
DIMENSION IN(V,1:10,0:10)
4 INVPAC(I,VLVL),INVPAC(I,VLVL),CONCA,
EQUIVALENCE (CONCA),CONCA)
5 SUBS=PPSUB( MASTER )
C Since DGEAR requires a population vector but we want to
C work with a v, J population, array we must unpack the array supplied
C by DGEAR (NVI) at the beginning of MASTER and pack the array required
C by DGEAR at the end of MASTER (CONV)
  IVLVL = 0
  NTOT = 0,D3
  DO 100 V = 0,MAXV
    DO 100 J = 0,MAXJ(V)
      INVLVL = IVLVL + 1
      NV(I,J) = INVPAC(I,VLVL)
      NVTOT=N(I,J)+NTOT
100 CONTINUE
  NVA = CONCA - (CONCB - NVTOT)
  DO 200 V=0,MAXV
    DO 200 J=0,MAXJ(V)
      DNVA=0.0D0
      NV(I,V,J)
      DO 300 IC = 1,ICOLL
        IF(EXIST(V-1,J,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J)-DNVA
          DNVA=DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-1,V,J)) THEN
          DNVA= DNVA+FUN3(IC,V-1,J-1)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-2,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-2)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-3,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-3)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-4,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-4)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-5,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-5)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-6,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-6)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-7,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-7)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-8,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-8)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-9,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-9)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        IF(EXIST(V-1,J-10,V,J)) THEN
          DNVA=DNVA+FUN3(IC,V-1,J-10)-DNVA
          DNVA= DNVA+FUN3(IC,V,J) + DNVA
        ENDF
        DNVA= DNVA+CONC(ICOLL,IC)
300 CONTINUE
      DNVA=DNVA+NVA*FUN3IM(V,J) + DNVA
      DN(V,J)=DNVA
200 CONTINUE
  IVLVL = 0
  C Now we must pack the v, J matrix of derivatives into a vector.
  DO 400 V = 0,MAXV
    DO 400 J = 0,MAXJ(V)
      INVLVL = IVLVL + 1
      INVPAC(I,VLVL) = DN(V,J)
400 CONTINUE
  WRITE(6,*) ' INVPAC: ',(INVPAC(I),I=1,MIN(I,VLVL,10))
  WRITE(6,*) ' CONVPAC: ',(CONVPAC(I),I=1,MIN(I,VLVL,10))
  RETURN
END
  
```

```

SUBROUTINE JACOB(I,VLVL,T,INVPAC,PD)
C This (dummy) subroutine supplies the Jacobian matrix (if req'd)
C for the DGEAR subroutine. Arguments are as specified in IMSL
C documentation for DGEAR.
IMPLICIT CHARACTER*6 (S)
INTEGER I,VLVL
DOUBLE PRECISION INVPAC (/), PD(I,VLVL,I,VLVL),T
5 SUBS=PPSUB( JACOB )
RETURN
END
  
```

```

SUBROUTINE DUMP (IVLVL, T, NVFAC, NENT)
C It is not obvious that the sum of DNV(V, J) over the V and J
C levels gives DNTOT. The following example shows that this is
C the case:
C Given a 3-level system where 40 molecules/sec leave the
C first level, 4 to react and 36 to go to the second level,
C 35 molecules/sec leave the third level--13 react and 22 fall
C back to the second level.
C Therefore, 53 molecules/sec enter the second level and
C 10 from the third, 4 from the second--4
C (41-53-25) must be leaving the system.
C That is: DNV(1)+DNV(2)+DNV(3) = DNTOT.
C
IMPLICIT DOUBLE PRECISION (B-H, F, N-O, W-Z)
IMPLICIT INTEGER (A, I, J, L-M, V)
IMPLICIT CHARACTER*6 (S)
COMMON/POP/NV(0:10) 0:20, NTOT, CONCA, CONCB, CONCM, CONCL, S
COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUM, INPTS, ICOLL
COMMON/STAT/STATBIN(0:10) 0:20
DIMENSION DNV(0:10) 0:20, NVFAC(23), INTPAC(23)
SUBSPRUB( DUMP )
C Master will calculate the DNVFAC vector given the INPAT vector
CALL MASTER (IVLVL, T, NVFAC, DNVFAC)
C Now unpack the DNVFAC, NVFAC vectors.
IVLVL = 0
DO 100 V = 0, MAXV
  DO 100 J = 0, MAXJ(V)
    IVLVL = IVLVL + 1
    DNV(V, J) = DNVFAC(IVLVL)
    NV(V, J) = NVFAC(IVLVL)
100 CONTINUE
NTOT = 0.0D0
DNTOT = 0.0D0
VTVV = 0.0D0
DO 200 V = 0, MAXV
  NLUMP = 0.0D0
  DO 201 J = 0, MAXJ(V)
    NTOT = NV(V, J) + NTOT
    NLUMP = NV(V, J) * NLUMP
  CONTINUE
  VTVV = VTVV + NV(V, J) * NV(V, J)
  WRITE (UNIT=V, 701) V, J, T, DNV(V, J), NLUMP
701 FORMAT(2I5, 3I2X, D16.10)
201 CONTINUE
DNLUMP = 0.0
DO 202 J = 0, MAXJ(V)
  DNTOT = DNV(V, J) + DNTOT
  DNLUMP = DNV(V, J) * DNLUMP
  WRITE (UNIT=V+10, 701) V, J, T, DNV(V, J), DNLUMP
202 CONTINUE
200 CONTINUE
VTVV = VTVV / NTOT
VVA = CONCA + (CONCB - NTOT)
VPHEN = -DNTOT / (NTOT + VVA)
KTEXT = (CONCB - NTOT) / NTOT
WRITE (1, 702) T, KTEXT, DNTOT, VPHEN, VTVV
702 FORMAT(5I2X, D16.4)
WRITE (6 *) 'Time, Total Population: ', T, NTOT
RETURN
END

```

```

SUBROUTINE EIGEN (ICTRI)
IMPLICIT DOUBLE PRECISION (B-H, F, N-O, W-Z)
IMPLICIT INTEGER (A, I, J, L-M, V)
IMPLICIT CHARACTER*6 (S)
COMPLEX*8 W(200), Z(100) 200
DIMENSION WF(40400)
C ... this dim. based on MAXPLV(200) dim. MAXMLV(2) * MAXPLV
COMMON/RYB/RUP(0:10) 0:20, RPD(0:10) 0:20, RPD(0:10) 0:20, RPD(0:10) 0:20
1 RPD(0:10) 0:20, RPD(0:10) 0:20, RPD(0:10) 0:20
2 RPD(0:10) 0:20, RPD(0:10) 0:20, RPD(0:10) 0:20
3 RPD(0:10) 0:20
COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUM, INPTS, ICOLL
C
C EIGEN: WF and Z are dummy arguments and are in the argument list
C above to allow the two arrays in question to have variable
C dimensions - otherwise the arrays would be inefficiently large.
C
SUBSPRUB( EIGEN )
MAXMLV = 200
IF (IDEBUG.EQ.1) THEN
  WRITE (6, *) 'order of KLUMP = ', ICTRI
  ICMIN = MIN(ICTRI, 5)
  WRITE (6, *) 'KLUMP = ', ((KLUMP(I, J) I=1, ICMIN; J=1, ICMIN))
ENDIF
CALL EIGRF (KLUMP, ICTRI, MAXMLV, 2, W, Z, MAXMLV, WF, IER)
C ... pls see DMSLDLIB documentation for explanation of EIGRF arguments
IF (IER.GT.128) THEN
  WRITE (IOUT, *) 'Error in IMSL routine EIGRF.'
  CALL ITERM(IER, SSUB)
ENDIF
WRITE (IOUT, *) 'Eigenvalues of the master equation are'
  (complex numbers):
DO 10 I=1, ICTRI
  WRITE (IOUT, *) Z, W(I)
10 CONTINUE
WRITE (IOUT, *) 'The eigenvector matrix according to'
  (row, column) is complex notation is:
LROW = MIN(ICTRI, 10)
DO 200 M = 1, LROW
  WRITE (IOUT, *) '*****'
  DO 200 L = 1, LROW
    WRITE (IOUT, *) M, L, Z(M, L)
  CONTINUE
200 CONTINUE
RETURN
END

```

```

SUBROUTINE TABOUT
C Tables created are suitable for input to a plotting routine
C or are TEK files suitable for incorporation into documents.
IMPLICIT DOUBLE PRECISION (B-H, F, N-O, W-Z)
IMPLICIT INTEGER (A, I, J, L-M, V)
IMPLICIT CHARACTER*6 (S)
DIMENSION X(100), NV(0:10) 0:20
COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUM, INPTS, ICOLL
COMMON/RYB/RUP(0:10) 0:20, RPD(0:10) 0:20, RPD(0:10) 0:20, RPD(0:10) 0:20
COMMON/STAT/STATBIN(0:10) 0:20
SUBSPRUB( TABOUT )
C For the moment the following is a subset
101 IFLAG = 0
990 WRITE (IOUT, *) 'For which V, J level do you wish to plot?'
  WRITE (IOUT, *) 'Default is 0 & 0 level.'
  READ (1, *) END=991, VV, JJ
  GO TO 991
991 IF (IOUT.EQ.IOUT) THEN
  INP = 5
  IOUT = 6
  GO TO 990
ELSE
  CALL DFAULT(VV, JJ)
  CALL DFAULT(JJ, JJ)
ENDIF
992 WRITE (ISAVE, *) VV, JJ
  IF (VV.LT.0 OR VV.GT.10) THEN
    WRITE (IOUT, *) 'V level incorrectly specified: V must lie'
    * ' between 0 and 10 inclusive.'
    IFLAG = IFLAG + 1
    IF (IFLAG.LE.1) GO TO 101
    CALL ITERM(VV, SSUB)
  ENDF
  IFLAG = 0
  IF (JJ.LT.0 OR JJ.GT.10) THEN
    WRITE (IOUT, *) 'J level incorrectly specified: J must lie'
    * ' between 0 and 100 inclusive.'
    IFLAG = IFLAG + 1
    IF (IFLAG.LE.1) GO TO 101
    CALL ITERM(JJ, SSUB)
  ENDF
DO 100 I=1, ISEVER
  READ (UNIT=V, *) END=991, X
  NV(V, I) = X
  WRITE (IOUT, *) I, NV(V, I)
100 CONTINUE
101 WRITE (IOUT, *) 'If you want output for more V levels, type '
  READ (1, *) END=991, IANS
  GO TO 991
993 IF (IOUT.EQ.IOUT) THEN
  INP = 5
  IOUT = 6
  GO TO 990
ELSE
  CALL DFAULT(IANS, JJ)
ENDIF
994 WRITE (ISAVE, *) IANS
  IF (IANS.EQ.1) GO TO 101
  RETURN
END

```

```

SUBROUTINE PLOTOUT
C DISPLA is the plotting program used. Modifications needed to use
C a different plotting routine should be minor.
C The routine will eventually have the capability to output
C plots directly to the plotting device or to create a file suitable
C for incorporation into a larger document.
IMPLICIT DOUBLE PRECISION (B-H, F, N-O, W-Z)
IMPLICIT INTEGER (A, I, J, L-M, V)
IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUM, INPTS, ICOLL
C For the moment a dummy routine.
SUBSPRUB( PLOTOUT )
WRITE (IOUT, *) 'PLOTOUT has been called.'
RETURN
END

```

```

SUBROUTINE PRINTV (INT, IPRAC)
IMPLICIT DOUBLE PRECISION (B-H, F, N-O, W-Z)
IMPLICIT INTEGER (A, I, J, L-M, V)
IMPLICIT CHARACTER*6 (S)
C assumed size ... does this save on memory?
DIMENSION DPRE(0:10)
COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INCV, VONLY, VUNIT(0:23)
COMMON/COUNT/MAXV, MAXJ(0:10), MAXUM, INPTS, ICOLL
C For the moment a dummy routine.
SUBSPRUB( PRINTV )
RETURN
END

```

```

FUNCTION EDEL (I, J, K, L)
IMPLICIT DOUBLE PRECISION (B-H, F, N-O, W-Z)
IMPLICIT INTEGER (A, I-M, V)
IMPLICIT CHARACTER*6 (S)
COMMON/ENERGY/ELEV(0:10) 0:20, TEMP
SUBSPRUB( EDEL )
EDEL = ELEV(K, L) - ELEV(I, J)
RETURN
END

```

```

FUNCTION SPRSUB(SNAME)
  IMPLICIT CHARACTER*6 (S)
  COMMON/IO/INP,IOOUT,ISAVE,ISUP,IDERUG,INDAT,INOV,WOONLY,VUNIT(0:23)
  $SUB=SPRSUB
  IF (IDEBUG.EQ.1) WRITE(IOUT,701) SNAME
701 FORMAT('*** Routine ',A6,' called. ')
  SPRSUB = SNAME
  RETURN
END

SUBROUTINE OPNFIL
  IMPLICIT INTEGER (A I-J,L-M,V)
  IMPLICIT CHARACTER*6 (S)
  COMMON/IO/INP,IOOUT,ISAVE,ISUP,IDERUG,INDAT,INOV,WOONLY,VUNIT(0:23)
  COMMON/CONF/MAKV
  OPEN FILE on units 10 ---> 20 reserved for INOV
  40 ---> 50 reserved for INOV
  31, 51 reserved for INTOT and ENTOT resp.
  This subroutine will eventually have machine-readable files.
  Instead of formatted files (which is actually TABOUT's function)
  but for ease of debugging we have kept these files being written
  in an inefficient but humanly-readable way.
  File must be opened for direct access.
  RECL must be specified for direct access.
  Records must be written direct access in order to be
  read direct access.
  Syntax of the READ cmd is READ(un.fmt,REC=rec)
  First entry in each record is TIME; remaining entries are
  pop. (der. pop.). Each file has information about one
  v level and its J sub-levels. Record length is not
  assuming double precision (8 bytes).
  $SUB=SPRSUB(OPNFIL)
  DO 10 V = 0, MAKV
    WRITE(VUNIT(V),*) ' v J time n(v,J)'
    OPEN(VUNIT(V),FORM='UNFORMATTED',ACCESS='DIRECT',
  4 RECL=MAKV(V),11*8)
    WRITE(VUNIT(V+21),*) ' v J time dn(v,J)'
    OPEN(VUNIT(V+21),FORM='UNFORMATTED',ACCESS='DIRECT',
  4 RECL=MAKV(V),11*8)
  10 CONTINUE
  WRITE(8,*) ' time total population'
  WRITE(9,*) ' time rate of decrease of total population'
  RETURN
END

```

```

SUBROUTINE SIU(INTYPE,VALUE,UNIT)
  *** FOR ALL CALLS TO SIU, MAKE SURE INTYPE IS CONSISTENT:
  This subroutine converts units specified for the following
  types to SI MKS units, according to the tens-digit of the
  variable INTYPE:
  (1) energy
  (2) rate constants
  (3) length inverse length
  (4) mass
  (5) temperature
  (6) population
  (7) pressure
  (8) concentration
  For example: CALL SIU(13,10.0,CM) means that the type 1
  energy, since integer(13/10)=11 has been selected. The value
  of 10.0 cm-1 has been passed to subroutine SIU, which will
  convert it to MKS units (namely KJ/mol).
  Conversion factors are based on W.H. Flygare(1978):
  IMPLICIT DOUBLE PRECISION(A-H,J-Z)
  IMPLICIT INTEGER*4 (I)
  IMPLICIT CHARACTER*6 (S)
  CHARACTER*3 UNIT
  INTEGER VUNIT,WOONLY
  COMMON/IO/INP,IOOUT,ISAVE,ISUP,IDERUG,INDAT,INOV,WOONLY,VUNIT(0:23)
  PARAMETER (AVOGAD=6.02205E23)
  PARAMETER (EVTOJO=1.60219E-19*AVOGAD, CATOJO=1.96648E-23*AVOGAD,
  1 KCTOJO=6.95246E-21*AVOGAD, CATOJO=6.95246E-24*AVOGAD,
  2 HTOJO=1.0E-03, HATOJO=0.4359815E-17*AVOGAD)
  PARAMETER (MITOSE=1.0E+00/60.0E0, HRTOSE=1.0E+00/3600.0E0)
  PARAMETER (BOTOME=5.28177E0E-11, AOTOME=1.0E-10,
  1 NMTOME=1.0E-09, CMTOME=1.0E-02,
  2 PNTOME=1.0E-12)
  C Values considered are for grams/mole 'not' per molecule.
  PARAMETER (AOTOG=1.0E-03, COTOG=1.0E-03)
  PARAMETER (CETOKE=273.15E-01)
  PARAMETER (TOTOPA=133.189, KPTOPA=1.0E-3,
  1 ATOPA=1.013E+5, LBCTOPA=6894.16,
  2 BPTOPA=1.0E+5, MNTOPA=TOTOPA)
  C Suggestion for future improvement: for the symbol AB read in
  C for the units, this could in fact be AB, ab, Ab or aB.
  C We should construct a CASE function or subroutine, which
  C will automatically convert AB to upper case.
  C Then, the IF stmts may be shortened.
  $SUB=SPRSUB(SIU)
  INTYPE= (INTYPE/10)
  C The reason that the number representing the type of conversion
  C needed must be integer-divided by 10
  C is that this number serves to store more than one
  C type of information: not only the kind of quantity (e.g., energy
  C rate constant, etc.), but the physical
  C meaning associated with that number (for purposes of reading in
  C values in READW and READIN. For example there is more than one type
  C of rate constant (quantity type 2) that is read in but each must have
  C the same units-conversion applied and therefore the code associated
  C with those values (READ) is numbered 21, 22, 23, etc.
  C This has the drawback that the program will have to be changed
  C substantially if, for example, there were more than 10 types of rate
  C constants between which we wished to distinguish.
  C The advantage of this approach is that the program has consistency
  C in the codes used by subroutines READIN, READW, SIU, NOVICE
  C PRINTV, and routines that call these.
  IFLAG=0
  101 IF (INTYPE.EQ.1) THEN
    IF (UNIT.EQ.'EV'.OR.UNIT.EQ.'eV') THEN
      VALUE=VALUE*EVTOJO
      RETURN
    ELSEIF (UNIT.EQ.'CM'.OR.UNIT.EQ.'cm') THEN
      VALUE=VALUE*CMTOJO
      RETURN
    ELSEIF (UNIT.EQ.'JO'.OR.UNIT.EQ.'jo') THEN
      RETURN
    ELSEIF (UNIT.EQ.'KC'.OR.UNIT.EQ.'kc') THEN
      VALUE=VALUE*KCTOJO
      RETURN
    ELSEIF (UNIT.EQ.'CA'.OR.UNIT.EQ.'ca') THEN
      VALUE=VALUE*CATOJO
      RETURN
    ELSEIF (UNIT.EQ.'KJ'.OR.UNIT.EQ.'kj') THEN
      VALUE=VALUE*KJTOJO
      RETURN
    ELSEIF (UNIT.EQ.'HA'.OR.UNIT.EQ.'ha') THEN
      VALUE=VALUE*HATOJO
      RETURN
    ELSEIF (UNIT.EQ.'ER'.OR.UNIT.EQ.'er') THEN
      VALUE=VALUE*ERTOJO
      RETURN
    ELSE
      1 WRITE(IOUT,*) 'Units input for energy cannot be con-
      2 verted. Only EV, CM, KC, CA, KJ, JO, HA, ER
      GO TO 100
    ENDIF
  ELSEIF (INTYPE.EQ.2) THEN
    IF (UNIT.EQ.'SE'.OR.UNIT.EQ.'se') THEN
      RETURN
    ELSEIF (UNIT.EQ.'MI'.OR.UNIT.EQ.'mi') THEN
      VALUE=VALUE*MITOSE
      RETURN
    ELSEIF (UNIT.EQ.'HR'.OR.UNIT.EQ.'hr') THEN
      VALUE=VALUE*HRTOSE
      RETURN
    ELSE
      1 WRITE(IOUT,*) 'Units for rate constant cannot be con-
      2 verted. Only SE, MI or HR allowed.'
      GO TO 100
    ENDIF
  ELSEIF (INTYPE.EQ.3) THEN

```

```

IF (UNIT.EQ. BO .OR. UNIT.EQ. be) THEN
  VALUE=VALUE*BOTOME
  RETURN
ELSEIF (UNIT.EQ. AG .OR. UNIT.EQ. ag) THEN
  VALUE=VALUE*ASTOME
  RETURN
ELSEIF (UNIT.EQ. IM .OR. UNIT.EQ. im) THEN
  VALUE=VALUE*IMTOME
  RETURN
ELSEIF (UNIT.EQ. CM .OR. UNIT.EQ. cm) THEN
  VALUE=VALUE*CMTOME
  RETURN
ELSEIF (UNIT.EQ. PM .OR. UNIT.EQ. pm) THEN
  VALUE=VALUE*PMTOME
  RETURN
ELSEIF (UNIT.EQ. ME .OR. UNIT.EQ. me) THEN
  RETURN
ELSE
  WRITE(IOUT,*) 'Units for distance cannot be converted.'
  1 GO TO 100
ENDIF
ELSEIF (ITYPE.EQ.4) THEN
  IF (UNIT.EQ. AU .OR. UNIT.EQ. au) THEN
    VALUE=VALUE*AUTOC
    RETURN
  ELSEIF (UNIT.EQ. KG .OR. UNIT.EQ. kg) THEN
    RETURN
  ELSEIF (UNIT.EQ. CM .OR. UNIT.EQ. cm) THEN
    VALUE=VALUE*CMTOCK
    RETURN
  ELSE
    WRITE(IOUT,*) 'Units for mass cannot be converted.'
    1 GO TO 100
  ENDIF
ELSEIF (ITYPE.EQ.5) THEN
  IF (UNIT.EQ. FA .OR. UNIT.EQ. fa) THEN
    VALUE=(VALUE*32.0)*5.0/9.0 * CETOVE
    RETURN
  ELSEIF (UNIT.EQ. CE .OR. UNIT.EQ. ce) THEN
    VALUE=VALUE * CETOVE
    RETURN
  ELSEIF (UNIT.EQ. KE .OR. UNIT.EQ. ke) THEN
    RETURN
  ELSE
    WRITE(IOUT,*) 'Units for temperature cannot be converted.'
    1 GO TO 100
  ENDIF
ELSEIF (ITYPE.EQ.6) THEN
  IF (UNIT.EQ. MO .OR. UNIT.EQ. mo) THEN
    RETURN
  ELSEIF (UNIT.EQ. NO .OR. UNIT.EQ. no) THEN
    VALUE=VALUE /AVOGAD
    RETURN
  ELSE
    WRITE(IOUT,*) 'Units for population cannot be converted.'
    1 GO TO 100
  ENDIF
ELSEIF (ITYPE.EQ.7) THEN
  IF (UNIT.EQ. PA .OR. UNIT.EQ. pa) THEN
    RETURN
  ELSEIF (UNIT.EQ. TO .OR. UNIT.EQ. to) THEN
    C **** conversion for torr to SI units?
    VALUE=VALUE*760PA
    RETURN
  ELSEIF (UNIT.EQ. AT .OR. UNIT.EQ. at) THEN
    VALUE=VALUE*ATTCPA
    RETURN
  ELSEIF (UNIT.EQ. KP .OR. UNIT.EQ. kp) THEN
    VALUE=VALUE*PRTCPA
    RETURN
  ELSE
    WRITE(IOUT,*) 'Units for pressure cannot be converted.'
    1 GO TO 100
    2 GO TO 100
  ENDIF
ELSEIF (ITYPE.EQ.8) THEN
  IF (UNIT.EQ. MO .OR. UNIT.EQ. mo) THEN
    RETURN
  ELSEIF (UNIT.EQ. MO .OR. UNIT.EQ. mo) THEN
    C **** additional concentration units??
    ELSE
    WRITE(IOUT,*) 'Units for concentration cannot be converted.'
    1 GO TO 100
  ENDIF
ELSE
  WRITE(IOUT,*) 'Internal error in SIU.'
  WRITE(IOUT,*) 'SI unit conversion not performed.'
  WRITE(IOUT,*) 'Type is inappropriate. ITYPE = ',ITYPE
  RETURN
ENDIF
C****
C****
100 WRITE(IOUT,*) 'Please respecify unit. ' UNIT
READ(INP,*) END=901) UNIT
GO TO 901
901 IF (IOUT.EQ. ISUP) THEN
  INP=5
  IOUT=6
  GO TO 100
ELSE
  C The defaults are hard to specify here, because such a variety
  C of units is available.
  CALL DFALTD(VALUE,1.0D0)
  IF (ITYPE.EQ.1) CALL DFALTC(UNIT, DE)
  IF (ITYPE.EQ.2) CALL DFALTC(UNIT, SE)
  IF (ITYPE.EQ.3) CALL DFALTC(UNIT, AG)
  IF (ITYPE.EQ.4) CALL DFALTC(UNIT, CM)
  IF (ITYPE.EQ.5) CALL DFALTC(UNIT, CE)
  IF (ITYPE.EQ.6) CALL DFALTC(UNIT, MO)
  IF (ITYPE.EQ.7) CALL DFALTC(UNIT, PA)

```

```

IF (ITYPE.EQ.9) CALL SFALTC(UNIT, MO)
ENDIF
902 WRITE(IGAVE,*) VALUE, UNIT
WRITE(IOUT,*) VALUE, UNIT
IFLAG=IFLAG+1
IF (IFLAG.EQ.1) GO TO 101
WRITE(IOUT,*) 'Error in respecifying. Program terminating. SSUB
STOP 1001
END

```

```

SUBROUTINE NOVICE (INTYPE)
  This subroutine prompts for units specified for the following
  types to SI MGS units according to the tens-digit of INTYPE:
  (1) energy
  (2) rate constants
  (3) distance inverse distance
  (4) mass
  (5) temperature
  (6) population
  (7) pressure
  (8) concentration

```

For example, CALL NOVICE(10) means that the first kind of unit (since int(12/10)=1) has been selected.

```

IMPLICIT DOUBLE PRECISION(A-H,J-Z)
IMPLICIT CHARACTER*(3)
INTEGER*4 UNIT, VONLY
COMMON/IO/INP, IOUT, ISAVE, ISUP, IDEBUG, INDAT, INGV, VONLY, VUNIT(10,21)
SSUB=SSUB+1) NOVICE)
ITYPE= (INTYPE/10)
IF (ITYPE.EQ.1) THEN

```

```

  WRITE(IOUT,*) 'Units which are acceptable for energy are:'
  WRITE(IOUT,*) 'EV or eV - electron volts'
  WRITE(IOUT,*) 'CM or cm - wavenumbers'
  WRITE(IOUT,*) 'KC or kc - kilocalories/mole'
  WRITE(IOUT,*) 'CA or ca - calories/mole'
  WRITE(IOUT,*) 'KJ or kj - kilojoules/mole'
  WRITE(IOUT,*) 'JO or jo - joules/mole'
  WRITE(IOUT,*) 'HA or ha - hartree (atomic units)'
  WRITE(IOUT,*) 'ER or er - erg/mole'
ELSEIF (ITYPE.EQ.2) THEN
  WRITE(IOUT,*) 'Units for rate constant allowed are:'
  WRITE(IOUT,*) 'SE or se - moles/sec'
  WRITE(IOUT,*) 'MI or mi - moles/minute'
  WRITE(IOUT,*) 'HR or hr - moles/hour'
ELSEIF (ITYPE.EQ.3) THEN
  WRITE(IOUT,*) 'Units for distance or inverse distance are:'
  WRITE(IOUT,*) 'ME or mE - meters'
  WRITE(IOUT,*) 'BO or bo - bohrs (atomic units)'
  WRITE(IOUT,*) 'AO or aO - angstroms'
  WRITE(IOUT,*) 'NM or nM - nanometers'
  WRITE(IOUT,*) 'CM or cM - centimeters'
  WRITE(IOUT,*) 'PM or pM - picometers'
ELSEIF (ITYPE.EQ.4) THEN
  WRITE(IOUT,*) 'Units for mass allowed are:'
  WRITE(IOUT,*) 'AU or aU - atomic mass units'
  WRITE(IOUT,*) 'KG or kg - kilograms'
  WRITE(IOUT,*) 'GM or gM - grams'
ELSEIF (ITYPE.EQ.5) THEN
  WRITE(IOUT,*) 'Units for temperature allowed are:'
  WRITE(IOUT,*) 'FA or fA - Fahrenheit'
  WRITE(IOUT,*) 'CE or ce - Celsius'
  WRITE(IOUT,*) 'KE or ke - Kelvin'
ELSEIF (ITYPE.EQ.6) THEN
  WRITE(IOUT,*) 'Units for population are:'
  WRITE(IOUT,*) 'MO or mo - moles'
  WRITE(IOUT,*) 'NO or no - number of molecules'
ELSEIF (ITYPE.EQ.7) THEN
  WRITE(IOUT,*) 'Units for pressure are:'
  WRITE(IOUT,*) 'PA or pA - pascal'
  WRITE(IOUT,*) 'TO or to - torr (cm Hg)'
  WRITE(IOUT,*) 'AT or at - atmosphere'
  WRITE(IOUT,*) 'KP or kP - kilopascal'
ELSEIF (ITYPE.EQ.8) THEN
  WRITE(IOUT,*) 'Units for concentration are:'
  WRITE(IOUT,*) 'MO or mo - molar'
ELSE
  WRITE(IOUT,*) 'Internal error.'
  WRITE(IOUT,*) 'Type is inappropriate. ITYPE = ',ITYPE
  CALL ITERM(ITYPE,ISUB)
  RETURN
ENDIF
WRITE(IOUT,*)
RETURN
END

```

```

SUBROUTINE READIN(IREAD,DATA)
  IMPLICIT DOUBLE PRECISION (B-H,K,N,O,W,Z)
  IMPLICIT INTEGER (A,I,J,L,M,V)
  IMPLICIT CHARACTER*(*)
  LOGICAL SYVAL(10,20)
  COMMON/IO/INP, IOUT, ISAVE, ISUP, IREPS, IINDAT, INOV, VONLY, VUNIT(0:23)
  COMMON/COUNT/MAXV, MAXK(0:10), MAXIN, IIPTR, ICOLL
  DIMENSION YVALUE(0:20), DATA(0:10,0:20)
  CHARACTER*2 UNIT
  CHARACTER*70 SHEADR

  C This subroutine is flagged, according to what kind of values
  C will be read in:
  C
  C IREAD          UNIT FOR READING IN FILE
  C 11 -- V EIV(J=0)      1
  C 21 -- V Ksp(V, 0)     2
  C 34 -- V Kbin(V, 0)    3
  C 61 -- V N(V, 0)       4
  C
  C Please see the general comment in the main program regarding
  C the IREAD numbering convention.
  C
  C Any data file consisting of triples (s.g. V,J, EIV(J)) is read in
  C through subroutine READWJ.
  C
  SSUB=SPRSUB(IREADIN)
  IFIT=0
  IFIRST=0
  C Files for read-in data are assigned to units as noted
  IF(IREAD.EQ.11) IINDAT=1
  IF(IREAD.EQ.21) IINDAT=2
  IF(IREAD.EQ.34) IINDAT=3
  IF(IREAD.EQ.61) IINDAT=4
  C Is MAXV already known? If not known, it will still be zero,
  C and within this subroutine, we will have to determine MAXV by a
  C standard comparison technique. IFIRST acts as a flag.
  IF(MAXV.EQ.0) IFIRST=1
  DO 100 J=0, 20
    YVALUE(J)=0.0D0
    SYVAL(J)=.FALSE.
  100 CONTINUE
  IF(IREAD.EQ.11) THEN
    WRITE(IOUT,*) 'What is the unit for energy in the file?'
    SHEADR=' ' <default: 'kJ' >
    IF(INOV.EQ.1) CALL NOVICE(IREAD)
    READ(INP,*,END=900) UNIT
    GO TO 901
  900 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 902
  ELSE
    CALL DFALTC(UNIT, 'E')
  ENDF
  WRITE(ISAVE,*) UNIT
  SHEADR=' ' & V & EIV
  901 ELSE IF(IREAD.EQ.21) THEN
    WRITE(IOUT,*) 'What is the unit for k(--->v=1) rate &
    & constants in the file?'
    SHEADR=' ' <Def.: 'se' >
    IF(INOV.EQ.1) CALL NOVICE(IREAD)
    READ(INP,*,END=904) UNIT
    GO TO 905
  904 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 903
  ELSE
    CALL DFALTC(UNIT, 'SE')
  ENDF
  WRITE(ISAVE,*) UNIT
  SHEADR=' ' & V & EIV & 'v=1'
  905 ELSE IF(IREAD.EQ.34) THEN
    WRITE(IOUT,*) 'What is the unit for kbin(v) in the file?'
    SHEADR=' ' <Def.: 'se' >
    IF(INOV.EQ.1) CALL NOVICE(IREAD)
    READ(INP,*,END=907) UNIT
    GO TO 908
  907 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 906
  ELSE
    CALL DFALTC(UNIT, 'SE')
  ENDF
  WRITE(ISAVE,*) UNIT
  SHEADR=' ' & V & Kbin(V)
  908 ELSE IF(IREAD.EQ.61) THEN
    WRITE(IOUT,*) 'What is the unit used for population?'
    SHEADR=' ' <Def.: 'mc' >
    IF(INOV.EQ.1) CALL NOVICE(IREAD)
    READ(INP,*,END=910) UNIT
    GO TO 911
  910 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 909
  ELSE
    CALL DFALTC(UNIT, 'mc')
  ENDF
  WRITE(ISAVE,*) UNIT
  SHEADR=' ' & V & N(V)
  911 ELSE
    CALL ISTERM(IREAD,SSUB)
  ENDF
  C As the data are read in, they are converted into s.i. units.
  C Data conversion in SIU depends on whether data are energy (IREAD/10
  C = 1) or rate constants(IREAD/10 = 2) or population (IREAD/10 = 6).
  C
  C The first value of each pair must by definition be an integer.

```

```

  C This loop will determine the size of the array and automatically
  C arrange the points in the array according to the value of v.
  C
  C The array SYVAL is intended simply to denote which locations in
  C the array are filled(.true.) and which are not (.false.).
  C
  VINDMAX=0
  DO 200 I=0, 20
    READ(INDAT,*,END=300) VIN, DATAIN
    IF(SYVAL(VIN)) THEN
      WRITE(6,*) Duplicate in array at location VIN.
      & Most recent value supplied will be used.
    ELSE
      SYVAL(VIN)=.TRUE.
    ENDF
    IF(VIN.GT.VINDMAX) VINDMAX=VIN
    CALL SIU(READ,DATAIN,UNIT)
    YVALUE(VIN) = DATAIN
  200 CONTINUE
  C There are IDATA x,y pairs in the
  C array, and the counters go from 0 to IDATA-1.
  300 WRITE(IOUT,*) I. data pairs read in.
  C If this is the first call to READIN (i.e. MAXV has not been
  C set beyond zero), then the max. V value read in will become
  C MAXV. Otherwise, check to see if the max. v read in
  C corresponds to the MAXV the user has previously specified.
  IF(IFIRST.EQ.1) THEN
    MAXV=VINDMAX
  ELSE IF(MAXV.GT.VINDMAX) THEN
    WRITE(IOUT,*) File read in does not go as high as max. v
    & value expected: MAXV,VINDMAX
  ELSE IF(MAXV.LT.VINDMAX) THEN
    WRITE(IOUT,*) File read in goes beyond the max. v
    & value expected: MAXV,VINDMAX
  ENDF
  C
  C Now, we must check to see if any v levels are missing, and then
  C interpolate to obtain the missing value by calling an
  C INSL subroutine, which is accessed through FITTR1.
  C (The '1' in the subroutine name refers to the one-dimensional
  C cubic spline function.)
  DO 500 I=0, MAXV
    IF(.NOT.SYVAL(I)) IFIT = 1
  500 CONTINUE
  C There are two types of one-dimensional fitting subroutines:
  C (1) a smoothing fitter ICSSCV
  C (2) an interpolating fitter ICSCCU
  C The routine called depends on how closely the user wants the
  C fitting function to 'stick' to the data.
  C The most accurate data should interpolate, more questionable
  C points may hav. an 'improved fit' by allowing the fit to smooth the
  C data. (To be decided by user.)
  IFLAG=0
  IF(IFIT.NE.0) THEN
    IF(IFIT.EQ.1) THEN
      WRITE(IOUT,*) Some points are missing from data file.
      & Would you like the data to be smoothed (1)
      & or interpolated (0)?
      READ(INP,*,END=931) ANSMO
      GO TO 932
    931 IF(IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 930
    ELSE
      CALL DFALTI(ANSMO, 0)
    ENDF
    WRITE(ISAVE,*) ANSMO
    CALL FITTR1(SYVAL, YVALUE, ANSMO)
  ELSE
    933 WRITE(IOUT,*) No points are missing from the data
    WRITE(IOUT,*) Would you like the data to be smoothed anyway?
    WRITE(IOUT,*) '1 = Yes; 0 = no'
    READ(INP,*,END=934) ANSMO
    GO TO 935
  934 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 933
  ELSE
    CALL DFALTI(ANSMO, 0)
  ENDF
  935 WRITE(ISAVE,*) ANSMO
  IF(ANSMO.EQ.1) CALL FITTR1(SYVAL, YVALUE, 1)
  ENDF
  WRITE(IOUT,*) SHEADR
  DO 600 I=0, MAXV
    DATA(I,0)=YVALUE(I)
    WRITE(IOUT,*) I, DATA(I,0)
  600 CONTINUE
  RETURN
  END

```

```

SUBROUTINE FITTR1(SYVAL,VVALUE,ANSMS)
  IMPLICIT DOUBLE PRECISION (D-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*) (C)
  LOGICAL*1 SYVAL(0:20)
  PARAMETER (IYSIZE=20)
  DIMENSION XIN(0:20),YIN(0:20),XOUT(0:20),YOUT(0:20),
  & VVALUE(0:20),SPLCOF(0:20,3),WF(3:108)
  C All we need to document calc for wk size.
  COMMON/IO/ IOUT,ISAVE,ISUP,ITERM,INRAT,INOV,MONLY,VUNIT(0:23)
  COMMON/COU/ MAXV,MAXJ(0:10),MAXIN,INPTS,ICOLL
  C There are two types of one-dimensional fitting subroutines:
  C (1) a smoothing filter ICSSCV
  C (2) an interpolating filter ICSCU
  C The routine called depends on how closely the user wants the
  C fitting function to "stick" to the data.
  SSUB = FITTR1
  ICOUNT=0
  IFIT=0
  ICSS=1
  IERR=0
  C Build the 'in' arrays which will be
  C used to create the spline coefficients.
  DO 100 I=0,IYSIZE
    IF(SYVAL(I)) THEN
      XIN(ICOUNT)=DBLE(I)
      YIN(ICOUNT)=VVALUE(I)
      ICOUNT=ICOUNT+1
    ENDIF
  ENDDO
100 CONTINUE
  IF(ICOUNT.GE.4.AND.ANSMS.EQ.1) THEN
    CALL ICSSCV(XIN,YIN,ICOUNT,YOUT,SPLCOF,IYSIZE-1,IERR)
  ELSE IF(ICOUNT.GE.3.AND.ANSMS.EQ.0) THEN
    CALL ICSCCV(XIN,YIN,ICOUNT,SPLCOF,IYSIZE-1,IERR)
  ELSE
    WRITE(IOUT,*) 'Too few points for fitting.'
  ENDIF
  IF(IERR.GT.128) CALL ITERM(IERR,SSUB)
  C Now use the spline coefficients to find the points needed.
  C (We will recalculate the y-values of even those points
  C whose input values are known, since this is easier than sorting
  C the leaves from the have-nots.)
  C For now, we evaluate the whole array. If this proves to be other
  C than harmless (extrapolation goes wild, for example) then the
  C 2nd-last parameter in the call to icsevu will have to be changed
  C to limit the extent of the extrapolation.
  CALL ICSEVU(XIN,YIN,ICOUNT,SPLCOF,IYSIZE-1,XOUT,VVALUE,IYSIZE-1
  & IERR)
  IF(IERR.GT.128) CALL ITERM(IERR,SSUB)
  RETURN
END

SUBROUTINE READIN(IPEAD, DATA)
  IMPLICIT DOUBLE PRECISION (D-H,K,N-O,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*) (C)
  LOGICAL*1 SYVAL(0:10,0:20)
  COMMON/IO/ IOUT,ISAVE,ISUP,ITERM,INRAT,INOV,MONLY,VUNIT(0:23)
  COMMON/COU/ MAXV,MAXJ(0:10),MAXIN,INPTS,ICOLL
  DIMENSION XVAL(1:10,0:20),DATA(0:10,0:20),JINMAX(0:10)
  CHARACTER*3 UNIT
  CHARACTER*70 SHEADS
  SSUB = READIN
  IPEAD = 0
  IPRINT = 0

  C This subroutine is similar to REACTIN, except that it reads in
  C a three-dimensional array.
  C Please see the general comment in the main program regarding
  C the IPEAD numbering convention.

  IPEAD UNIT FOR READINS IN FILE
  10 -- V J EIV J) 1
  20 -- V J MIN(V,J) 3
  30 -- V J NIV J) 4

  C Files for read-in data are assigned to units as noted
  IF(IPEAD.EQ.10) INRAT=1
  IF(IPEAD.EQ.15) INRAT=3
  IF(IPEAD.EQ.20) INRAT=4
  C Determine if MAXV has been set or if the current call to this
  C routine should set it AND the MAXJ array.
  IF (MAXV.EQ.0) IPRINT = 1

  C redimension here
  DO 100 V=1,10
    DO 100 J=0,20
      XVAL(V,J)=0.000
      XVAL(V,J) = .FALSE.
  100 CONTINUE
  IF(IPEAD.EQ.10) THEN
    WRITE(IOUT,*) 'What is the unit for energy in the file?'
    WRITE(IOUT,*) ' <default: "ke">'
    IF(INOV.EQ.1) CALL NOVICE(IPEAD)
    READ(INP,*) END=901: UNIT
    GO TO 902
  901 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 900
  ELSE
    CALL DPALTC(UNIT,*)
  ENDIF
  900 WRITE(IGAVE,*) UNIT
  SHEADS = ' V J EIV J)'
  ELSE IF(IPEAD.EQ.15) THEN
    WRITE(IOUT,*) 'What are
    * the units for the rate constants in the file?'
    WRITE(IOUT,*) ' <default: "no">'
    IF(INOV.EQ.1) CALL NOVICE(IPEAD)
    READ(INP,*) END=911: UNIT
    GO TO 912
  911 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 910
  ELSE
    CALL DPALTC(UNIT,*)
  ENDIF
  910 WRITE(IGAVE,*) UNIT
  SHEADS = ' V J MIN(V,J)'
  ELSE IF(IPEAD.EQ.20) THEN
    WRITE(IOUT,*) 'What are
    * the units for population in the file?'
    WRITE(IOUT,*) ' <default: "no">'
    IF(INOV.EQ.1) CALL NOVICE(IPEAD)
    READ(INP,*) END=921: UNIT
    GO TO 922
  921 IF(IOUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 920
  ELSE
    CALL DPALTC(UNIT,*)
  ENDIF
  920 WRITE(IGAVE,*) UNIT
  SHEADS = ' V J NIV J)'
  ELSE
    CALL ITERM(IPEAD,SSUB)
  ENDIF

  C As the data are read in, they are converted into S.I. units.
  C Date conversion in SIU depends on whether data are energy (IPEAD/10
  C = 1) or rate constants (IPEAD/10 = 2) or population (IPEAD/10 = 3).
  C The first two values of each triplet must by definition be integers.
  C This loop will determine the size of the array and automatically
  C arrange the points in the array according to the value of V and J.
  C The array XVAL is intended simply to denote which locations in
  C the array are filled (.true.) and which are not (.false.).
  VJMAX = 0
  DO 150 I = 0, 10
    JINMAX(I) = 0
  150 CONTINUE
  DO 200 I=0,10
    READ(INRAT,*) END=300: VIN, JIN, DATIN
    IF(SYVAL(VIN, JIN)) THEN
      WRITE(6,*) Duplicate in array at location (, VIN, , JIN
      * Most recent value supplied will be used.
    ELSE
      SYVAL(VIN, JIN) = .TRUE.
    ENDIF
  200 CONTINUE
  IF (VIN.GT.VJMAX) THEN

```

```

VINMAX = VIN
IF (JIN.GT.JINMAX(VIN)) JINMAX(VIN) = JIN
ENDIF
CALL SMLFREAD(DATIN,UNIT)
XVAL(VIN, JIN) = DATIN
CONTINUE
100 WRITE(ROUT,*) ' data template read in.
IF (IFIRST.EQ.1) THEN
    MAXV = VINMAX
    DO 350 I = 0, 10
        MAXJ(I) = JINMAX(I)
    CONTINUE
350 ELSE
    IF (MAXV.GT.VINMAX) WRITE(ROUT,*) ' Interpolation coming up.
    v values between MAXV and VINMAX are missing.
    IF (MAXV.LT.VINMAX) WRITE(ROUT,*) ' Warning...some input v
    values are beyond the maximum, MAXV
    for this run. They will be ignored.
    DO 350 IV = 0, MAXV
        DO 360 IJ = 0, MAXJ(IV)
            IF (MAXJ(IV).GT.JINMAX(IV)) WRITE(ROUT,*)
            ' Interpolation coming up. J values between MAXJ(IV)
            and JINMAX(IV) are missing.
            IF (MAXJ(IV).LT.JINMAX(IV)) WRITE(ROUT,*)
            ' Warning...some input J value are beyond the maximum
            for this run. They will be ignored.
            (MAXJ(IV), IJ) for this run. They will be ignored.
360 CONTINUE
ENDIF
CONTINUE
ENDIF
C
C Now we must check to see if any v or J levels are missing, and then
C interpolate to obtain the missing value by calling an
C IMSL subroutine, which is accessed through FITTR2.
C (The '2' in the subroutine name refers to the two-dimensional
C cubic spline function.)
DO 500 IV=0, MAXV
    DO 500 IJ=0, MAXJ(IV)
        IF (.NOT.SXVAL(IJ,IJ)) IFIT = 1
    CONTINUE
500 CONTINUE
C There are two types of two-dimensional fitting subroutines:
C (1) a smoothing filter ? analogous to ICSSCV
C (2) an interpolating filter ? analogous to ICOCOU
C The routine called depends on how closely the user wants the
C fitting function to "stick" to the data.
C The most accurate data should interpolate; more questionable
C points may have an improved fit by allowing the fit to smooth the
C data. (To be decided by user.)
IFLAG=0
IF (IFIT.NE.0) THEN
    WRITE(ROUT,*) ' Some points are missing from data file.
    WRITE(ROUT,*) ' Would you like the data to be smoothed (1)
    or interpolated (0) ?
    READ(INP,*,END=931) ANSMO
    GO TO 932
931 IF (ROUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 930
    ELSE
    CALL DFALTI(ANSMO,0)
    ENDF
932 WRITE(ISAVE,*) ANSMO
    CALL FITTR2(SXVAL,XVAL,ANSMO)
    ELSE
933 WRITE(ROUT,*) ' No points are missing from the data file.
    WRITE(ROUT,*) ' Would you like the data to be smoothed anyway?
    WRITE(ROUT,*) ' 1 = yes, 0 = no'
    READ(INP,*,END=934) ANSMO
    GO TO 935
934 IF (ROUT.EQ.ISUP) THEN
    INP=5
    IOUT=6
    GO TO 933
    ELSE
    CALL DFALTI(ANSMO,0)
    ENDF
935 WRITE(ISAVE,*) ANSMO
    IF (ANSMO.EQ.1) CALL FITTR2(SXVAL,XVAL,1)
    ENDF
    WRITE(ROUT,*) ' SHEADR
    DO 600 IV=0, MAXV
        DO 600 IJ=0, MAXJ(IV)
            DATA(IV,IJ)=SXVAL(IV,IJ)
            WRITE(ROUT,*) IV, IJ, DATA(IV,IJ)
600 CONTINUE
RETURN
END

```

```

SUBROUTINE FITTR2(SXVAL,XVAL,ANSO)
IMPLICIT DOUBLE PRECISION (D,B,H,F,N,U,W,C)
IMPLICIT INTEGER (A,I,J,L,M,V)
IMPLICIT CHARACTER*(*)
COMMON /1/ SXVAL(0:10,0:10)
PARAMETER (INSIDE=10, IYSIDE=20)
DIMENSION XIN(0:10), YIN(0:10), XINT(0:10), YINT(0:10)
* XIN(0:10) XINT(0:10) 0:10)
* YIN(0:10) YINT(0:10) 0:10)
* XVAL(0:10,0:10) IWK(3555) WF(6666)
C** document the IWK and WF array sizes.
COMMON /2/ INP, IOUT, ISAVE, ISUP, IZEROS, INCAT, INOV, VONLY, VUNIT(0:10)
ISUB = FITTR2
ICINP=0
C
C counter for no. of points with known values
ICINT=0
IERR=0
LENGTH=(INSIDE-1)*(IYSIDE-1)
C Build the in arrays, XIN, YIN and ZIN, which will be
C used to create the spline coefficients.
DO 100 I=0, INSIDE
    XIN(I)=DBLE(I)
    YIN(I)=DBLE(I)
    DO 100 J=0, IYSIDE
        IF (I.EQ.0) YINT(J)=DBLE(J)
        IF (SXVAL(I,J)) THEN
            XIN(ICINP)=DBLE(I)
            YIN(ICINP)=DBLE(J)
            ZIN(ICINP)=SXVAL(I,J)
            ICINP=ICINP+1
        ENDIF
    CONTINUE
    IYMAX=0
    DO 300 V=0, MAXV
        find max. of all the OMAX
        IF (MAXJ(V).GT.IYMAX) IYMAX=MAXJ(V)
    CONTINUE
300 CONTINUE
C
C We will recalculate the y-values of even those points
C whose input values are known, since this is easier than sorting
C the have's from the have-nots.
C For now we evaluate the whole array. If this proves to be other
C than harmless (extrapolation goes wild, for example) then we will
C have to limit the extent of the extrapolation.
IF (ICINP.GE.4) THEN
    CALL ICSSCV(XIN,YIN,ZIN,LENGTH,XINT,MAXV-1,YINT,IYMAX-1,
    * ZINT,INSIDE-1,IWK,WF,IWK)
    ELSE
    WRITE(ROUT,*) ' Too few points for fitting.'
    ENDF
C
C write out the matrix found by interpolation
WRITE(6,*)
WRITE(6,99)
FORMAT(1X, V=0, TX, V=1, TX, V=2, TX, V=3, TX, V=4, TX, V=5,
* TX, V=6, TX, V=7, TX, V=8, TX, V=9, TX, V=10)
WRITE(6,*)
DO 400 J=0, IYMAX
    WRITE(6,98) J, (XINT(I,J), I=0, MAXV)
    WRITE(6,98) J, (XVAL(L,J), L=0, MAXV)
99 FORMAT(1X, J= 13 2X 11E10.4)
WRITE(6,*)
400 CONTINUE
IF (IERR.GT.100) CALL ISTERM(IERR,ISUB)
RETURN
END

SUBROUTINE DFALTI(VAR,VALUE)
DOUBLE PRECISION VAR, VALUE
COMMON /3/ INP, IOUT, ISAVE, ISUP, IZEROS, INCAT, INOV, VONLY, VUNIT(0:10)
VAR = VALUE
CLOSE (INP)
RETURN
END

SUBROUTINE DFALTI(VAR,VALUE)
CHARACTER*(*) VAR, VALUE
COMMON /3/ INP, IOUT, ISAVE, ISUP, IZEROS, INCAT, INOV, VONLY, VUNIT(0:10)
VAR = VALUE
CLOSE (INP)
RETURN
END

SUBROUTINE DFALTI(VAR,VALUE)
INTEGER VAR, VALUE
COMMON /3/ INP, IOUT, ISAVE, ISUP, IZEROS, INCAT, INOV, VONLY, VUNIT(0:10)
VAR = VALUE
CLOSE (INP)
RETURN
END

```

```

SUBROUTINE ISTERM (ANS, INAME)
  IMPLICIT INTEGER (I)
  IMPLICIT CHARACTER*6 (S)
  C A program termination for incorrect integer input.
  ISUB=SPRSUB/ISTERM
  WRITE(6,*) 'Specification for value in subroutine ', INAME,
  4 ' is invalid. IANS'
  WRITE(6,*) 'Program terminating in ', INAME, ' .
  STOP 1001
  END

SUBROUTINE ASTEM (ANS, INAME)
  IMPLICIT CHARACTER*6 (S)
  C A program termination for incorrect real number input.
  DOUBLE PRECISION ANS
  ISUB=SPRSUB/ASTERM
  WRITE(6,*) 'Specification for value in subroutine ', INAME,
  4 ' is invalid. ANS'
  WRITE(6,*) 'Program terminating in ', INAME, ' .
  STOP 1701
  END

SUBROUTINE DUMP (IVLVL, T, NVFAC, NEND, XTENT)
  C It is not obvious that the sum of DNV(V, J) over the V and J
  C levels gives dN/dT. The following example shows that this is
  C the case:
  C Given a 3-level system where 42 molecules/sec leave the
  C first level, 4 to react, and 36 to go to the second level,
  C 25 molecules/sec leave the third level--10 react and 15 fall
  C back to the second level.
  C Therefore 15 molecules/sec enter the second level and
  C 10 from the third + 4 from the second=or
  C (42-53+25) must be leaving the system.
  C That is, DNV(1)+DNV(2)+DNV(3) = DNDT.
  IMPLICIT DOUBLE PRECISION (B-H, Y, N-U, W-Z)
  IMPLICIT INTEGER (A, I-J, L-M, V)
  IMPLICIT CHARACTER*6 (S)
  REAL SDUMMY
  COMMON/GEAR/DUMMY(48), SDUMMY(4), IDUMMY(38)
  COMMON/POP/NV(0:10, 0:20), NTOT, CONCA, CONCB, CONCC, JCOLL(3)
  COMMON/IO/INP, IOU, ISAVE, ISU, IEREG, INDT, INOV, VONLY, VUNIT(0:23)
  COMMON/CONF/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
  COMMON/MAST/MAV, MAM(0:10, 0:10)
  COMMON/UNDERF/IUNDER
  DIMENSION DNV(0:10, 0:20), NVFAC(31), DNVFAC(31)
  CALL INCT
  ISUB=SPRSUB/DUMP
  C Master will calculate the DNVFAC vector, given the NVFAC vector
  C CALL MASTER(IVLVL, T, NVFAC, DNVFAC)
  C now unpack the DNVFAC NVFAC vectors.
  IVLVL = 0
  DO 100 V = 0, MAXV
    DO 100 J = 0, MAXJ(V)
      IVLVL = IVLVL + 1
      DNV(V, J) = DNVFAC(IVLVL)
      NV(V, J) = NVFAC(IVLVL)
100 CONTINUE
  NTOT = 0.0D0
  DNDT = 0.0D0
  VVVV = 0.0D0
  DO 200 V = 0, MAXV
    NLUMP = 0.0D0
    DO 201 J = 0, MAXJ(V)
      NTOT = NV(V, J) + NTOT
      NLUMP = NV(V, J) + NLUMP
      VVVV = VVVV + NV(V, J) * NV(V, J)
  C ...lumped over J values
  701 WRITE(VUNIT(V), 701) V, J, T, NV(V, J), NLUMP
  701 FORMAT(2I5, 3(DX, D16.10))
  701 CONTINUE
  701 ENLUMP = 0
  DO 202 J = 0, MAXJ(V)
    DNDT = DNV(V, J) + DNDT
    ENLUMP = DNV(V, J) + ENLUMP
  202 WRITE(VUNIT(V+12), 701) V, J, T, DNV(V, J), ENLUMP
  202 CONTINUE
  VVVV = VVVV/NTOT
  NVA = CONCA - CONCBC + NTOT
  KPHEN = -DNDT/NTOT/NVA
  WRITE(6,*) '(P, 7) (X, D16.3), 0P: (VVVV, KPHEN, DNV(0, 0), DNV(1, 0),
  1 DNV(2, 0), DNV(3, 0), DNV(4, 0))
  C xtent is amount reacted/amount available to react...
  XTENT = (CONCBC-NTOT)/NEND
  WRITE(21, 702) XTENT, KPHEN, DUMMY(8), IDUMMY(6),
  1 IDUMMY(7), IUNDER
  702 FORMAT(1P, 3(D14.7), 3(I10))
  WRITE(22, 703) NV(0, 0)/NTOT, NV(1, 0)/NTOT,
  1 NV(2, 0)/NTOT, NV(3, 0)/NTOT, NV(4, 0)/NTOT
  703 FORMAT(1P, D20.13, 4(D15.8))
  IF(IRESET.EQ.1) IRESET = 2
  IF(XTENT.GT.1.E-1.AND.IRESET.EQ.0) THEN
    IRESET = 1
    WRITE(6,*) 'Resetting to...'
    ERSEP
  IF(IRESET.EQ.2) IRESET = 4
  IF(XTENT.GT.1.E-1.AND.IRESET.EQ.2) THEN
    IRESET = 3
    WRITE(6,*) 'Resetting to...'
    ERSEP
  IF(IRESET.EQ.3) IRESET = 5
  IF(XTENT.GT.1.E-3.AND.IRESET.EQ.4) THEN
    IRESET = 5
    WRITE(6,*) 'Resetting to...'
    ERSEP
  WRITE(6,*) 'Time, Total Population: ', T, NTOT
  RETURN
  END

```

```

SUBROUTINE INIT
  C This subprogram initializes the physical constants used.
  C The constants are given for SI units.
  C NO DOCUMENT default changeability.
  IMPLICIT DOUBLE PRECISION (B-H, Y, N-U, W-Z)
  IMPLICIT INTEGER (A, I-J, L-M, V)
  IMPLICIT CHARACTER*6 (S)
  REAL*8 ARAD, AMASS
  DOUBLE PRECISION AVOGAD
  CHARACTER*36 SPRSUB(3)
  COMMON/STRES/STRESS, RESULTS, RMS, SLEN, PLINT, PI, IAVGAD
  COMMON/INVT/INVT, ISAVE, ISU, IEREG, INDT, INOV, VONLY, VUNIT(0:23)
  COMMON/CONF/MAXV, MAXJ(0:10), MAXUN, INPTS, ICOLL
  COMMON/POP/NV(0:10, 0:20), NTOT, CONCA, CONCB, CONCC, JCOLL(3)
  COMMON/ENERGY/ELEV(0:10, 0:10), TEMP
  COMMON/MASS/MAV, MAM(0:10, 0:10)
  PARAMETER (AVOGAD=6.022E23)
  EQUIVALENCE(ISTRES(1), IEREG(1))
  C The long strings which are specific for the various kinds of
  C collision are written into a '36 character string which is
  C then effectively broken into 6 = '6 character strings by means
  C of underlining with an array SLEN of dimension 6.
  C The string-length is chosen because CHARACTER*6 is conventionally
  C used in other subroutines.
  ERSEP is called because underflow errors may occur in the program.
  ERSEP is an IBM system program. Other adaptations of this
  program to different systems will require a call to a different
  underflow-adjustment system routine. If no such subroutine is
  available, the user may have to modify the program.
  Potential trouble spots are:
  **Chris Vic will determine this by running the pgm without
  ** ERSEP being called.
  CALL ERSEP (208, 0, -1, 1, 1)
  MAXUN = 11
  INP = 5
  IEREG = 1
  C ...debug toggled on for the time being
  IOU = 6
  C IOU and IEREG must be set for SPRSUB to work
  ISUB=SPRSUB/INIT
  C If this program has been run once, the answers will be saved
  C on unit ISAVE so that a restart option is available.
  C To avoid meaningless prompting on the terminal, the
  C output will be suppressed by writing it to unit ISUP.
  ISUP=99
  ISUB=98
  IFLAG=0
  WRITE(IOU,*) ' Defaults are explicitly noted, or are
  4 ' indicated by a "*" on the menus.'
  WRITE(IOU,*) ' Defaults may be obtained by entering only a
  4 ' carriage return after the prompt.'
  WRITE(IOU,*)
  904 CONTINUE
  904 WRITE(IOU,*) ' Is this a restart? (*Yes = 1, No = 0):
  ANRES = 1
  GO TO 913
  C P11 CALL REPTI(ANRES, 1)
  911 IF (ANRES.EQ.0.GE. ANRES.GT.1) THEN
    WRITE(IOU,*) ' Error in restart specification: *ANRES
    4 ' Choose 0 or 1
    IFLAG=IFLAG+1
    IF (IFLAG.EQ.1) GO TO 904
    CALL ISTERM(ANRES, SPRUB)
  ENDP
  IF (ANRES.EQ.1) THEN
    INP=5
  C ...to avoid colliding with filesets for v-levels
  IOU=ISUP
  ENDP
  PLANCH=6.63618D-34
  KBOLTZ=1.38065D-23
  KBOLTZ=1.38065D-23
  K obtained from Rev. Mod. Phys. vol. 27, p. 363 (1955).
  HBAR=KBOLTZ * AVOGAD
  CLIGHT=2.997925D+08
  KEVEP=1000.0000
  CEXPL=1.0D0
  PI=3.141592653589793
  EMASS=0.0D0
  CPASS=0.0D0
  AMASS=0.0D0
  REMAS=0.0D0
  BRAD=0.0D0
  CRAD=0.0D0
  ARAD=0.0D0
  KE=0.0D0
  C The maximum number of output units is set by MAXUN. For
  C v-levels ranging from 0 to 10, this means 11 output files
  C each of which will contain population information for
  C appropriate J sub-levels. The rate-of-population-change
  C information is contained on 11 more output files.
  C 2 additional files will monitor total population
  C and total rate-of-population-change.
  NV data for V=0...10 unit 10...20 VUNIT index 6-10
  IN data summed (NTOT) 21 11
  DNV data for V=0...10 units 40...50 13-22
  DNDT data unit 51 23
  MAKV=0
  DO 20 V=0,10
    MAXJ(V)=0
    VUNIT(V)=10+V
    VUNIT(V+12)=40+V
  20 CONTINUE
  C The matrix ELEV must have one extra level, so that the

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C      energy difference can be calculated in EREL for a given level.
      DO 13 V = 0 10
      DO 13 J = 0 10
      ELEV(V,0)=0.0
10  CONTINUE
C  MV  initialized Value unimportant normalized later.
      MV(0,0) = 1.0

      NOTES ON DEFAULTS:

      The default value is either explicitly denoted as such
      or it is indicated with a * in front of the menu number.

      If the input is from the terminal a carriage return will
      signify that the default value is desired.
      If the input is from a file then reset the input unit
      to that of the terminal so that the user can pick up
      from where s/he left off...

      The following default subroutines are called:
      DFAULT - for double precision numbers
      DFAULT - for integer numbers
      DFAULT - for character input.

      If the user wishes to change the default value only the
      prompting default reminder and the second argument in the
      DFAULT call must be changed.

900 WRITE(IOUT,*) 'Type a *1' if prompting for units is
      * required. Any other number--no prompting. <default: 0>
      (The routine for prompting for units is NOVICE.)
      READ(INP,*,END=901) INOV
      GO TO 900
901 IF(IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 900
      ELSE
      CALL DFAULT(INOV,0)
      ENDF
902 WRITE(ISAVE,*) INOV
      JCOLL(1)=6
      JCOLL(2)=9
      JCOLL(3)=0
      SSPEN(1)=* For collisions between A and BC;
      SSPEN(2)=* For collisions between BC and BC;
      SSPEN(3)=* For collisions between M and BC;
      IF(INOV.NE.1) RETURN
      WRITE(IOUT,*) 'This program accepts many kinds of units.
      WRITE(IOUT,*) 'The user must know which two-letter codes refer
      WRITE(IOUT,*) 'to which units, e.g., EV means electron volts.
      WRITE(IOUT,*) 'The letter code for inverse units is identical to
      WRITE(IOUT,*) 'that of the non-inverted units. The user must
      WRITE(IOUT,*) 'input values and units with the units enclosed
      1 ' in single quotes.
      2 ' and separated by commas e.g. '4.0 cm''.
      WRITE(IOUT,*)
      RETURN
      END

```

```

SUBROUTINE NOISCL(ANN01)
C  In the calculation of the rate coefficient for v<-->v+1 scaling,
C  the value from which all other values are scaled must first be
C  obtained. This routine prompts for K(0,0->1,0)
C  and if optional levels exist (i.e., vonly=0) it prompts
C  for K(0,0->1,1) and K(0,1->1,0).
C  When ANN01 is 1 Landau-Teller scaling is desired.
C  If the argument ANN01 is 0 use an alternate form of scaling.
      IMPLICIT DOUBLE PRECISION (B-H,F,N-U,W-Z)
      IMPLICIT INTEGER (A,I,O,L-M,V)
      IMPLICIT CHARACTER*(3)
      CHARACTER** UNIT
      COMMON/SPACES/ISPEC (SPECID,3)
      COMMON/IMP/IMPUP(0:10,0:10) IMPD(0:10,0:10) IMPD(0:10,0:10)
      COMMON/IMP/IMPDOWN(0:10,0:10) IMPDOWN(0:10,0:10)
      COMMON/IMP/IMPDOWN(0:10,0:10) IMPDOWN(0:10,0:10)
      COMMON/IMP/IMPDOWN(0:10,0:10) IMPDOWN(0:10,0:10)
      COMMON/IMP/IMPDOWN(0:10,0:10) IMPDOWN(0:10,0:10)
      COMMON/ENERGY/ELEV(0:10,0:10) TEM
      COMMON/PHYSCT/CLANCF KCOLTE RGAS SDEKPI CLIGHT PI INEVEN
      COMMON/COUNT/MAXV,MAXJ(0:10) MAXUN INPTS IICLL
      SUBSPREPUB/NOISCL
      WRITE(IOUT,*) 'ISPEC/ID ISPEC: 1 1 1'
901 FORMAT(1X,6A6)
902 WRITE(IOUT,*) 'Input the units to be used for v<-->v+1
      * rate constants: <default: '5'>'
      IF(INOV.EQ.1) CALL NOVICE(00)
      READ(INP,*,END=901) UNIT
      GO TO 902
901 IF(IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 900
      ELSE
      CALL DFAULT(UNIT,5)
      ENDF
902 WRITE(ISAVE,*) UNIT
      ... dummy call to SIU checks for valid unit
      CALL SIU(0,0,00) UNIT
C  Originally we prompted for K01 but this turned out to be inconvenient
C  (the literature usually cites K10) so now we prompt for K10 and
C  immediately convert to K01 so that the remainder of the code is not
C  affected. Users may wish to prompt for K01...
903 WRITE(IOUT,*) 'Input K for v=1,J=0 -> v=0,J=0 transition:
      WRITE(IOUT,*) <default: 1.44E7>
      READ(INP,*,END=904) K1000
      K1000 = GETV(0,ISPEC-1)
      GO TO 905
904 IF(IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 903
      ELSE
      CALL DFAULT(K1000,1.44E7)
      ENDF
905 WRITE(ISAVE,*) K1000
      CONST=1.0/PIAS*TEMP
      KUP(0,0) = K1000*DEXP(EDEL(0,0,1,0)*CONST)
      IF (IDEVEL.EQ.1) WRITE(OUT,*) KUP(0,0) KUP(0,0)
      CALL SIU(00,KUP(0,0) UNIT)
      IF (VONLY.EQ.0) THEN
906 WRITE(IOUT,*) 'Input K for v=1,J=0 -> v=0,J=0 transition:
      WRITE(IOUT,*) <default: 1.8D7>
      READ(INP,*,END=907) K1000
      GO TO 908
907 IF(IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 906
      ELSE
      CALL DFAULT(K1000,1.8D7)
      ENDF
908 WRITE(ISAVE,*) K1000
      KUP(0,0) = K1000*DEXP(EDEL(0,0,1,1)*CONST)
      CALL SIU(00,KUP(0,0) UNIT)
909 WRITE(IOUT,*) 'Input K for v=1,J=0 -> v=0,J=1 transition:
      WRITE(IOUT,*) <default: 1.2D7>
      READ(INP,*,END=910) K1000
      GO TO 911
910 IF(IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 909
      ELSE
      CALL DFAULT(K1000,1.2D7)
      ENDF
911 WRITE(ISAVE,*) K1000
      KUP(0,0) = K1000*DEXP(EDEL(0,0,2,1)*CONST)
      CALL SIU(00,KUP(0,0) UNIT)
      Must find KUP(0 all J) and KUP(0 all J) and KUP(0 all J)
      --this will be done using a modified KACTUP routine.
      CALL KACT
      ENDF
      IF(ANN01.EQ.1) THEN
      CALL KACT
      ELSEIF(ANN01.EQ.0) THEN
      CALL KACT
      ELSE
      CALL ISTERM(ANN01) $SUB)
      ENDF
      RETURN
      END

```

```

SUBROUTINE TEMPER
IMPLICIT DOUBLE PRECISION(A-H,C-O)
IMPLICIT INTEGER (I)
IMPLICIT CHARACTER*6 (S)
CHARACTER*2 UNIT1
INTEGER UNIT1, VONLY
COMMON/IO/INP,ICUT,ISAVE,ISUP,ISUP,ISPEC,INSTAT,INOV,VONLY,UNIT1(2)
COMMON/ENERGY/ELEV(0:10,0:20) TEMP
ISUB=SPRSPR*ICUTP
115 WRITE(OUT,*) 'Input the temperature. <Default: 300 K>'
IF(INOV.EQ.1) CALL NOVICE(50)
READ(INP,*) END=903) TEMP,UNIT1
TEMP = GETTMP(I)
GO TO 904
903 IF(OUT.EQ.ISUP) THEN
INP=5
ICUT=6
GO TO 115
ELSE
CALL DFALD(TEMP,300,IND)
CALL SFALD(UNIT1, 'K')
ENDIF
904 WRITE(ISAVE,*) TEMP,UNIT1
CALL STU(50,TEMP,UNIT1)
RETURN
END
    
```

```

SUBROUTINE CONCEN
IMPLICIT DOUBLE PRECISION(A-H,C-O)
IMPLICIT INTEGER (I)
IMPLICIT CHARACTER*6 (S)
CHARACTER*2 UNIT1
INTEGER UNIT1, VONLY, JCOLL
COMMON/IO/INP,ICUT,ISAVE,ISUP,ISUP,ISPEC,INSTAT,INOV,VONLY,UNIT1(2)
COMMON/POP/AV(0:10,0:20),NTOT,CONC(3),JCOLL(3)
ISUB=SPRSPR*CONCEN
110 WRITE(OUT,*) 'What is the unit of concentration used?'
WRITE(OUT,*) '<Default: mo/mole>'
IF(INOV.EQ.1) CALL NOVICE(50)
READ(INP,*) END=903) UNIT1
GO TO 901
900 IF(OUT.EQ.ISUP) THEN
INP=5
ICUT=6
GO TO 110
ELSE
CALL DFALD(UNIT1, 'M')
ENDIF
901 WRITE(ISAVE,*) UNIT1
! Dummy call to STU, just to check if units for concentration
! are kosher.
CALL STU(50,1.0D0,UNIT1)
111 WRITE(OUT,*) 'What is the concentration of species A?'
WRITE(OUT,*) '<Default: 1.0D-6>'
READ(INP,*) END=903) CONC1
GO TO 903
902 IF(OUT.EQ.ISUP) THEN
INP=5
ICUT=6
111a What happened to IFLAG...?
GO TO 111
ELSE
CALL DFALD(CONC1,1.0D-6)
ENDIF
903 WRITE(ISAVE,*) CONC1
CALL STU(50,CONC1,UNIT1)
112 WRITE(OUT,*) 'What is the concentration of species B?'
WRITE(OUT,*) '<Default: 1.0D-3>'
READ(INP,*) END=904) CONC2
GO TO 905
904 IF(OUT.EQ.ISUP) THEN
INP=5
ICUT=6
GO TO 112
ELSE
CALL DFALD(CONC2,1.0D-3)
ENDIF
905 WRITE(ISAVE,*) CONC2
CALL STU(50,CONC2,UNIT1)
113 WRITE(OUT,*) 'What is the concentration of relaxant gas?'
WRITE(OUT,*) '<Default: 1.0D-1>'
READ(INP,*) END=906) CONC3
GO TO 907
906 IF(OUT.EQ.ISUP) THEN
INP=5
ICUT=6
GO TO 113
ELSE
CALL DFALD(CONC3,1.0D-1)
ENDIF
907 WRITE(ISAVE,*) CONC3
CALL STU(50,CONC3,UNIT1)
! The three gases present in the default concentrations give a
! combined pressure of about 60 torr.
RETURN
END
PROGRAM BS10
IMPLICIT DOUBLE PRECISION (A-H,K,N,U,W-O)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/IO/INP,ICUT,ISAVE,ISUP,ISUP,ISPEC,INSTAT,INOV,VONLY,UNIT1(2)
COMMON/ENERGY/ELEV(0:10,0:20) TEMP
COMMON/COUNT/MAXV,MAXJ(0:10),MAXN,INPTS,ICOLL
COMMON/SPECIES/ISPEC,ISPEC(6,3)
COMMON/POP/AV(0:10,0:20),NTOT,CONC(3),JCOLL(3)
PRINT*, 'Version 664 of BMSIM, 10 May 1990'
PRINT*
CALL D050
STOP 1006
END
    
```

This program simulates the kinetics of bimolecular reactions.  
 Rotational levels included.  
 Variables:  
 The assumptions made for the allowed  $v, j \rightarrow v', j'$  transitions are: a change of 0 or 1 or -1 in the vibrational level; and a change of 0 or 2 or -2 in the rotational levels.  
 KUPUP, KUPD, KUPV --  
 These two-dimensional arrays contain the energy transfer rate constants in the 'up' direction. The convention is as follows: KUPUP(v,j) => the transition  $v,j \rightarrow v+1,j+1$   
 KUPD(v,j) => the transition  $v,j \rightarrow v+1,j-1$   
 KUPV(v,j) => the transition  $v,j \rightarrow v+1,j$   
 KDOWNU, KDOWND, KDOWNV --  
 Similarly, these two-dimensional arrays contain the energy transfer rate constants in the 'down' direction. The convention is as follows:  
 KDOWNU(v,j) => the transition  $v,j \rightarrow v-1,j+1$   
 KDOWND(v,j) => the transition  $v,j \rightarrow v-1,j-1$   
 KDOWNV(v,j) => the transition  $v,j \rightarrow v-1,j$   
 The  $J \rightarrow J-2$  transition rate coefficients can be determined by microscopic reversibility (along with  $v \rightarrow v-1$  rates).  
 The following relationships exist through microscopic reversibility:  
 KUPUP(v,j) <----> KDOWND(v,j)  
 KUPD(v,j) <----> KDOWNU(v,j)

```

SUBROUTINE BIMSINITOL INIT RESET TFIRST TROUT IEP:
IMPLICIT DOUBLE PRECISION (D,H,F,N,W)
IMPLICIT INTEGER (A,I,J,L,M,V)
IMPLICIT CHARACTER*(*)
COMMON/INIT/ IOUT ISAVE ISUP ISEBEG INEAT INOV INONLY MONITR(10)
COMMON/CONST/ MAX(10,10) MAXM(10) INFW(10) ICOLL
COMMON/ENERGY/ELEV(0:10,0:10) TEMP
COMMON/SPECIES/ISPEC(6,3)
COMMON/POP/IN(1:10,0:10) NTOT CONCN(1) JCOLL(3)
... For concentrations CONCA CONCB CONCM respectively

IOUT and ISEBEG must have values when SPRES is called but
their values
are assigned proper in subroutine INIT.
ISEBEG = 0
IOUT = 2
ISUP ISEBEG BIMSIN

PRINT* 'Version 064 of BIMSIN 30 May 1990'
PRINT*
CALL INIT
CALL ENERGY
IF(INV.EQ.1) CALL ASSUME
CALL TROUT
The temperature should be known first, in case the user specifies
partial pressure in the CONC subroutine.
CALL CONC
CALL PIRER(ANNU)
For each species in the reaction, the rate coefficients
must be found.

The vibrational structure of diatomic BC is of interest
but not that of anything such as the relaxant gas M which
collide with BC, and thus PIRER, where the potential energy
curve is specified, is not within the DO 100 loop.
CALL PIRER(ANNU)

Now that the ELEV array for BC has been calculated and filled in
from (0:10) to (maxv,max): we can consider the effect of collisions
on BC.
ICOLL=0
DO 100 ISPEC=1,3
...the rate coefficients depend on what SPECIES
collide with BC: A BC itself or M.
901 WRITE(IOUT,*)
WRITE(IOUT,*) '*****'
WRITE(IOUT,*)
WRITE(IOUT,*) '(SPECIES ISPEC = 1:1 6)'
902 PIRER(1,ISPEC)
WRITE(IOUT,*) 'Type 1 to continue with these collision partners.'
WRITE(IOUT,*) 'Otherwise, type another number.' default = 1
READIN * END=902) ANCOLL
GO TO 903
903 IF(IOUT.EQ.ISEBEG) THEN
...
CALL DFALT/ANCOLL 1)
ENDIF
WRITE(ISAVE,*) ANCOLL
IF(ANCOLL.NE.1) THEN
GO TO 100
ELSEIF(IOUT.EQ.ISEBEG) THEN
READIN *
READIN *
GO TO 100
ELSE
ICOLL=ICOLL+1
JCOLL/ICOLL=ISPEC
The values of the array JCOLL are set up according to the following
collision partners with BC:
1 - A + BC
2 - BC + BC
3 - Relaxant M + BC
e.g. if we only want to consider A + BC and M + BC cases
then JCOLL(1) = 1 and JCOLL(3) = 3.
The JCOLL array is initialized to zero so that the total number of
collision types may be determined by finding the position of
the last non-zero value.
ENDIF
Calculation of KOUF vector for each v level...
where KOUF(v) means KIV(v)-KIV(v-1)
only called if ro-vib. solution wanted)
IF(VONLY.EQ.0) CALL PIRER(ANKGUP)
The J--v-1 transition rate coefficients can be determined
by microscopic reversibility (along with v--v-1 rates).
IF(ANNU.NE.1) THEN
ANNU=1 means that K01 values are determined
CALL PIRER(ANK01)
ELSE
CALL K01SC(1)
... arg. 1 means Landau-Teller scaling to be done
ENDIF
CALL CHECKR
CALL CHECKR
MICREV must be called before COLLECT:
CALL MICREV
CALL COLLECT
100 CONTINUE
CALL PIRER(ANNU)
After the user has specified the method of obtaining the
microscopic rate coefficient in this subroutine, and the KRXIV(J)
are found in the various subroutines PIRER branches to
then the KRX should be examined to see which values are
not responsible for the system of interest.
For a bimolecular reaction this would imply that those KRX
which exceed the collision frequency possible, according

```

```

SUBROUTINE BIMSINITOL INIT RESET TFIRST TROUT IEP:
IMPLICIT DOUBLE PRECISION (D,H,F,N,W)
IMPLICIT INTEGER (A,I,J,L,M,V)
IMPLICIT CHARACTER*(*)
COMMON/INIT/ IOUT ISAVE ISUP ISEBEG INEAT INOV INONLY MONITR(10)
COMMON/CONST/ MAX(10,10) MAXM(10) INFW(10) ICOLL
COMMON/ENERGY/ELEV(0:10,0:10) TEMP
COMMON/SPECIES/ISPEC(6,3)
COMMON/POP/IN(1:10,0:10) NTOT CONCN(1) JCOLL(3)
... For concentrations CONCA CONCB CONCM respectively

IOUT and ISEBEG must have values when SPRES is called but
their values
are assigned proper in subroutine INIT.
ISEBEG = 0
IOUT = 2
ISUP ISEBEG BIMSIN

PRINT* 'Version 064 of BIMSIN 30 May 1990'
PRINT*
CALL INIT
CALL ENERGY
IF(INV.EQ.1) CALL ASSUME
CALL TROUT
The temperature should be known first, in case the user specifies
partial pressure in the CONC subroutine.
CALL CONC
CALL PIRER(ANNU)
For each species in the reaction, the rate coefficients
must be found.

The vibrational structure of diatomic BC is of interest
but not that of anything such as the relaxant gas M which
collide with BC, and thus PIRER, where the potential energy
curve is specified, is not within the DO 100 loop.
CALL PIRER(ANNU)

Now that the ELEV array for BC has been calculated and filled in
from (0:10) to (maxv,max): we can consider the effect of collisions
on BC.
ICOLL=0
DO 100 ISPEC=1,3
...the rate coefficients depend on what SPECIES
collide with BC: A BC itself or M.
901 WRITE(IOUT,*)
WRITE(IOUT,*) '*****'
WRITE(IOUT,*)
WRITE(IOUT,*) '(SPECIES ISPEC = 1:1 6)'
902 PIRER(1,ISPEC)
WRITE(IOUT,*) 'Type 1 to continue with these collision partners.'
WRITE(IOUT,*) 'Otherwise, type another number.' default = 1
READIN * END=902) ANCOLL
GO TO 903
903 IF(IOUT.EQ.ISEBEG) THEN
...
CALL DFALT/ANCOLL 1)
ENDIF
WRITE(ISAVE,*) ANCOLL
IF(ANCOLL.NE.1) THEN
GO TO 100
ELSEIF(IOUT.EQ.ISEBEG) THEN
READIN *
READIN *
GO TO 100
ELSE
ICOLL=ICOLL+1
JCOLL/ICOLL=ISPEC
The values of the array JCOLL are set up according to the following
collision partners with BC:
1 - A + BC
2 - BC + BC
3 - Relaxant M + BC
e.g. if we only want to consider A + BC and M + BC cases
then JCOLL(1) = 1 and JCOLL(3) = 3.
The JCOLL array is initialized to zero so that the total number of
collision types may be determined by finding the position of
the last non-zero value.
ENDIF
Calculation of KOUF vector for each v level...
where KOUF(v) means KIV(v)-KIV(v-1)
only called if ro-vib. solution wanted)
IF(VONLY.EQ.0) CALL PIRER(ANKGUP)
The J--v-1 transition rate coefficients can be determined
by microscopic reversibility (along with v--v-1 rates).
IF(ANNU.NE.1) THEN
ANNU=1 means that K01 values are determined
CALL PIRER(ANK01)
ELSE
CALL K01SC(1)
... arg. 1 means Landau-Teller scaling to be done
ENDIF
CALL CHECKR
CALL CHECKR
MICREV must be called before COLLECT:
CALL MICREV
CALL COLLECT
100 CONTINUE
CALL PIRER(ANNU)
After the user has specified the method of obtaining the
microscopic rate coefficient in this subroutine, and the KRXIV(J)
are found in the various subroutines PIRER branches to
then the KRX should be examined to see which values are
not responsible for the system of interest.
For a bimolecular reaction this would imply that those KRX
which exceed the collision frequency possible, according

```

```

C to the temperature and pressure of the system.
C Thus, the subroutine DIFFUS is called.
C In this subroutine, the collision frequency is calculated
C and any krxn(V,J) which exceeds it is set equal to it.
C If the user is modelling a unimolecular reaction, or an ideal
C reaction which is not limited by the rate of diffusion, then
C the upper limit in the subroutine DIFFUS should be specified.
      IF (ANMETH.NE.3) THEN
        CALL PKKPOP(ANPOP)
        CALL DIFFUS
        CALL SOLVER(TOL,HINIT,RESET,TFIRST,TMOUT,IEP)
        RETURN
      ENDIF
      IF (ANMETH.NE.1) THEN
        CALL FENIT(ICTR1)
        CALL ELGEN(ICTR1)
      ENDIF
907 WRITE(IOUT,*) 'Type 1 if you want tabulated output. <Default: 1>'
      READ(INP,*,END=905) ANTAB
      GO TO 908
905 IF (ICOLL.EQ.ISUP) THEN
      INP=5
      IOUT=4
      GO TO 907
    ELSE
      CALL DFALTI(ANTAB,1)
    ENDIF
906 WRITE(ISAVE,*) ANTAB
      IF (ANTAB.EQ.1) CALL TABOUT
908 WRITE(IOUT,*) 'Type 1 if you want output plotted <Def.: 1>'
      READ(INP,*,END=909) ANPLOT
      GO TO 910
909 IF (IOUT.EQ.ISUP) THEN
      INP=5
      IOUT=6
      GO TO 908
    ELSE
      CALL DFALTI(ANPLOT,0)
    ENDIF
910 WRITE(ISAVE,*) ANPLOT
      IF (ANPLOT.EQ.1) CALL PLTOUT
      RETURN
      END

SUBROUTINE DO30
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
COMMON/BSDATA/HUSED(500),IQUSED(500),ISTEP(500),IUNDR(500)
1Y(500),Y1(500),Y2(500),Y3(500),Y4(500)
COMMON/POP/NV(0:10,0:20),NTOT,CONCA,CONCBC,CONCM,JCOLL(3)
COMMON/UNDER/IUNDR
EXTERNAL TRAKUF
CALL ERRSET (108,0,-1,1,TRAKUF,207)
CALL TBLTOP
OPEN(4,FORM='UNFORMATTED')
INNER = 0
10 READ(1,*,END=20) KHTMP,CONCA,CONCBC,CONCM,TMP
1 ICOLL,K10(1),K10(2),TOL,HINIT,RESET,TFIRST,TMOUT
SCALEC = 1.D0
CONCA = CONCA*SCALEC
CONCBC = CONCBC*SCALEC
CONCM = CONCM*SCALEC
IF (ICOLL.EQ.1) THEN
  X = CONCA/CONCBC
ELSE
  X = CONCA/CONCM
ENDIF
DO 30 M=0,99
  IF (M.EQ.1.OR.M.EQ.2.OR.M.EQ.4.OR.M.EQ.5) GOTO 30
  REWIND(M)
  CONTINUE
  LOOPCT = 0
  INNER = INNER + 1
  IUNDR = 0
  ISTEP(1) = 0
  WRITE(6,'(1X,12 1X,4(E8.1,1X),11 1X,2(E8.1))')
1 INNER,CONCA,CONCBC,CONCM,TMP,ICOLL,K10
  WRITE(6,*)HINIT,RESET,TFIRST
  CALL BIMSTM(TOL,HINIT,RESET,TFIRST,TMOUT,IER)
  CALL OUTPUT(KHTMP,IER)
  GOTO 10
20 CONTINUE
RETURN
END

SUBROUTINE TBLTOP
WRITE(2,*)('A10,1X),4(A7,1X),A8,1X,A3)')
$ 'K_min/K_eq','HT results',CONCA,CONCBC,
$ 'CONCM','Temp','Collider',IER)
RETURN
END

```

```

SUBROUTINE TBLOUT(KHT,IER)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
COMMON/POP/NV(0:10,0:20),NTOT,CONCA,CONCBC,CONCM,JCOLL(3)
COMMON/BSDATA/HUSED(500),IQUSED(500),ISTEP(500),IUNDR(500)
1Y(500),Y1(500),Y2(500),Y3(500),Y4(500)
CHARACTER*3 COLLID
IF (ICOLL.NE.1) THEN
  COLLID = 'HCl'
ELSE
  COLLID = 'He'
ENDIF
TMP = TMP
CALL USMNM(KPH,LOOPCT,1,KMIN,KMAX)
KHT = KMIN/KPH(1)
ISTEPS = ISTEP/LOOPCT
KDIFF = (FRAT-KHT)/KHT
WRITE(6,'(1X,12 1X,4(E8.1,1X),11 1X,2(E8.1))')
$ 'FRAT,KHT,CONCA,CONCBC,CONCM,ISTEP,COLLID,IER'
RETURN
END

DOUBLE PRECISION FUNCTION GETCA(I)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
COMMON/POP/NV(0:10,0:20),NTOT,CONCA,CONCBC,CONCM,JCOLL(3)
GETCA = CONCA
RETURN
END

DOUBLE PRECISION FUNCTION GETCBC(I)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
COMMON/POP/NV(0:10,0:20),NTOT,CONCA,CONCBC,CONCM,JCOLL(3)
WRITE(6,*) 'in getcbc',CONCBC
GETCBC = CONCBC
RETURN
END

DOUBLE PRECISION FUNCTION GETCM(I)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
COMMON/POP/NV(0:10,0:20),NTOT,CONCA,CONCBC,CONCM,JCOLL(3)
GETCM = CONCM
RETURN
END

FUNCTION IGETCP(I)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
IGETCP = ICOLL
RETURN
END

DOUBLE PRECISION FUNCTION GETTMP(I)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
GETTMP = TMP
RETURN
END

DOUBLE PRECISION FUNCTION GETI0(ISPEC)
IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
IMPLICIT INTEGER (A,I-J,L-M,V)
IMPLICIT CHARACTER*6 (S)
COMMON/BS30WS/X,TMP,K10(2),KPH(500),XTENT(500),LOOPCT,ICOLL
GETI0 = K10(ISPEC)
RETURN
END

SUBROUTINE OUTPUT(KHT,IER)
REAL*8 KHT
CALL GETDAT
CALL TBLDAT(KHT,IER)
RETURN
END

```

```

SUBROUTINE GETDAT
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  COMMON/BSJWSX TNP F10(2) FPH(500) XTENT(500) LOOPCT ICOLL
  COMMON/BSDATA/HUSED(500) IUSED(500) ISTEP(500) IUNDR(500)
  Y1(500) Y2(500) Y3(500) Y4(500)
  CHARACTER*80 STRING
  READ(2) I
  READ(2) (A80) STRING
  WRITE(6) 'STRING LOOPCT'
  DO 10 I = 1, LOOPCT
    READ(2) (3D14.7) F10(I) FPH(I) XTENT(I)
    I = IUSED(I) ISTEP(I) IUNDR(I)
    READ(2) (2D15.8) Y1(I) Y2(I) Y3(I) Y4(I)
    XTENT(I) = XTENT(I)
  CONTINUE
  RETURN
END

SUBROUTINE PLTDAT
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  COMMON/BSJWSX TNP F10(2) FPH(500) XTENT(500) LOOPCT ICOLL
  COMMON/BSDATA/HUSED(500) IUSED(500) ISTEP(500) IUNDR(500)
  Y1(500) Y2(500) Y3(500) Y4(500)
  COMMON/POP/AV(10,10,0,20) NTOT,CONCA,CONCB,CONCM,ICOLL(3)
  CALL USCON(KPH,LOOPCT,1,EMIN,KMAX)
  EPS = 1.E-5
  DO 11 I = 1, LOOPCT
    IF (FPH(I)-KMIN)/FPH(I).LT.EPS) WRITE(6) 'YES XTENT(I)
    IF (KPH(I)-KMIN)/KPH(I).GE.EPS) WRITE(6) 'NO XTENT(I)
    WRITE(4) LOOPCT,ICOLL,TNP,X
    WRITE(4) XTENT(I),I=1,LOOPCT
    WRITE(4) KPH(I),I=1,LOOPCT
    WRITE(4) HUSED(I),I=1,LOOPCT
    WRITE(4) IUSED(I),I=1,LOOPCT
    WRITE(4) ISTEP(I),I=1,LOOPCT
    WRITE(4) IUNDR(I),I=1,LOOPCT
    WRITE(5) Y1(I),I=1,LOOPCT
    WRITE(5) Y2(I),I=1,LOOPCT
    WRITE(5) Y3(I),I=1,LOOPCT
    WRITE(5) Y4(I),I=1,LOOPCT
  RETURN
END

SUBROUTINE INCTT
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  IMPLICIT CHARACTER*(*)
  COMMON/BSJWSX TNP F10(2) FPH(500) XTENT(500) LOOPCT ICOLL
  LOOPCT = LOOPCT
  IF (LOOPCT.GT.500) THEN
    WRITE(6) 'loops...arrays are full'
    STOP
  ENDIF
  RETURN
END

SUBROUTINE DIFFUS
  The krxn(v,j) should be examined to see which values are
  not reasonable for the system of interest.
  For a bimolecular reaction this would imply that those krxn
  which exceed the collision frequency possible according
  to the temperature and pressure of the system.
  should have an upper limit which is equal to or less
  than the rate of diffusion. The rate-limiting step is
  no longer the speed of chemical conversion but the
  speed at which one molecule can meet another molecule.
  In this subroutine the collision frequency is calculated
  and any krxn(v,j) which exceeds it is set equal to it.
  If the user is modelling a unimolecular reaction, or an ideal
  reaction which is not limited by the rate of diffusion then
  the upper limit in the subroutine DIFFUS should be changed.
  IMPLICIT DOUBLE PRECISION (B-H,K,N-U,W-Z)
  IMPLICIT INTEGER (A,I-J,L-M,V)
  REAL*8 AVOGAD,AMASS,ARAD
  PARAMETER (AVOGAD=6.02205D23)
  COMMON/IC/INP,ICUT,ISAVE,ISUP,ISUPG,INPAT,INOV,VOINV,VUNIT(0:23)
  COMMON/ICOUNT/MAXV,MAX(10:10) MAXUN INPTS ICOLL
  COMMON/MASST/KVJBM(0:10,0:10)
  COMMON/ENERGY/ELEV(0:10,0:10) TENG
  COMMON/PHYSCT/PLANCY,REOLTE,RGAS,SCEXP1,CLIGHT,PI,IEVER
  COMMON/MASS/BMASS,CMASS,AMASS,REDMAS,BRAD,CRAD,ARAD,RE
  COMMON/POP/AV(10,10,0,20) NTOT,CONCA,CONCB,CONCM,ICOLL(3)
  IFLAG = 0
  WRITE(10) 'Should the microscopic rate constants
  * have an upper limit equal to that of the maximum collision
  * frequency possible?'
  WRITE(10) ' 1) yes, establish this as an upper limit
  WRITE(10) ' 2) the upper limit is proportional to
  * the collision frequency
  WRITE(10) ' 3) no upper limit (e.g. unimolecular rxn)'
  READ(10) 'END=903) ANDIFF
  GOTO 904
903 IF (ICUT.EQ.ISUP) THEN
  INP = 5
  IOUT = 6
  GOTO 100
ELSE
  CALL DPALTD(ANDIFF,1)
  ENDF
904 WRITE(10) 'Error in upper limit specification: ANDIFF
  IF (ANDIFF.EQ.3) THEN
  GO TO 104
ELSEIF (ANDIFF.NE.1.AND.ANDIFF.NE.3) THEN
  WRITE(10) 'Error in upper limit specification: ANDIFF
  WRITE(10) 'Specify 1, 2 or 3.'
  IFLAG = IFLAG + 1
  IF (IFLAG.LE.1) GOTO 100
  CALL ISTERM(ANDIFF,ISUB)
  ENDF

C
C The rate of collision, Z, may be expressed in a form analogous to
C the rate of reaction with a diffusion coefficient, k_diff:
C Z = k_diff (A) (B)
C where (A) and (B) are the concentrations of the species involved
C in collision.
C For each v,j level
C we want to compare
C k_diff with the k_vjbm coefficient input to the
C program. Since only collisional processes are considered in this
C simulation, no k_vjbm should exceed k_diff.
C The formula for Z is from J.N. Levine, Physical Chemistry
C (Mcgraw-Hill, 1971), p. 411:
C Z = pi^1/2 (A + r_BC)^2 * (8*kT/pi)^1/2 * (1/M_A + 1/M_BC)^-1/2 *
C * (N_A/V) * (N_BC/V)
C Therefore k_diff = Z / ((A) * (B) * N_AV)
C where the additional N_AV in the denominator comes in because we are
C considering the number of "moles" of collisions rather than the
C number of collisions.
C or
C k_diff = pi^1/2 * (A + r_BC)^2 * (8*kT/pi)
C * RECIPROCAL-REDUCED-MASS^-1/2 * (1/2) * N_AV
C This will be used to estimate the upper limit for
C the bimolecular rate coefficient.
C The molecular diameter is estimated as the sum of atomic
C radii whose values are passed through common block MASS.
C The atomic masses and radii are found in subroutines RKNMYS.
C Since we are concerned only with reactive collisions (A-BC)
C we ignore M-BC collisions in determining k_diff.
C REC = (BRAD+CRAD)
C BMASS=BMASS*CMASS
C RECM = (1.0D0/AMASS + 1.0D0/BMASS)*AVOGAD
C AMASS and BMASS come in on a "per mole" basis...we need them
C on a "per molecule" basis so we divide the mass by N_AV which
C is equivalent to multiplying the reciprocal by N_AV as above.
C KDIFF=PI^1/2*(ARAD+RBC)**2*DSQRT((8.0D0*RGAS*TENG/PI)*RECM)*AVOGAD
  WRITE(10) '700) KDIFF
  FORMAT(' Collision coefficient (k_diff) is .D13.4 sec^-1')
  IF (ANDIFF.EQ.2) THEN
105 WRITE(10) '700) What is the ratio of the upper limit to the
  collision coefficient? <Default: 1.0>'
  READ(10) 'END=908) RATIO
  GOTO 908
908 IF (ICUT.EQ.ISUP) THEN
  INP = 5
  IOUT = 6
  GOTO 100
ELSE
  CALL DPALTD(RATIO,1.0D0)
  ENDF
909 KDIFF=RATIO*KDIFF
  ENDF
DO 100 V=0, MAXV
  DO 100 J=0, MAXJ(V)
  IF (KVJBM(V,J).GT.KDIFF) THEN
  WRITE(6) '701) V,J,KVJBM(V,J),KDIFF
  KVJBM(V,J)=KDIFF
701 FORMAT(' Kvjbm(',D13.4) exceeds the collision

```

# APPENDIX C. SOURCE CODE

```
4      coefficient: 0.010.4.0 > 0.010.4.0 Value reset.0
      ENDIF
100 CONTINUE
104 RETURN
      END

SUBROUTINE TRAPF(I1,I2,D1,D2)
COMMON/UNDERF/IUNDER
REAL*8 D1
IUNDER = IUNDER - 1
I1 = 0
C WRITE(6,*) underflow: IUNDER
RETURN
END
```

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